

## Intramolecular Cyclizations of $\alpha$ -Stannyli Radicals to Acylsilanes: Regiospecific Syntheses of Five-membered Cyclic Silyl Enol Ethers

Yeun-Min Tsai\* and Sheng-Yueh Chang

Department of Chemistry, National Taiwan University, Taipei, Taiwan 106, Republic of China

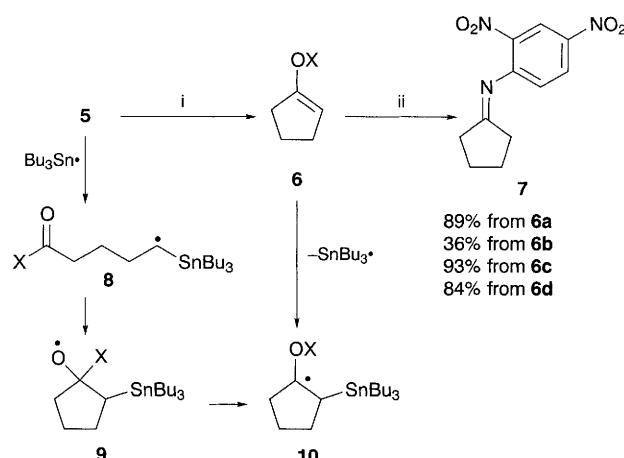
$\alpha$ -Stannyli radicals generated from acylsilanes **5**, **13** and **14** cyclize to give good yields of cyclic silyl enol ethers after sequential cyclizations, radical Brook rearrangements and  $\beta$ -scissions.

Acylsilanes have gained more and more attention in recent years.<sup>1</sup> Radical cyclization reactions of acylsilanes involve the formation of  $\beta$ -silyl alkoxy radicals **1** which quickly rearrange to give  $\alpha$ -silyloxy radicals **2**.<sup>2</sup> This special feature creates opportunities for further elaboration.<sup>3</sup> Here we report the use of  $\alpha$ -stannyli radicals to cyclize with acylsilanes and give cyclic silyl enol ethers in a regiospecific way.

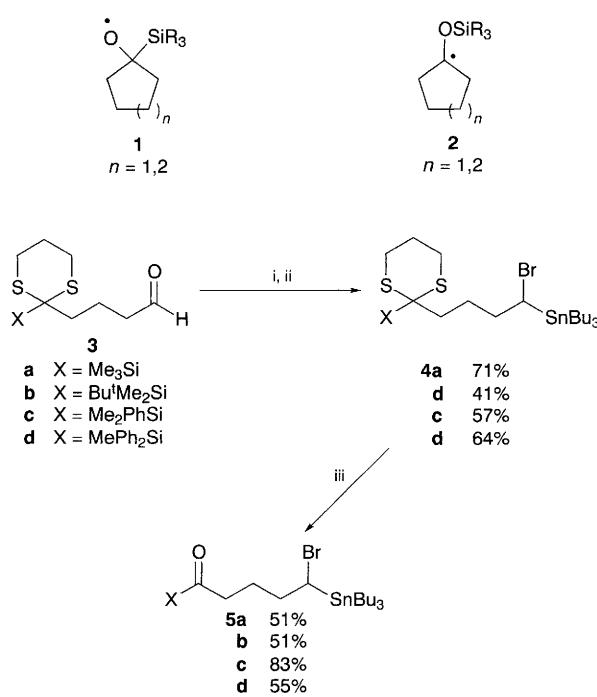
Aldehydes **3**<sup>3a</sup> were treated with tributyltin anions and the resulting  $\alpha$ -stannyli alcohols were converted to  $\alpha$ -stannyli bromides **4** using triphenylphosphine and tetrabromomethane (Scheme 1). Hydrolysis of **4** with ceric ammonium nitrate (CAN)<sup>4</sup> gave acylsilanes **5** in moderate yields over the three steps.

Cyclizations of **5** in refluxing benzene with slow addition of a catalytic amount of tributyltin hydride (0.15 equiv.) and AIBN (0.05 equiv.) gave silyl enol ethers **6** (Scheme 2).<sup>†</sup> The presence of **6** was confirmed by comparing their crude <sup>1</sup>H NMR spectra with those of the authentic samples prepared from cyclopentanone. However, the crude products of **6** were converted to the 2,4-dinitrophenylhydrazone of cyclopentanone to estimate the yields of the cyclizations. In the cases of **6a**, **c** and **d**, the overall conversion yields were very good. The low yield in the case with a bulky *tert*-butyldimethylsilyl group **6b** indicated that the cyclization was not efficient. This may attributable to the steric interactions between the silyl group and the tributylstannyli group. The whole process involves the generation of  $\alpha$ -stannyli radicals **8** which cyclize to give radicals **9**. These radicals undergo radical Brook rearrangements<sup>5</sup> to give radicals **10** which are set up to perform facile  $\beta$ -scissions and give silyl enol ethers **6** with concomitant formation of a tributyltin radical. The tin radical then abstracts the bromine atom from **5** to regenerate radicals **8**, thus completes the cycle.

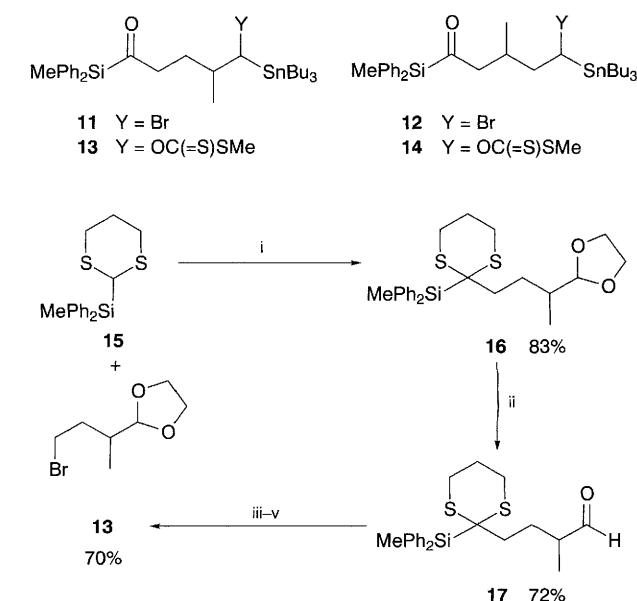
To demonstrate the regiospecific nature of this approach for the preparation of cyclic silyl enol ethers, we intended to synthesize bromides **11** and **12**. However, the same route used for the synthesis of **5** gave low yields in these cases. This was probably due to the instability of the structures of type **4** in which a nucleophile and an electrophile coexist.<sup>6</sup> Therefore, a different route was developed. Alkylation of the 2-silyl-1,3-dithiane **15** with 2-(3-bromo-1-methylpropyl)-1,3-dioxolane<sup>7</sup> gave **16** in 83% yield (Scheme 3). The aldehyde **17** obtained from hydrolysis of **16** was treated with tributyltin lithium, and the resulting alkoxide was trapped sequentially with carbon disulfide and methyl iodide to construct the xanthate moiety. The dithiane was then hydrolysed with



Scheme 2 Reagents and Conditions: i, Bu<sub>3</sub>SnH (0.15 equiv.), AIBN (0.05 equiv.), C<sub>6</sub>H<sub>6</sub>, 80 °C; ii, 2,4-dinitrophenylhydrazine, H<sup>+</sup>, EtOH



Scheme 1 Reagents: i, Bu<sub>3</sub>SnLi; ii, Ph<sub>3</sub>P, CBr<sub>4</sub>; iii, CAN, MeCN, H<sub>2</sub>O



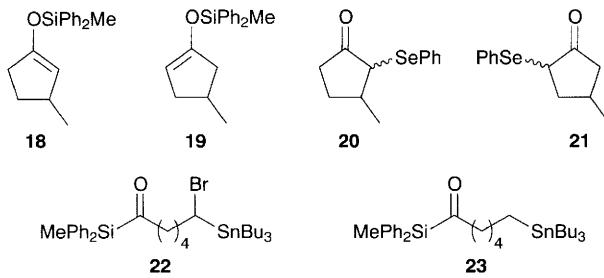
Scheme 3 Reagents: i, BuLi; ii, p-Me-C<sub>6</sub>H<sub>4</sub>SO<sub>3</sub>H (cat.), H<sub>2</sub>O, THF; iii, Bu<sub>3</sub>SnLi; iv, CS<sub>2</sub>, MeI; v, (CF<sub>3</sub>CO<sub>2</sub>)<sub>2</sub>IC<sub>6</sub>H<sub>5</sub>, H<sub>2</sub>O, MeCN, THF, NaHCO<sub>3</sub>

iodobenzene bis(trifluoroacetate) to give **13** in 70% yield from the aldehyde.<sup>8</sup> The xanthate **14** was synthesized accordingly in similar yields.

Radical cyclizations of **13** and **14** under the same conditions as stated above gave the two regioisomeric silyl ethers **18** and **19**, respectively. To confirm the structures, the crude products were treated with phenylselenyl bromide in dichloromethane at  $-78^{\circ}\text{C}$ . The selenide **20** was obtained in 68% yield from **13** as a mixture of *cis/trans* (1:2) isomers.<sup>9</sup> The selenide **21** was obtained in 77% yield from **14** as an 8.8:1 mixture of two stereoisomers.<sup>‡</sup>

Attempted radical cyclization of **22** initiated with tributyltin hydride (0.15 equiv.) and AIBN (0.05 equiv.) was not successful. Only the reduction product **23** was observed along with mostly unreacted **22**. With a full equivalent of tributyltin hydride, the acylsilane **23** was isolated in 80% yield. It has been reported for radical cyclizations of acylsilanes that 1,5-hydrogen atom transfer was a competing process with 1,6-cyclization.<sup>3a</sup> In the case of **22**, the unfavourable interactions between the silyl and stannylyl groups probably blocked the cyclization completely.

In summary, the radical cyclizations described herein represent a novel regiospecific approach to the syntheses of five-membered cyclic silyl enol ethers. This report also demonstrates that  $\alpha$ -stannyl radicals can be very useful if



properly designed. The potential of  $\alpha$ -stannyl radicals is currently under investigations.

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## Footnotes

<sup>†</sup> Typical procedure: Tributyltin hydride (21  $\mu\text{l}$ , 0.078 mmol) and AIBN (5 mg, 0.03 mmol) in benzene (5 ml) were added over 1 h to a solution of **5c** (295 mg, 0.503 mmol) in benzene (5 ml) heated at  $80^{\circ}\text{C}$ . The resulting solution was heated for a further hour and then concentrated *in vacuo* to obtain the crude product which was used directly in the subsequent reaction.

<sup>‡</sup> The major isomer was presumably the *trans* isomer; however, the stereochemistry was not determined. The characteristic  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ) signal for the major isomer appears at  $\delta$  3.82 (br d,  $J$  6.6 Hz), and that of the minor one appears at 3.72 (br dd,  $J$  11, 8.4 Hz).

## References

- A. Ricci and A. Degl'Innocenti, *Synthesis*, 1989, 647; P. C. B. Page, S. S. Klair and S. Rosenthal, *Chem. Soc. Rev.*, 1990, **19**, 147; P. F. Cirillo and J. S. Panek, *Org. Rep. Procedure Int.*, 1992, **24**, 553.
- Y.-M. Tsai and C.-D. Cherng, *Tetrahedron Lett.*, 1991, **32**, 3515.
- (a) Y.-M. Tsai, K.-H. Tang and W.-T. Jiaang, *Tetrahedron Lett.*, 1993, **34**, 1303; (b) D. P. Curran, W.-T. Jiaang, M. Palovich and Y.-M. Tsai, *Syn. Lett.*, 1993, 403.
- T.-L. Ho, H. C. Ho and C. M. Wong, *J. Chem. Soc., Chem. Commun.*, 1972, 791; H.-J. Cristau, B. Chabaud, R. Labaudinière and H. Christol, *Synth. Commun.*, 1981, **11**, 423.
- J. C. Dalton and R. A. Borque, *J. Am. Chem. Soc.*, 1981, **103**, 699; J. M. Harris, I. MacInnes, J. C. Walton and B. Maillard, *J. Organomet. Chem.*, 1991, **403**, C25.
- A. E. Davey, A. F. Parsons and R. J. K. Taylor, *J. Chem. Soc., Perkin Trans. I*, 1989, 1853; Z. Sui, P. S. Furth and J. J. De Voss, *J. Org. Chem.*, 1992, **57**, 6658.
- D. J. Collins and A. M. James, *Aust. J. Chem.*, 1989, **42**, 223.
- G. Stork and K. Zhao, *Tetrahedron Lett.*, 1989, **30**, 287.
- T. Toru, T. Okumura and Y. Ueno, *J. Org. Chem.*, 1990, **55**, 1277.