

Synthesis of Heterobimetallic Allene Complexes

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The first dinuclear complex containing a four-electron $\mu\text{-}\eta^2\text{-}\eta^2$ -allene ligand was synthesized from the reaction of allene with $[\text{Cp}(\text{CO})_2\text{Mo}]_2$ ($\text{Cp} = \eta^5\text{-C}_5\text{H}_5$) and was reported in 1977.¹ Thereafter, a number of complexes of this type were prepared,² notably by photolysis of dinuclear complexes with a bridging cyclopropylidene ligand.^{2b,c} Several dinuclear and trinuclear metal complexes containing allenyl ligands in various bonding modes have been prepared by initially using a propargyl ligand as a template.³ Despite these and other reports,⁴ methods that lead to formation of a heterodinuclear $\mu\text{-}\eta^2\text{-}\eta^2$ -allene complex are still lacking.⁵ In exploring the chemistry of a tungsten β -(chlorocarbonyl)allylic complex $\text{Cp}(\text{CO})_2\text{W}(\eta^3\text{-CH}_2\text{C}(\text{COCl})\text{CH}_2)$ (**1a**),⁶ we prepared several dinuclear complexes by replacing the Cl atom with another metal group and found photolysis of such complexes provided a systematic approach to the dinuclear $\mu\text{-}\eta^2\text{-}\eta^2$ -allene complexes. Herein, we report several complexes synthesized via this approach and their characterization.

The reaction of $\text{Cp}(\text{CO})_2\text{W}(\eta^3\text{-CH}_2\text{C}(\text{COCl})\text{CH}_2)$ (**1a**)⁶ with $\text{Cp}(\text{CO})_3\text{W}$ in THF resulted in the formation of a homonuclear complex $\text{Cp}(\text{CO})_2\text{W}[\mu\text{-}\eta^3\text{-}\eta^1\text{-CH}_2\text{C}(\text{CO})\text{CH}_2]\text{W}(\text{CO})_3\text{Cp}$ (**2a**). Heteronuclear analogues $\text{Cp}(\text{CO})_2\text{M}(\mu\text{-}\eta^3\text{-}\eta^1\text{-CH}_2\text{C}(\text{CO})\text{CH}_2)\text{M}'$ (**2b**, $\text{M} = \text{Mo}$, $\text{M}' = \text{W}(\text{CO})_3\text{Cp}$; **2c**, $\text{M} = \text{W}$, $\text{M}' = \text{Fe}(\text{CO})_2\text{Cp}$; **2d**, $\text{M} = \text{W}$, $\text{M}' = \text{Re}(\text{CO})_5$) were prepared similarly. However, attempts to prepare similar complexes using the $\text{Cp}(\text{CO})_3\text{Mo}$ anion were unsuccessful. We believe that metal anions with weak nucleophilicity⁷ may not be applicable for the preparation of **2** employing this synthetic strategy. On the basis of the EAN rule, the two metal centers in **2** bridged by a $\mu\text{-}\eta^3\text{-}\eta^1$ - β -carbonylallylic ligand should have no M–M bond. Complexes **2** were characterized through a combination of microanalytical data and IR, ¹H NMR, and ¹³C NMR spectroscopy.⁸ For example, in the ¹H NMR spectrum of **2a** at room temperature, the two inequivalent Cp groups displayed a sharp resonance and a broad resonance and the allylic *syn* and *anti* protons appeared as two broad resonances at δ 2.76 and 1.38, respectively. These broad resonances are due to the fluxional behavior of the *endo* and *exo* configurations of the allylic group. At 240 K, the two broad resonances at δ 2.76 and 1.38 of **2a** were resolved into two sets of allylic protons at δ 2.78, 1.16 and 2.68, 1.58,⁸ assignable to those of the *endo* and *exo* isomers, respectively.

Photolysis of complexes **2a–c** with UV irradiation in benzene resulted in the loss of two CO ligands and afforded the dinuclear

allene complexes $\text{Cp}(\text{CO})_2\text{M}(\mu\text{-}\eta^2\text{-}\eta^2\text{-CH}_2\text{C}=\text{C}=\text{CH}_2)\text{M}'$ (**4a**, $\text{M} = \text{W}$, $\text{M}' = \text{W}(\text{CO})_2\text{Cp}$; **4b**, $\text{M} = \text{Mo}$, $\text{M}' = \text{W}(\text{CO})_2\text{Cp}$; **4c**, $\text{M} = \text{W}$, $\text{M}' = \text{Fe}(\text{CO})\text{Cp}$); see Scheme I. Complexes **4** were characterized by spectroscopic methods. In addition, the molecular structure of **4c** was determined by an X-ray diffraction analysis.⁹ In each unit cell, there are two independent molecules which are an enantiomeric pair and differ only slightly in their bond lengths and angles. Figure 1 shows one of the molecules. As expected, the C₃ unit is nonlinear with a C–C–C bond angle of 140(4)°. The three-carbon allene unit sits above the W–Fe axis in a criss-cross arrangement. This geometry allows interaction of one π -bond of the allene with the W center and of the second π -bond of the allene with the Fe center. With such a structure, the heteronuclear system should display two sets of inequivalent geminal protons, i.e. proximal and distal with respect to the M–M' bond. Indeed, in the ¹H NMR spectra of heteronuclear allene complexes, four allene protons displayed four multiplet resonances (δ 4.50, 3.02, 2.66, 2.40 for **4c** and δ 4.22, 3.69, 2.67, 2.45 for **4b**). In contrast, in the homonuclear system,¹ there is a C₂ axis passing through the center carbon of the allene ligand and the midpoint of the M–M vector. Therefore, in the ¹H NMR spectrum of **4a**, a deceptively simple triplet pattern for the methylene protons was observed.

In the photolytic decarbonylation reactions of **2a** and **2b**, an intermediate formulated as $\text{Cp}(\text{CO})_2\text{M}(\mu\text{-}\eta^3\text{-}\eta^1\text{-CH}_2\text{CCH}_2)\text{W}(\text{CO})_3\text{Cp}$ (**3a**, $\text{M} = \text{W}$; **3b**, $\text{M} = \text{Mo}$) was observed. Presumably, the two metal centers in **3** are bridged by a $\eta^3\text{-}\eta^1$ -allylic ligand as shown in the Scheme I.¹⁰ Complexes **3a** and **3b** were observed but not isolated.⁸ Interestingly, for rhenium complex **2d**, only one decarbonylation was attained and the complex $\text{Cp}(\text{CO})_2\text{W}(\mu\text{-}\eta^3\text{-}\eta^1\text{-CH}_2\text{CCH}_2)\text{Re}(\text{CO})_5$ (**3d**) was isolated in 45% yield. In the ¹H NMR spectrum of **3d**, the resonances of the allylic protons at δ 3.09, 1.67 are typical for *syn/anti* resonances of an allylic system. The mass spectrum of **3d** displayed parent peaks as well as fragmentations due to loss of CO.⁷ W–Re metal–metal bonds have been observed in several complexes.¹¹ Reasons for no further decarbonylation of **3d** are not clear.

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- (8) Preparations and spectroscopic data of these complexes are given in the supplementary material.

- (9) Yellow crystals of **4c** were grown by slow diffusion of *n*-hexane into a dichloromethane solution of **4c** at 4 °C. Crystal data: space group $P2_12_12_1$, $a = 7.606(2)$ Å, $b = 12.333(6)$ Å, $c = 31.072(7)$ Å, $Z = 4$. X-ray data were collected at room temperature on an Enraf-Nonius CAD4 diffractometer using graphite-monochromated Mo $K\alpha$ radiation. A total of 2205 reflections were collected with use of the ω - 2θ scanning technique between 2.0 and 49.9° in 2θ . A total of 1553 reflections with $I > 2\sigma(I)$ were used in the full-matrix least-squares refinement. The structure was solved by using heavy-atom methods with the NRCC computing package. The final values of the agreement indices were $R = 0.062$, $R_w = 0.060$, and GOF = 2.20.
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- (12) Yellow crystals of **5** were grown by slow diffusion of *n*-hexane into a dichloromethane solution of **5** at 4 °C. Crystal data: space group $P2_1/c$, $a = 8.681(3)$ Å, $b = 13.420(5)$ Å, $c = 17.630(9)$ Å, $\beta = 103.48(4)^\circ$, $Z = 4$. X-ray data were collected at room temperature on an Enraf-Nonius CAD4 diffractometer using graphite-monochromated Mo $K\alpha$ radiation. A total of 3514 reflections were collected with use of the ω - 2θ scanning technique between 2.0 and 49.9° in 2θ . A total of 2408 reflections with $I > 2\sigma(I)$ were used in the full-matrix least-squares refinement. The structure was solved by using heavy-atom methods with the NRCC computing package. The final values of the agreement indices were $R = 0.044$, $R_w = 0.032$, and GOF = 2.43.

Scheme I

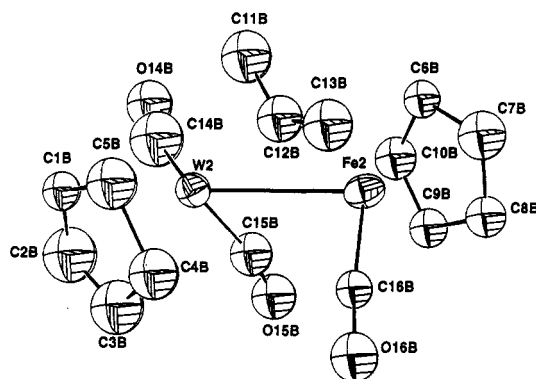
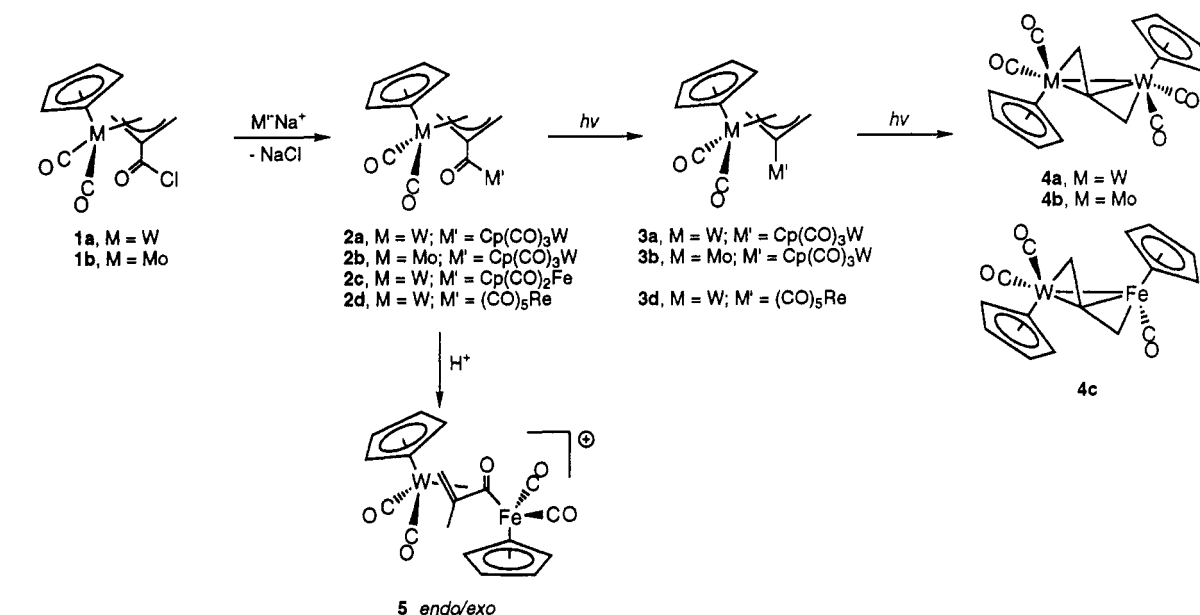


Figure 1. ORTEP drawing of $\text{Cp}(\text{CO})_2\text{W}(\mu\text{-}\eta^2, \eta^2\text{-CH}_2\text{=C=CH}_2)\text{Fe}(\text{CO})\text{Cp}$ (**4c**) with thermal ellipsoids shown at the 50% probability level. Selected bond distances (Å) and bond angles (deg) are as follows: Fe–C(12B), 1.91(4); Fe–C(13B), 2.08(5); W–C(11B), 2.34(4); W–C(12B), 2.07(4); C(12B)–C(13B), 1.34(6); C(11B)–C(12B), 1.45(6); C(11B)–C(12B)–C(13B), 140(4).

Treatment of **2c** with HBF_4 resulted in the formation of a cationic enone complex $[\text{Cp}(\text{CO})_2\text{W}(\mu\text{-}\eta^4, \eta^1\text{-CH}_2\text{=C}(\text{CH}_3)\text{CO})\text{Fe}(\text{CO})_2\text{Cp}]\text{BF}_4$ (**5**).⁸ Complex **5** in solution gave a mixture of *endo/exo* isomers with a ratio of 6.4:1. The structure of **5** was also confirmed by X-ray diffraction analysis.¹² As shown in Figure 2, the enone group adopts an *s-cis* configuration and is in an *endo* conformation with respect to the Cp group.¹³ The η^4 -coordination of the enone group is revealed by the one W–O and three W–C bonding distances. A slightly longer W–C(2) separation of 2.49(1) Å as compared to the other W–C separations (2.24(1) and 2.28(1) Å) may be attributed to the steric effect of the iron moiety and to a relatively short W–O(3) distance of 2.141(7) Å. For a cationic molybdenum butadiene complex,¹⁴ four isomers (*endo/exo* and *cis/trans*) have been observed at low temperature.

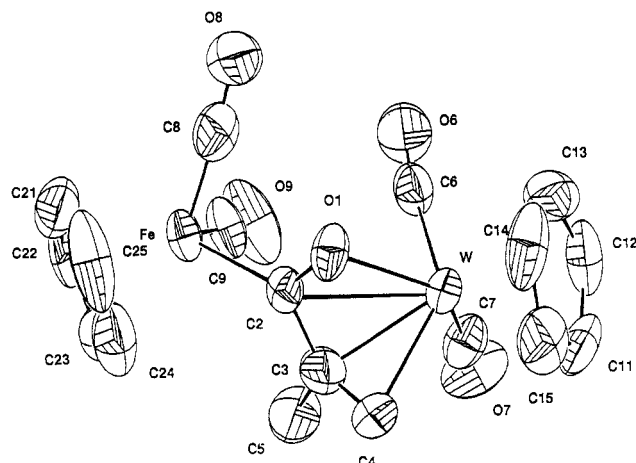


Figure 2. ORTEP drawing of $\text{Cp}(\text{CO})_2\text{W}(\mu\text{-}\eta^4, \eta^1\text{-CH}_2\text{=C}(\text{CH}_3)\text{-CH=O})\text{Fe}(\text{CO})_2\text{Cp}$ (**5**) with thermal ellipsoids shown at the 50% probability level. Selected bond distances (Å) and bond angles (deg) are as follows: Fe–C(2), 1.946(1); W–O(1), 2.141(7); W–C(2), 2.49(1); W–C(3), 2.28(1); W–C(4), 2.24(1); C(2)–O(1), 1.32(1); C(2)–C(3), 1.43(2); C(3)–C(4), 1.42(2); C(3)–C(5), 1.49(2); C(2)–C(3)–C(4), 115(1); O(1)–C(2)–C(3), 110.7(9); C(4)–C(3)–C(5), 121(1).

In our case, the *s-trans* configuration was not observed probably due to the steric bulk of the metal moiety.

Using β -(chlorocarbonyl)allylic metal complexes as starting materials, we have demonstrated the preparation of a heteronuclear complex with a β -(carbonyl)allylic bridging ligand. Photolysis of this product led efficiently to the heteronuclear $\mu\text{-}\eta^2, \eta^2$ -allene complex. Reactivity of allene complexes with heteronuclear metal centers is currently under investigation.

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Supplementary Material Available: A textual presentation of the syntheses and spectroscopic data for **2a**, **2b**, **2c**, **2d**, **3b**, **4b**, **4c**, and **5** and tables of X-ray crystallographic data, positional parameters, anisotropic thermal parameters, and selected bond distances and angles for **4c** and **5** (11 pages). Ordering information is given on any current masthead page.

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