行政院國家科學委員會專題研究計畫 期中進度報告

含胺 - 亞胺配基的化學(2/3)

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complexes $[R^1R^2NCMe_2CH=N(2,6-R^3{}_2C_6H_3)]NiBr_2$ ($R^1=R^2=Me$ $R^3=Me$ (**Ni-1b**), $R^3=^iPr$ (**Ni-1c**), $R^1=R^2=Et$ $R^3=H$ (**Ni-2a**), Me (**Ni-2b**), iPr (**Ni-2c**), $R^1=R^2=^nPr$ $R^3=^iPr$ (**Ni-3c**), $(R^1R^2)=c$ - C_3H_6 $R^3=^iPr$ (**Ni-4c**), $(R^1R^2)=c$ - C_4H_8 $R^3=^iPr$ (**Ni-5c**)).

Scheme I
$$R^{3}$$

$$R^{1}R^{2}NH$$

$$R^{1}R^{2}N$$

$$R^{2}N$$

$$R^{1}R^{2}N$$

$$R^{1}R^{2}N$$

$$R^{2}N$$

$$R^{3}N$$

The violet dibromonickel complexes are generally soluble in CH_2Cl_2 or $CHCl_3$, and appear to suffer deterioration by exposing to ambient conditions. The SQUID measurement for Ni-2c indicates a ground state of triplet. The single-crystals of Ni-1b were grown from CH_2Cl_2/Et_2O . Its molecular structures in distorted tetrahedral geometry are unequivocally confirmed by X-ray crystallography, and the ORTEP drawing is shown in Figure 1. The chelation of α -aminoaldeimine to nickel constitutes a five-membered metallacycle. The N1-Ni-N2 and Br1-Ni-Br2 angles are $82.2(4)^{\circ}$ and $114.8(3)^{\circ}$ respectively, being comparable with those found in the diimine nickel catalysts. The bond distances of Ni-N_{im} and Ni-N_{am} are 1.988(9) and 2.059(9) Å; and C-N_{im} and C-N_{am} are 1.26(2) and 1.50(2) Å, respectively. As observed in other square-planar catalysts with coordinating diimines, the ortho-substituted imino phenyl is prone to dispose perpendicular to the plane of imine. Such a feature is supposed to facilitate the olefin polymerization. The amine of sp^3 configuration that is distinguishable from the imine of sp^2 results in the non-planar metallacycle. As consequence, the substituents of both the amino nitrogen and its adjacent carbon should be enabled to sterically affect the axial coordination sites as well as the vicinal