

Fabrication and Magnetic Studies of (Co, Zn)-doped γ -Fe₂O₃ Thin Films

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Abstract—A fabrication process for the sputtered (Co, Zn)-doped γ -Fe₂O₃ thin films with quite a short reduction time (10 minutes) at temperature between 350-370 °C and with oxidation time about 20 minutes at temperature between 310-330 °C has been established. It has been observed that, with a few percent of Co & Zn content, the Zn dopant has the effect of producing higher saturation magnetization and better thermal stability of coercivity, as well as better suppressing grain growth effect during the heat treatments when these films are compared to γ -Fe₂O₃ thin films doped with Co only.

I. INTRODUCTION

Both Co-based metal films [1,2] and Fe-based oxide films [3-6] are attractive as a high density magnetic recording disk media. However, the oxide films have the advantage of better corrosion and wear resistance. The physical properties of oxide films depend significantly on the amount and kind of dopants as well as their preparation conditions. Various techniques have been used to prepare the iron oxide films; such as the reactive evaporation [3], the reactive sputtering [4] and the chemical vapor deposition [5] techniques. Most of the iron oxide thin films studied before have been doped with a certain amount of impurities (i.e. dopants); for example, the addition of Co can increase the coercivity [6]; Ti has been employed as a dopant to improve the squareness of the films [4]; and Cu, Os, and Al dopants have the effect of suppressing grain growth during heat treatments [6].

II. EXPERIMENTAL

The (Co, Zn)-doped γ -Fe₂O₃ thin films about 2000 Å thick on silicon wafer or glass substrates were prepared by reactive rf sputtering from Fe targets which contained proper amount of Co and Zn in an atmosphere of a mixed gas of Ar and O₂. The substrate temperature T_s during sputter deposition was kept at 120 °C. The as-deposited α -Fe₂O₃ films were then reduced to Fe₃O₄ films under wet hydrogen atmosphere at temperatures varied between 300 and 400 °C and for times varied from 5 to 60 minutes. Oxidation from Fe₃O₄ films to γ -Fe₂O₃ films was performed by heating films in air for temperatures between 310 and 330 °C for 20 minutes. The film thicknesses were measured with stylus profilometer. The Co and Zn contents, defined as the amount of Co and Zn substituted for Fe in the films, were determined by x-ray photoelectron spectroscopy (ESCA) and electron

probe microanalysis (EPMA). The grain size were determined by means of a transmission electron microscope (TEM). The crystal structure of the films was confirmed by means of a x-ray diffractometer. The magnetic properties of the films were measured with both vibration sample magnetometer (VSM) and superconducting quantum interference device (SQUID).

III. RESULTS AND DISCUSSION

The crystal grain size of each of the samples was investigated by TEM. The films with pure or Co-doped γ -Fe₂O₃ have an average grain size about 700 Å as shown in Fig.1 (a). However, with the doping of roughly above 2 wt. % of Zn, the average grain size decreased significantly. Fig.1 (b) shows the transmission electron micrographs of (Co, Zn)-doped γ -Fe₂O₃ films containing 5 wt. %Co & 4wt.%Zn. The average grain size (~300Å) of the (Co, Zn)-doped γ -Fe₂O₃ films was much smaller than that of the pure films. This means that the grain growth during the heat treatment is suppressed by the Zn addition.

From the measurement of the saturation magnetization M_s and the coercivity H_c of all the γ -Fe₂O₃ films as function of the reduction temperature T_R between 300 and 400 °C, and the reduction time t_R between 5 and 60 minutes. We observed several general behaviors: (1) All the film samples deposited on silicon wafer (100) have a little higher H_c and squareness ratio S; but roughly the same M_s as that on glasses. (2) The maximum values of both M_s vs T_R and H_c vs T_R for all the samples occur between 350 and

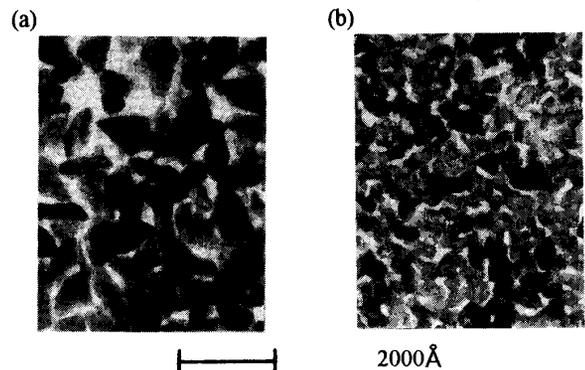


Fig. 1 The TEM micrographs of (a) pure γ -Fe₂O₃ films, and (b) (Co, Zn)-doped γ -Fe₂O₃ films containing 5wt.% Co & 4wt.% Zn.

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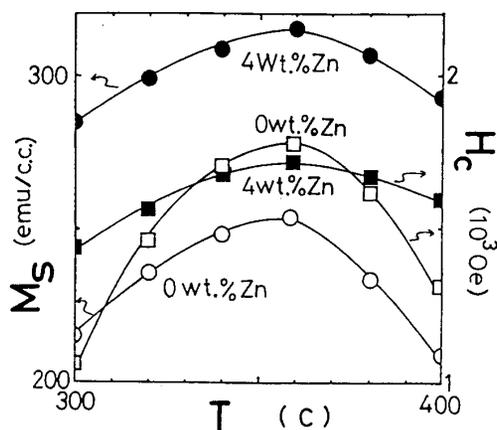


Fig.2 The M_s and H_c as functions of T_R at $t_R=10$ minutes for two (Co, Zn)-doped $\gamma\text{-Fe}_2\text{O}_3$ films with 5wt.% Co, and with both 5wt.% Co and 4wt.% Zn.

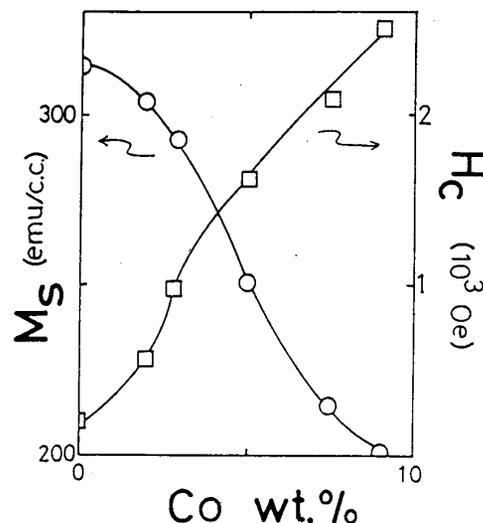


Fig.4 The M_s and H_c of Co-doped $\gamma\text{-Fe}_2\text{O}_3$ films as function of Co content.

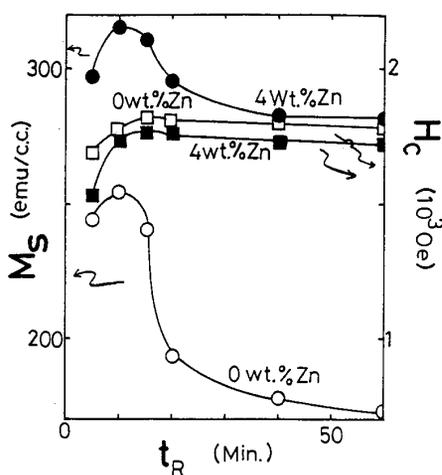


Fig.3 The M_s and H_c of (Co, Zn)-doped $\gamma\text{-Fe}_2\text{O}_3$ films as functions of t_R at $T_R=360^\circ\text{C}$ for two (Co, Zn)-doped $\gamma\text{-Fe}_2\text{O}_3$ films with 5 wt.% Co, and with both 5wt.% Co and 4wt.% Zn.

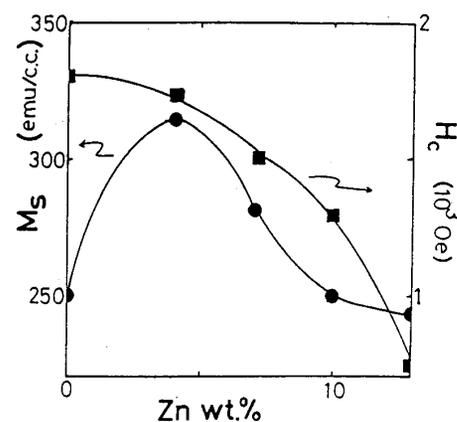


Fig.5 The M_s and H_c of (Co, Zn)-doped $\gamma\text{-Fe}_2\text{O}_3$ films containing 5wt.% Co as functions of Zn content.

370 °C, if the t_R is kept at 10 minutes. (3) For all the Zn-doped film samples, the H_c decreases and M_s increases, when they are compared with the undoped samples. (4) The higher value of both M_s vs t_R and H_c vs t_R for all the samples occur at roughly 10 minutes if the T_R is kept at 360 °C. As an example, The M_s and H_c as functions of both T_R and t_R for two $\gamma\text{-Fe}_2\text{O}_3$ thin films (one is doped with 5wt.% Co and the other is doped with both 5wt.% Co and 4wt.% Zn) are plotted in Fig.2 for T_R varyig between 300 and 400°C and keeping t_R at 10 minutes, and in Fig.3 for t_R varying

between 5 and 60 minutes and keeping T_R at 360 °C. We observed that for M_s vs T_R and H_c vs T_R , curves show maximum around 360°C; for M_s vs t_R and H_c vs t_R , M_s shows maximum at $t_R=10$ minutes, and H_c shows slowly decreasing after t_R roughly above 20 minutes. Why H_c decreases much slower than M_s for t_R higher than 20 minutes? We consider this is due to the formation of a small amount of pure Fe on surface during the reduction process for t_R longer than 10 minutes. After oxidation, a layer of

nonmagnetic FeO or α -Fe₂O₃ forms at the film surface; this will decrease Ms. However, the nonmagnetic layer and inner core of magnetic γ -Fe₂O₃ will produce an induced stress due to lattice mismatch, and this stress anisotropy may slow down the decreasing of Hc.

For studying the effect of Co and Zn dopants in γ -Fe₂O₃ thin films, the heat treatment conditions for all the samples were with reduction temperature roughly at 360°C for 10 minutes, and with oxidation temperature roughly at 320°C for 20 minutes.

The magnetic properties of Co-doped γ -Fe₂O₃ films as functions of Co content were plotted in Fig. 4. We found that doping of Co into γ -Fe₂O₃, results in enhancement of Hc of the undoped samples. Fig. 4 shows that Hc increases monotonically from roughly 200 Oe for undoped γ -Fe₂O₃ films to roughly 2500 Oe for γ -Fe₂O₃ films doped with 9wt.%Co. However, the Ms decreases monotonically from roughly 315 emu/c.c. for undoped γ -Fe₂O₃ films to roughly 200 emu/c.c. for 9.0wt.%Co-doped γ -Fe₂O₃ films. Therefore, we can vary the Co content below 9.0wt.% for get either higher Ms with lower Hc or vice versa for γ -Fe₂O₃ films.

For studying the effect of both Zn and Co dopants, we chose to fix the Co content of the films as 5.0 wt.%; and to vary the Zn content from 0 to 13.0 wt.%. The magnetic properties of the films as functions of Zn content are shown in Fig. 5. It is obvious that the Hc decreases monotonically with increasing the Zn content. Hc varies from 1800 Oe for 5wt.%Co-doped γ -Fe₂O₃ films to 740 Oe for 5wt.%Co-doped γ -Fe₂O₃ films with 13.0wt.%Zn. The increases of Ms with increasing Zn content roughly below 4wt.%Zn can be understood owing to the increase of the magnetic moment of the spinel lattice. We know that both the substitution of Fe³⁺ ions and vacancies by Zn²⁺ ions in the γ -Fe₂O₃ lattice, and the Fe³⁺ ions forced from tetrahedral A sites to octahedral B sites by the Zn²⁺ ions located at A sites can increase the magnetic moment of the γ -Fe₂O₃ system. The decrease of Hc with increasing Zn content can be also understood owing to the decrease of the magnetocrystalline anisotropy [7].

For (Co, Zn)-doped γ -Fe₂O₃ film samples with 5wt.%Co and with Zn content larger than 4wt.%, the decrease of Ms is due to the substitution of large amount of Zn²⁺ into the spinel structure of A sites, therefore the moments at A sites will be too weak to affect the B site moments, so that the net moments decrease [8]. It was found that the squareness of the (Co, Zn)-doped γ -Fe₂O₃ films was lower than that of the Co doped γ -Fe₂O₃ films. For example, it decreases from 0.85 for 9.0wt.%Co-doped γ -Fe₂O₃ film samples to 0.65 for samples with 9.0wt.%Co & 13.0wt.%Zn. However, the exact mechanism of these behaviors is not clear at present. Further research is underway, and will be reported later.

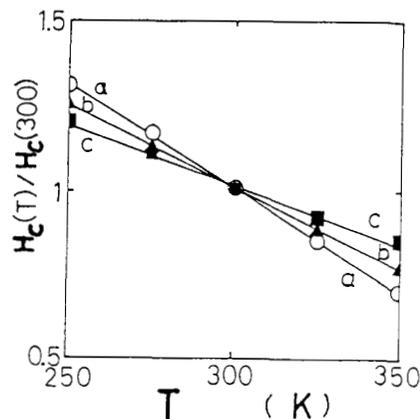


Fig.6 The normalized coercivity as functions of temperature between 250 and 350 K for three (Co, Zn)-doped γ -Fe₂O₃ film samples containing 5wt.%Co and with (a) 0wt.%Zn, (b) 6wt.%Zn, and (c) 13wt.%Zn

Fig. 6 shows the temperature dependence of the normalized coercivity $H_c(T)/H_c(300)$ of three (Co, Zn)-doped γ -Fe₂O₃ film samples containing 5wt.%Co and with (a) 0wt.%Zn, (b) 6wt.%Zn, and (c) 13wt.%Zn as functions of temperatures near room temperature (250~350 K). From this Fig. we find that the temperature dependence of Hc is decreased with increasing Zn content.

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