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## Capillary electrophoretic separation of 1 to 10 kbp sized dsDNA using poly(ethylene oxide) solutions in the presence of electroosmotic counterflow

DNA fragments of 1 to 10 kbp in length were separated by capillary electrophoresis (CE), using poly(ethylene oxide) (PEO) solutions in the presence of electroosmotic flow. The technique requires filling the capillary with the polymer solution by means of electroosmotic flow (EOF). Separation times of 6–7 min in PEO solutions ranging from  $0.3$  to  $8 \times 10^6 M_r$  at  $375$  V/cm were sufficient to separate the 11 components of the dsDNA ladder (0.5 to 10 kbp) by size. The migration behavior of the double-stranded (ds)DNA fragments, interpreted by “Ferguson plot analysis”, in the system is indistinguishable from that previously reported for capillary zone electrophoresis (CZE) in a polyacrylamide solution without EOF. Potential advantages of conducting CZE using polymer solutions in the presence of EOF are: (i) Possibility of long migration times on short columns; (ii) possibility of introducing relatively viscous, high  $M_r$  polymer solutions into narrow capillaries; (iii) possibility of establishing polymer concentration gradients in capillaries; (iv) possibility of concentrating the starting zone by balancing electrophoretic migration and electroosmotic transport.

### 1 Introduction

Previous CE analyses of kbp sized dsDNA in polymer solutions at concentrations above the entanglement threshold (“networks”) have always been conducted with the suppression of EOF in solutions of uncrosslinked polyacrylamide [1, 2], poly(*N*-acrylaminoethoxyethanol [3], cellulose derivatives [4–6] and agarose derivatives [7]. EOF was suppressed by inner wall coating in all of these applications presumably to avoid the flow of polymer solution out of the capillary and to prevent interactions of the analyte and/or contaminant proteins with the inner wall of the capillary. There are, however, potential advantages to conducting electrophoretic separations in the presence of EOF especially to the “semi-dilute” regime of polymer solutions, *i.e.* the concentration range which derives its resolving power from the commensurate dimensions of the screening length of the polymer network and the diameter of the analyte [8]. Such potential advantages are: (i) Filling of the capillary by EOF in lieu of the mechanical introduction of the polymer solution, an advantage in the application to viscous polymers and narrow capillaries; (ii) effective slowdown of migration allowing for long migration times (low field strength) in short capillaries; (iii) the possibility of establishing polymer concentration gradients in capillaries; and (iv) the possibility of starting zone concentration through a balance between migration and electroosmosis. The present work tests the effectiveness of CE separations of dsDNA fragments of 0.5 to 10 kbp in length, using PEO solutions in the “semi-dilute regime” when EOF is present. The effectiveness can be maximized through pretreatment of the silicate capillaries with NaOH,

which results in a reversal in the order of size separation when EOF exceeds the electrophoretic migration rate.

### 2 Materials and methods

#### 2.1 dsDNA fragments

Kilobase dsDNA marker (Pharmacia, Cat. No. 27-4004-01, Uppsala, Sweden) was used.

#### 2.2 Polymer

Poly(ethylene oxide) of  $M_r$   $8 \times 10^6$  (No. 37283-8),  $4 \times 10^6$  (No. 18-946-4),  $1 \times 10^6$  (No. 37278-1),  $0.6 \times 10^6$  (No. 18202-8),  $0.3 \times 10^6$  (No. 18200-1) was obtained from Aldrich (Milwaukee, WI, USA). The entanglement threshold,  $c^*$ , for PEO of the various  $M_r$ 's was given by Table 2 of [8]. PEO solutions were prepared at 85–90 °C with gradual addition of the solid to the stirred solutions, followed by at least 1 h of continued magnetic stirring. The solution was degassed by house vacuum, with a suction flask positioned in a ultrasonicator, and stored at 4 °C. Degassing was repeated prior to use. Solutions were used within 1 week of preparation.

#### 2.3 CE apparatus

The basic design of the separation system has previously been described [9]. A 30 kV power supply (Gamma High Voltage Research, Ormond Beach, FL, USA) was used. The entire detection system was enclosed in a black box with a HV interlock. The high voltage end of the separation system was put in a homemade plexiglass box for safety. A 1.5-mW He-Ne laser with 543.6 nm output (Melles Griot, Irvine, CA, USA) was used for excitation. The light was collected with a 10× objective (N. A. = 0.25). One RG 610 cutoff filter was used to block scattered light before the emitted light reaches the photomultiplier tube (Hamamatsu R928, Shizuoka-ken, Japan). The fluorescence signal was directly transferred through a 10-kΩ resistor to a 24-bit A/D interface at 10 HZ (Borwin, JMBS Developments, le

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**Abbreviations:** PEO, poly(ethylene oxide); TBE, Tris-boric acid-EDTA

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Fontanil, France) and stored in a PC. Capillaries of 75  $\mu\text{m}$  ID and 365  $\mu\text{m}$  OD and 40 cm length (Polymicro Technologies, Phoenix, AZ, USA) were used without modification. Prior to use, they were filled with 0.5 M NaOH for 1 h.

## 2.4 CE procedure

The capillary was pressure injected with 0.089 M Tris, 0.089 M boric acid, 0.002 M  $\text{Na}_2\text{EDTA}$  (1 $\times$  TBE) containing 5  $\mu\text{g}/\text{mL}$  ethidium bromide (EtB). A 1/10 diluted dsDNA preparation was injected at 1 kV for 5 s. The buffer vessels were exchanged for PEO solution and CE was conducted at 375 V/cm. The EtB labeled dsDNA fragments were detected by fluorescence optics.

## 2.5 Data processing

Mobilities ( $\text{cm}^2/\text{sV}$ ) were calculated from the computer output of migration times. Non-linear Ferguson plots (log mobility vs. polymer concentration) were obtained by EXCEL program. The local slopes of those plots,  $K'_R$ , were calculated as previously defined (Section 2.3 of [10]). Geometric mean radii of dsDNA fragments were calculated by Eq. (10) of [11]. EOF was evaluated according to the time  $t$ , at which the baseline shift from that of the buffer to that of the polymer solution where

$$\text{EOF} = (l \times l_d) / (t \times V)$$

and  $l$  is the length of capillary,  $l_d$  the length to the detector, and  $V$  the voltage.

## 3 Results

### 3.1 Separation of the constituent dsDNA fragments of the kilobase ladder in the presence of EOF in PEO solution networks formed by polymers of different $M_r$

The dsDNA fragment pattern of the ladder obtained at 375 V/cm in presence of EOF in 0.5–1.0% solutions of PEO in the  $M_r$  range of 8 to  $0.3 \times 10^6$  (Panels A to E) resolves the 11 dsDNA species uniformly (Fig. 1). Since EOF under the conditions used exceeds migration rates, the order of size separation is reversed compared to that observed without EOF. Differences in migration rates between CE at various PEO concentrations, shown in Panels A to E, relate to the irreproducibility of EOF (see Section 3.2). Minor variations in the gel pattern depend on whether the PEO concentrations lie above  $c^*$  (Panels A, B and C), at (Panel D) or below  $c^*$  (Panel E). The separations in the “semi-dilute” regime apparently resolve the small species 1 to 5, while in or at the “dilute” regime peak heights, 2 to 5 decrease with increasing length of dsDNA, suggesting that the peaks of the largest components contain unresolved mixtures. In particular, peak 1 appears unresolved. Suitable concentration ranges of PEO for separation of the components of the 11-member dsDNA ladder, using PEO of different  $M_r$ 's in the presence of EOF, are listed in Table 1.

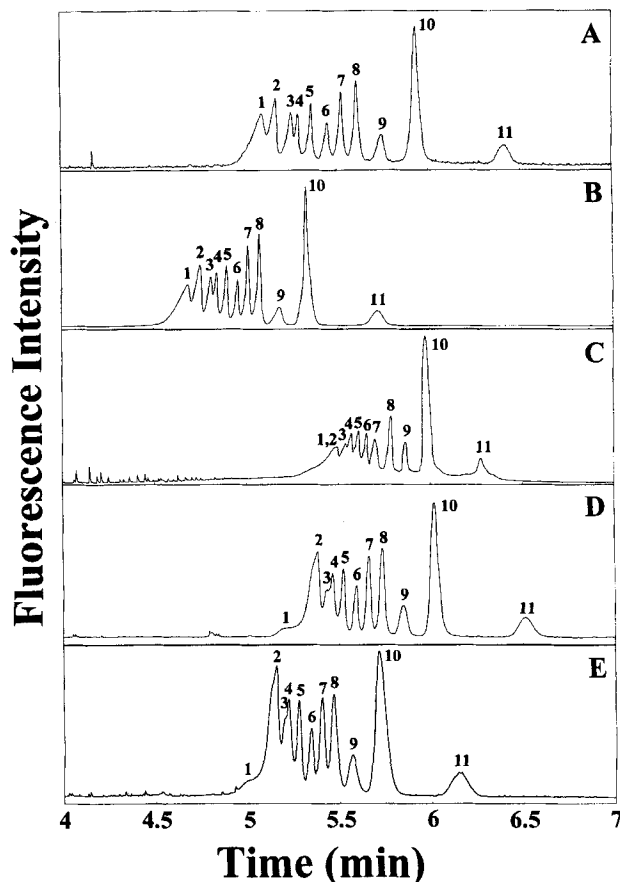


Figure 1. Representative CE separations of kbp sized dsDNA fragments in the presence of EOF, using PEO solutions: CE at 15 kV; capillary length: 40 cm; buffer: 1  $\times$  TBE containing 5  $\mu\text{g}/\text{mL}$  ethidium bromide; polymer: (A) PEO, of 0.5%,  $M_r$   $8 \times 10^6$ , (B) 0.5%,  $M_r$   $4 \times 10^6$ , (C) 0.5%,  $M_r$   $1 \times 10^6$ , (D) 1%,  $M_r$   $0.6 \times 10^6$  and (E) 0.5%,  $M_r$   $0.3 \times 10^6$ . dsDNA lengths (kbp): (1) 10, (2) 8, (3) 6, (4) 5, (5) 4, (6) 3, (7) 2.5, (8) 2, (9) 1.5, (10) 1, (11) 0.5.

Table 1. Suitable concentration ranges of PEO for separation of kbp-sized dsDNA in the presence of EOF

PEO $M_r \times 10^6$	Concentration range (%)	Separation window (min)
8	> 0.03 <sup>a)</sup> (0.5)	5.1–6.4
4	> 0.10 <sup>b)</sup> (1.0)	4.4–5.0
1	0.1–1.0 (0.5)	5.5–6.3
0.6	0.2–0.5 (1.0)	5.2–6.5
0.3	0.5–1.0 (1.0)	5.0–6.1

a) No data above 0.5% were obtained

b) No data above 1.0% were obtained

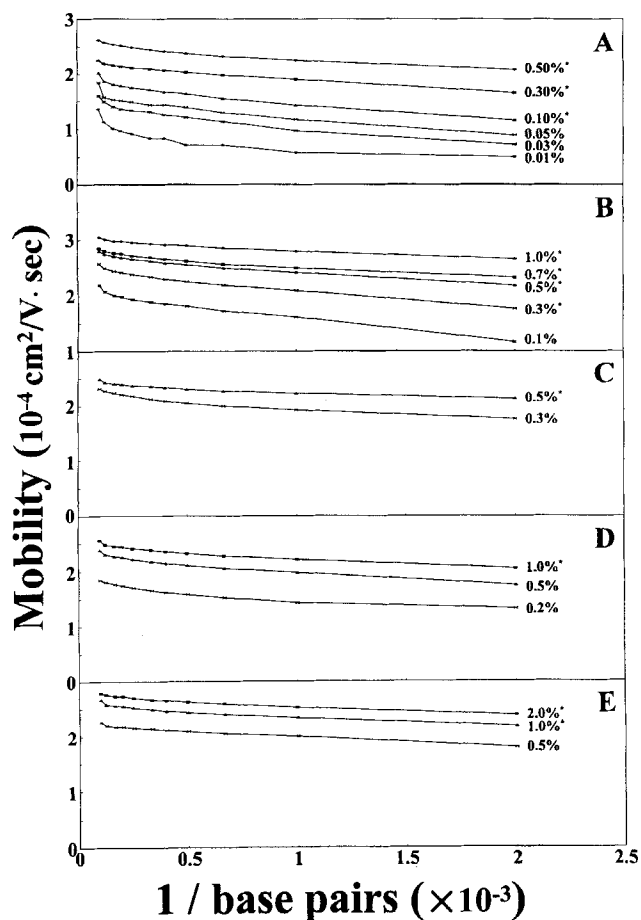
Numbers in parentheses designate the concentrations of the data in Fig. 1.

### 3.2 Short times required for separation of kbp-sized dsDNA fragments using CE in PEO solutions in the presence of EOF

The time required for separation of kbp-sized dsDNA, using CE in polymer solutions in the presence of EOF, compares favorably with separation times for previous CE separations without EOF (Table 2) even when differences in field

**Table 2.** Time required for separation of kbp-sized dsDNA, using CE in polymer solutions, in presence and in absence of EOF

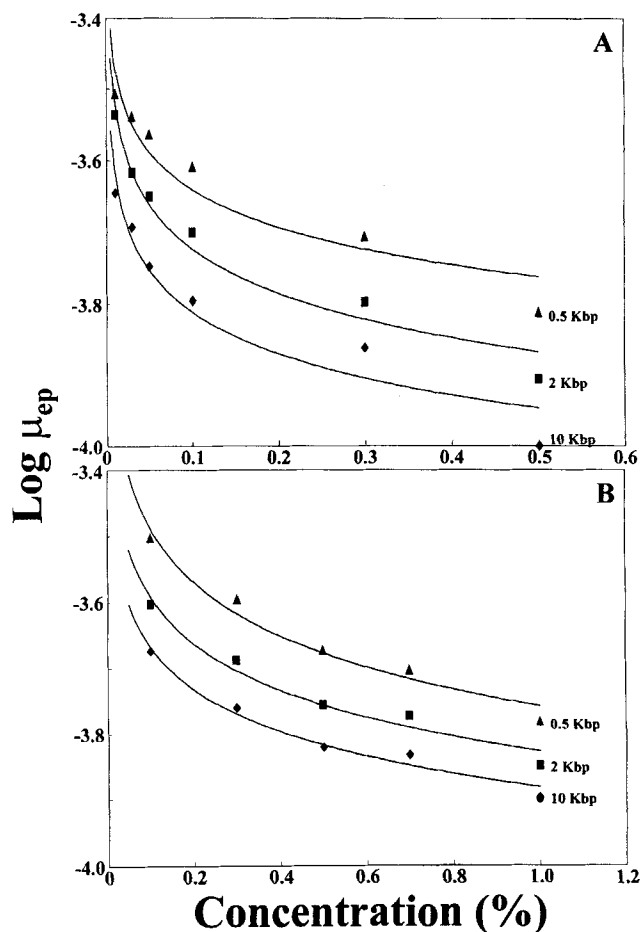
Polymer	$M_r$	Concentration (%)	dsDNA (kbp)	Separation time (min)	EOF	V/cm	Concentration regime	Reference
PEO	$8 \times 10^6$	0.5	0.5–10	< 7	+	375	Semidilute	This study
PEO	$4 \times 10^6$	1.0	0.5–10	< 6	+	375	Semidilute	This study
Polyacrylamide		4.5	0.075–12.2	120	–	100	Semidilute	[1]
Poly( <i>N</i> -acryloylamino-ethoxyethanol)		10.0	0.075–12.2	72	–	100	Semidilute	[3]

**Figure 2.** Migration rates of dsDNA fragments as a function of their length. Panel designations and CE conditions as in Fig. 1.

strength are taken into account. Note that the higher field strength used in the present study would presumably lead to failure of separation for the larger dsDNA species when EOF is suppressed.

### 3.3 Molecular sieving in CE, using PEO solutions in the presence of EOF

dsDNA mobility, corrected for EOF, decreases uniformly with increasing length of dsDNA independent of PEO  $M_r$ , in the range of 8 to  $0.3 \times 10^6$  (Fig. 2, Panels A to E). The decrease suggests a parallel molecular sieving effect when a polymer network is present in the “semi-dilute” concentration range (designated by \*) or in the “dilute” one.

**Figure 3.** Ferguson plots of dsDNA fragments in CE using PEO solutions in the presence of EOF. Panel designations and CE conditions as in Fig. 1.

### 3.4 Ferguson plot analysis of CE of dsDNA fragments in the kbp size range, using PEO solution in the presence of EOF

Mobility of dsDNA fragments, obtained in and corrected for the presence of EOF, decreases semi-hyperbolically with increasing PEO concentration when plotted semi-logarithmically (Fig. 3). The local slopes of that decreasing relationship, local retardation coefficients  $K'_R$ , decrease with polymer concentration (Fig. 4). The maximum rate of that decrease occurs at the lowest polymer concentrations and the largest dsDNA lengths.  $K'_R$  values are proportional to the geometric mean radius of the dsDNA fragments; the rate of increase of  $K'_R$  with dsDNA radius increases with PEO concentration (Fig. 5).

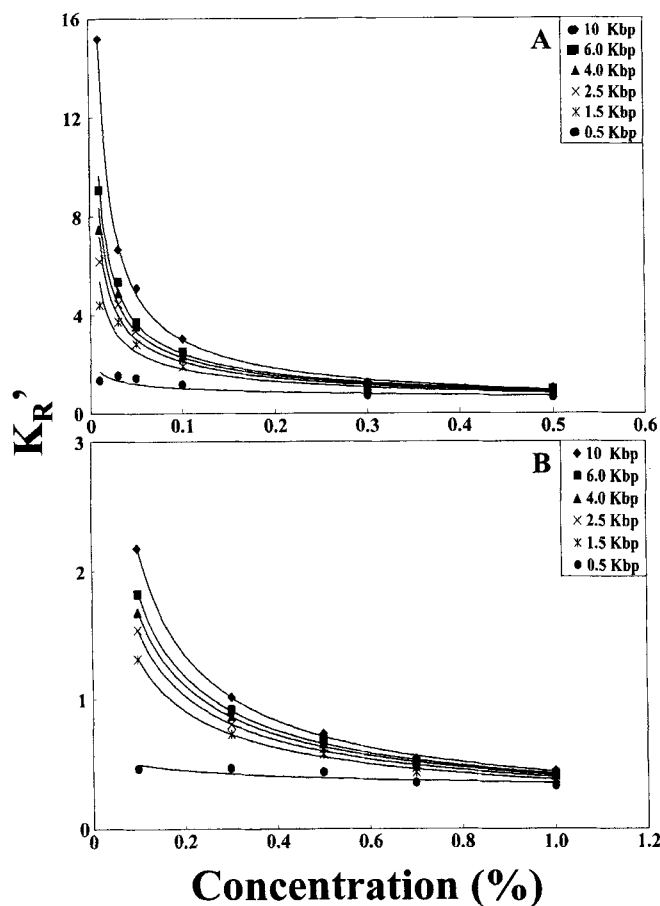


Figure 4. Local slopes of Ferguson plots as a function of PEO concentration. Conditions as in Fig. 3.

### 3.5 Reproducibility of mobilities derived from CE using polymer solutions in the presence of EOF

The migration time of the slowest migrating component (0.5 kbp) was determined within 2.8% RSD ( $n = 3$ ), using a 0.5% solution of  $8 \times 10^6 M_r$  of PEO in the presence of EOF. The reproducibility of EOF was determined as 2.3% RSD under the same conditions.

## 4 Discussion

### 4.1 Equivalence of resolving capacity for kbp-sized dsDNA fragments in the presence of EOF

Although previous reports on CE of DNA fragments in similar size ranges to those used in this study and using polymer solutions to achieve size separations (*e.g.* [1–3]) have expressed the need for suppression of EOF through wall coating. Uncoated capillaries treated with acid to protonate the silanol groups have been successfully applied to the separation of small dsDNA fragments of less than 0.6 kbp in length [12]. The present study demonstrates that when in CE under the polymer conditions of [12], EOF is maximized by alkaline treatment of the capillary to ionize the silanol groups, separations of kbp-sized dsDNA (Fig. 1) result which are qualitatively equivalent to those carried out with suppression of EOF by inner wall coating, *i.e.* Fig. 1 of

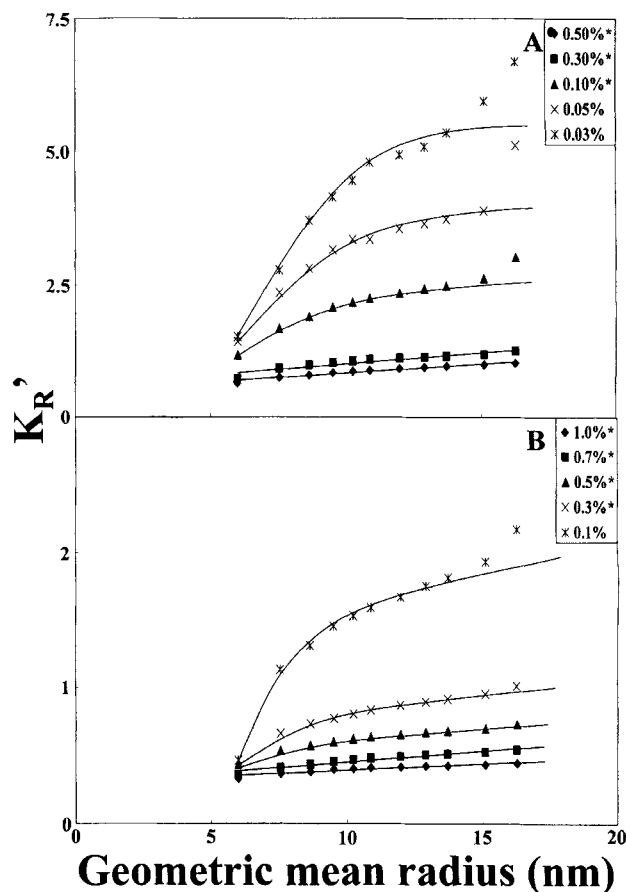


Figure 5. Proportionality between  $KR'$  and the geometric mean radius of the dsDNA fragments in CE using PEO solutions in the presence of EOF. Conditions as in Fig. 3.

[1], Fig. 1 of [3] or Fig. 1 of [6]. Similarly, the Ferguson plots shown in Fig. 3 are qualitatively equivalent to those previously reported for CE in agarose solutions (Fig. 1 of [13]). The relationships derived from the Ferguson plots *i.e.* the plot of  $K'_R$  vs. polymer concentration (Fig. 4) is qualitatively equivalent to those previously reported in Fig. 1 of [10] for polyacrylamide solutions. The plot of  $K'_R$  vs. the geometric mean radius of DNA (Fig. 5) cannot be distinguished qualitatively from that shown for polyacrylamide solutions in Fig. 2 of [10]. The linear relation between  $K_R$  and particle radius in polymer solutions has been observed in all previous applications of electrophoresis in polymer solutions (see Appendix of [10], Fig. 3 of [14]).

Band patterns, mobilities and derived functions do not indicate any difference whether they were obtained in the “semi-dilute” polymer concentration ranges (noted by an asterisk in the figures), or in the “dilute” concentration range below the entanglement threshold ( $c^*$ ). This insensitivity of the relations derived from the Ferguson plot was previously observed in the CE of particles larger than 30 nm in diameter [15]. It is presumably due to the relatively small differences within the scatter of mobility-derived parameters in the conventional plots used and has no qualitative significance.

#### 4.2 Potential advantages of conducting CE in the presence of EOF

We have discussed in the section above the fact that the presence of EOF in the CE using polymer solutions does not seem to bring about any disadvantages in separation of dsDNA fragments in the kbp-size range. However, actual advantages in the presence of EOF can be envisaged. One advantage is that the presence of EOF by increasing the time of passage through the capillary is in effect equivalent to increasing the migration path, thus allowing for relatively shorter capillaries or longer migration times. Table 2 also suggests that the time required for separation at least under the conditions of this study may be less than those previously reported. This is apparently due to the higher field strengths that can be applied to the separation of larger dsDNA species in the presence of EOF than is possible in its absence, without loss of resolution. The mechanism by which this advantage is achieved is not known. The filling of the capillary by EOF is certainly convenient. It may also be useful in the loading of capillaries with less than 75  $\mu\text{m}$  ID instead of pressure injection.

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#### 5 References

- [1] Chiari, M., Nesi, M., Righetti, P. G., *J. Chromatogr.* 1993, 652, 31–39.
- [2] Chiari, M., Nesi, M., Righetti, P. G., *Electrophoresis* 1994, 15, 616–622.
- [3] Heiger, D. N., Cohen, A. S., Karger, B. L., *J. Chromatogr.* 1990, 516, 33–48.
- [4] Braun, B., Blanch, H. W., Prausnitz, J. M., *Electrophoresis* 1997, 18, 1994–1997.
- [5] Baba, Y., Ishimaru, N., Samata, K., Tshuhako, M., *J. Chromatogr.* 1993, 653, 329–335.
- [6] Atha, D., *Electrophoresis* 1998, 19, 1428–1453.
- [7] Siles, B. A., Collier, G. B., Reeder, D. J., May, W. E., *Appl. Theor. Electrophor.* 1996, 6, 15–22.
- [8] Radko, S. P., Chrambach, A., *Biopolymers* 1997, 42, 183–189.
- [9] Chang, H.-T., Yeung, E. S., *J. Chromatogr. B* 1995, 669, 113–123.
- [10] Tietz, D., Chrambach, A., *Electrophoresis* 1993, 14, 185–190.
- [11] Stellwagen, N. C., *Adv. Electrophor.* 1987, 1, 177–230.
- [12] Fung, E. N., Yeung, E. S., *Anal. Chem.* 1995, 67, 1913–1919.
- [13] Tietz, D., Chrambach, A., *Electrophoresis* 1992, 13, 286–294.
- [14] Radko, S. P., Chrambach, A., *J. Phys. Chem.* 1996, 100, 19461–19465.
- [15] Radko, S. P., Chrambach, A., *Electrophoresis* 1996, 17, 1094–1102.