## 行政院國家科學委員會專題研究計畫成果報告

計畫名稱:有機金屬錯合物官能基轉換及催化反應(二)

**Novel Organometallic Functional Transformation** 

and Catalysis (II)

計畫編號: NSC87-2113-M002-001

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### 一、中文摘要

炔丙基及丙二烯基有機金屬錯合 物的親核加成反應形成氮代三亞甲 基甲烷,金屬環丁烯,及中央碳取 代的丙烯基錯合物。

關鍵詞: 炔丙基,丙二烯基,親核加 成反應

### **Abstract**

Addition of amines, phosphines, and electron-rich aromatic derivatives to the n3allenyl/propargyl complexes lead to the formation of n3-azatrimethylenemethane, metallacyclobutene, and central carbonsubstituted \(\eta^3\)-allyl complexes via C-N, C-P, and C-C bond formation, respectively.

Keywords: η3-allenyl/propargyl complexes, nucleophilic addition.

### 二、緣由與目的

The transition metal complexes with unsaturated hydrocarbon of  $\pi$ -coordination have long been the focus of organometallic chemistry. Those with non-classical organometallic interactions are of particular interest, for they often empower new activity to the organic moieties [1]. For example, organic allyls (C<sub>3</sub>R<sub>5</sub>-) often bond with metal in a n3-form rather than a n1-form. The chemistry involving \(\eta^3\)-allyl complexes has been extensively studied and proved to contribute crucially to organic synthesis [2]. Propargyl which may be conceived as the alkyne analog of allyl has a tautomeric

isomer of allenyl. Similar to allyl, propargyl and allenyl (C<sub>3</sub>R<sub>3</sub>-) which also have three-carbon skeletons and unsaturated hydrocarbyl functionalities, ought to allow rich chemistry with metal. transition metal complexes of allenyl and propargyl have been drawing attention for three decades, since the first reported synthesis of metal n1-allenyl [3]. Besides, organometallic propargyl and allenyl have revealed their synthetic utility in many organic reactions [3, 4]. mononuclear allenyl or propargyl metal complexes with non-classical  $\sigma,\pi$ -bonding were actually unexplored until recently.

It has been known that metal η1propargyl can transform into the thermodynamically more stable η1-allenyl tautomer The mechanistic studies indicate the involvement of 1,3-metal shift in this process of tautomerization. Comparing such a metal shift with the  $\eta 1$ -to- $\eta 3$ -to- $\eta 1$  fluxional motion of allyl [6], it would be chemically logical to expect the  $\eta$ 3-bonding interaction between metal and the allenyl and/or propargyl ligands. Recent development in metal \(\eta^3\)-allenyl/propargyl chemistry has filled such a category and opened a new page in organometallic chemistry. We choose to use the nomenclature of n3-allenyl/propargyl, because the organic ligands with a general C<sub>3</sub>R<sub>3</sub> form are shown to bond with metal through three carbon atoms with hybridization of allenyl and propargyl features. Such a bonding mode of  $\eta 3$ -C<sub>3</sub>R<sub>3</sub> groups remarkably enhances their electrophilic character, especially at the central carbon, and makes them as potent organometallic carbon electrophiles [7]. In fact, the

cationic mononuclear  $\eta 3$ -allenyl/propargyl complexes are found prone to external attack by various hard and soft nucleophiles. This report covers the addition of amines, phosphines, as well as electron-rich aromatic which strongly suggest the character of potent carbon electrophiles for the  $\eta 3$ -allenyl/propargyl complexes

### 三、報告內容

# I. Addition of Aromatic C-H Bond to An y3-Allenyl/Propargyl Complex. Organoplatinum-Induced Electrophilic Aromatic Substitution Reactions

The regioselective addition of a C-H bond of the pyrrole, indole, or benzene derivatives to the n3-allenyl/propargyl ligand of  $\{Pt(PPh_3)_2(\eta_3-C_3H_3)\}(BF_4)$  (1) generates the pyrrolyl-, indolyl-, or aryl-substituted \( \eta^3 - \)  $CH_2$ ]{ $BF_4$ } (Ar = 2-pyrrolyl **2a**, 2-N-methylpyrrolyl 2b, 3-indolyl 3a, 2-3-methylindolyl 3b, 4-dimethylamino-phenyl 4, 2,4dimetho-xyphenyl 5, 2,4,6-trimethoxyphenyl **6**), respectively. The regiochemistry of the C-H activation and a crossover labeling experiment support electrophilic substitution Further nucleophilic addition mechanism. to 6 leads to demetallation and achieves aryl vinylation.

### II. Synthesis of Metallacyclobutenes Complexes of Late Transition Metals via Nucleophilic Addition of Allenyl or Propargyl Complexes.

The regioselective addition of NEt<sub>3</sub>, PPh<sub>3</sub>, or pyridine to the central carbon of an  $\eta$ 3-allenyl/-propargyl complex of platinum  $\{Pt(PPh_3)_2(\eta_3-C_3H_3)\}(BF_4)$  (1) leads to the formation of the new cationic platinacyclobutenes  $\{(PPh_3)_2Pt(CH_2C(Nu)CH\}(BF_4)$  (Nu = NEt<sub>3</sub> (2a), PPh<sub>3</sub> (2b), C<sub>6</sub>H<sub>5</sub>N (2c)) via formation of a C-N or C-P bond, respectively. Complex 2c can transform into {cis-Pt(PPh<sub>3</sub>)<sub>2</sub>(Py)( $\eta$ 1-CHCCH<sub>2</sub>)}(BF<sub>4</sub>) (3). The reverse reaction has not been observed. It is

suggested that nucleophilic addition of **1** likely involves the external attack at the central carbon of the  $\eta$ 3-allenyl/propargyl ligand. Protonation of **1** yields  $\{Pt(PPh_3)_2 - (\eta_3-CH_2C(PPh_3)CH_2]\}(BF_4)_2$  (7). Addition of  $PPh_3$  to labile  $\eta$ 1-allenyliridium complex (OC-6-42)-Ir(Cl)( $PPh_3$ )<sub>2</sub>(OTf)(CO)( $\eta$ 1-CHC-CH<sub>2</sub>) (4) results in an iridacyclobutene  $\{(Cl)(PPh_3)_2(CO)Ir[CH_2C(PPh_3)CH]\}(OTf)$  (**5**). The single crystal X-ray structure of **5** has been determined.

### III. A New Stable Intermediary Mode between y3-2-Aminoallyl Complexes and Metallacyclobutanimines. Synthesis and Structural Characteristic of y3-Azatrimethylenemethane and N-Protonated, N-Alkylated, N-Arylated y3-Azatrimethylenemethane Complexes of Pt and Pd.

Regioselective addition of ammonia, primary or secondary amines, aniline, or amino derivatives either to a neutral (n1allenyl)platinum complex trans-Pt(Br)- $(PPh_3)_2(\eta_1-CHCCH_2)$  (1) or to a cationic  $\eta_3$ allenyl/propargyl platinum complex  $[Pt(PPh_3)_2(\eta_3-C_3H_3)](BF_4)$  (2) provide the synthesis of cationic N-protonated, alkylated, and N-arylated n3-azatrimethylenemethane complexes {Pt(PPh<sub>3</sub>)<sub>2</sub>[η3-CH<sub>2</sub>C- $(NRR')CH_2$ }(X) (R = H R' = H (3a), Me(**3b**), Et (**3c**), <sup>i</sup>Pr (**3d**), <sup>t</sup>Bu (**3e**), c-C<sub>6</sub>H<sub>11</sub> (**3f**), Ph (3g), CH<sub>2</sub>CH<sub>2</sub>OH (3h), R = R' = Et (3i), c- $C_3H_6$  (from azetidine 3j), Ph (3k), R = Me R' = Ph (3l); X = Br, BF<sub>4</sub>), respectively. Addition of amides to 1 gave a neutral \( \eta^3 - \) azatrimethylenemethane complex Pt(PPh<sub>2</sub>)<sub>2</sub>-[n3-CH<sub>2</sub>C(NSO<sub>2</sub>Ph)CH<sub>2</sub>] (4m).Similar reactions using palladium complexes yield  ${Pd(PPh_3)_2[\eta 3-CH_2C(NRR')CH_2]}(X) (R = H$  $R' = {}^{i}Pr$  (7d), Ph (7g), R = R' = Et (7i); X =Br, BF<sub>4</sub>, OTf),  $Pd(Br)(PPh_3)[\eta_3-CH_2C(NEt_2) CH_2$ ] (8i) and  $Pd(PPh_3)_2[\eta_3-CH_2C(NR)CH_2]$  $(R = SO_2Ph (9m), p-SO_2C_6H_4Me (9n)).$ Synthesis of three complexes, {M(PPh<sub>3</sub>)<sub>2</sub>- $[\eta_3\text{-CH}_2\text{C(NHR)CH}_2]^+$  (M = Pt R = SO<sub>2</sub>Ph (3m); M= Pd R = SO<sub>2</sub>Ph (7m), p-SO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>-Me (7n)), can not be done by hydroamination reactions, but has been succeeded using protonation of n3-N-TMM complexes 4m,

9m, and 9n, respectively. Spectroscopic and crystallographic characterizations indicate that these N-TMM species exhibit intermediary structural features between  $\eta$ 3-2-aminoallyl and metallacyclobutanimine complexes.

### **Summary**

We have extendeded the nucleophilic addition of  $\eta$ 3-allenyl/propargyl complexes to C-N, C-P, and C-C bond formation. The resulting products include  $\eta$ 3-azatrimethylenemethane, metallacyclobutene, and central carbon-substituted  $\eta$ 3-allyl complexes.

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