# 2-(N-Methylanilino)-2-phenylsulfanylacetonitrile, A Reagent Tested for Electrophilic, Nucleophilic and Radical Reactions 

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2-( $N$-Methylanilino)-2-phenylsulfanylacetonitrile 1 has been readily prepared from 2-( $N$ methylanilino) acetonitrile and diphenyl disulfide. Alkylation of the anion of 1 with halogenoalkanes resulted in concurrent elimination of benzenethiol to give conjugated $\alpha$-aminoalkenenitriles of $2 E$ configuration. Autoxidation of 1 in the presence of alkoxide ions afforded alkyl $N$-methyl- $N$ phenylcarbamates. Nucleophilic substitution of 1 with Grignard reagent or appropriate silyl compounds were promoted by Cul or Lewis acids to give varied $\alpha$-amino nitriles. The 4 -oxo- 2 amino nitriles 9 obtained by condensation of 1 and titanium enolates can be considered as derivatives of 1,3 -dicarbonyl compounds with the aldehyde group being activated to give an amino nitrile umpolung. When 1 was treated with tributylstannane, the corresponding amino nitrile $\alpha$ radical was formed and the self-coupling product was isolated.

Heteroatom-substituted acetonitriles such as cyanohydrins, amino nitriles and sulfanylacetonitriles are versatile reagents in synthetic chemistry. ${ }^{1}$ We reported recently as preliminary
communications ${ }^{2}$ the properties and use of $2-(N$-methyl-anilino)-2-phenylsulfanylacetonitrile 1. Compound 1, m.p. 61$61.5^{\circ} \mathrm{C}$, is readily prepared from the reaction of the $\alpha$-anion

of 2-( $N$-methylanilino)acetonitrile ${ }^{3}$ and diphenyl disulfide. It is stable, and no apparent decomposition occurs after storage for months under a nitrogen atmosphere at room temperature. In addition to the electrophilic reactions, we also studied the nucleophilic and radical reactions of 1 . The results are delineated in Scheme 1 and details are described as follows.

## Results and Discussion

Treatment of 1 with equimolar amounts of $\mathrm{Bu}^{t} \mathrm{OK}$ and halogenoalkanes in THF solution either at ambient temperature or with gentle heating gave the $\alpha$-aminoalkenenitriles 2am . The reaction is believed to proceed through the tandem alkylation of $1 \alpha$-anion and elimination of the benzenethiol. The anti elimination of the benzenethiol molecule can be promoted by the electron-donating amino group ${ }^{4}$ to give 2 with the $2 E$ configuration. If allylic bromides or allylic chlorides were used instead of iodoalkanes, the $\alpha$-amino dienenitriles were obtained.

Compound 1 was converted into the carbamates 3a-d by autoxidation in the presence of alkoxide ions and CuI. When Bu'OK was used as the base, the autoxidation of 1 gave the cyanoformamide 4 and the thiocarbamate 5 in 54 and $24 \%$ yields, respectively. The mechanism for the autoxidation of 1 to $3-5$ is not clear, though the reaction may be accounted for by substitution of the $\alpha$-proton, with an epoxy anion as a postulated intermediate. ${ }^{5}$
To use 1 as an equivalent of the cation of 2-( $N$-methylanilino)acetonitrile, we carried out the nucleophilic substitutions with organometallic or silyl compounds in the presence of CuI or Lewis acid. The reactions of 1 with Grignard reagent or BuLi were promoted by CuI to give amino nitriles $\mathbf{6 a - c}$ in high yields. No further substitution of the cyano group as that occurring in Bruylants reaction ${ }^{6}$ was observed. The reaction of 1 with allylsilane was effected by $\mathrm{TiCl}_{4}$ to give 2-( $N$-methylanilino)-pent-4-enenitrile 7 in $77 \%$ yield. ${ }^{7}$ The allylation was also realized by the catalysis of $\mathrm{AgBF}_{4}$, albeit in lower yield ( $54 \%$ ). Substitution of 1 with the titanium enolates generated from the silyl ketene acetal or ethyl ketones gave compounds 8 and $9 \mathbf{9}, \mathbf{b}^{8}{ }^{8}$ The amino nitrile-substituted ester and ketone have not been prepared by conventional methods. Compounds 9 can be considered as derivatives of 1,3-dicarbonyl compounds with the aldehyde group being activated to give an amino nitrile umpolung ${ }^{9}$ that may be used in organic synthesis.

Treatment of 1 with tributyltin hydride yielded 2-( $N$-methylanilino) acetonitrile and 2,3-bis( $N$-methylanilino)butane-1,4-dinitrile 10. These products are conceivably derived from an amino nitrile $\alpha$-radical intermediate by hydrogen abstraction or dimerization. Attempts to trap the radical intermediate with alkenes such as hexene, styrene or methyl acrylate failed, presumably the captodative radical ${ }^{10}$ being too stable to react with either electron-rich or electron-deficient alkenes.

The amino nitrile $\alpha$-radical generated from 11, however, underwent intramolecular cyclization effectively to give the 2-cyanoindoline 12 in $91 \%$ yield.


11


12

In summary, the title reagent $\mathbf{1}$ is useful for the preparation of $\alpha$-amino nitrile alkenes and dienes via tandem alkylationdehydrosulfanylation. The reaction is initiated by forming an $\alpha-$ carbanion, which can also be trapped with oxygen to give the
corresponding carbamate derivatives. Alternatively, $\mathbf{1}$ is used as an equivalent of an amino nitrile $\alpha$-cation to react with nucleophiles. The amino nitrile $\alpha$-radical can be generated from 1, though its reaction with alkenes failed.

## Experimental

Melting points (Yanaco micro melting point apparatus) are uncorrected. Elemental analyses were carried out on a PerkinElmer 240c or Hereaus CHN-O-RAPID elemental analyzer. IR spectra were run on a Perkin-Elmer 983G IR spectrophotometer. The ${ }^{1} \mathrm{H}$ NMR spectra were recorded at 200 or 300 MHz (Bruker AC-200 or AM-300WB spectrophotometer). Tetramethylsilane was used as internal standard ( $J$ values in Hz ). ${ }^{13} \mathrm{C}$ NMR spectra were recorded at 50 or 75 MHz . The mass spectra were recorded (using a Finnigan TSQ46c spectrometer) at an ionizing voltage of 70 or 20 eV . The high-resolution mass spectra (HRMS) were recorded on a JEOL JMS-HX 110 spectrometer. HPLC was carried out on a Hitachi L-6200 chromatograph using a $\mu$-Porasil column ( $7 \mathrm{~mm}, 25 \times 0.78 \mathrm{~cm}$ ) with $5 \mathrm{~cm}^{3} \mathrm{~min}^{-1}$ flow rate of elution.

2-(N-Methylanilino)-2-phenylsulfanylacetonitrile 1.-A solution of LDA was prepared by addition of BuLi $(1.33 \mathrm{~mol}$ $\mathrm{dm}^{-3}$ of hexane solution; $5 \mathrm{mmol}, 3.75 \mathrm{~cm}^{3}$ ) to diisopropylamine ( $5 \mathrm{mmol}, 0.75 \mathrm{~cm}^{3}$ ) in THF ( $10 \mathrm{~cm}^{3}$ ). A THF solution ( $3 \mathrm{~cm}^{3}$ ) of $2-(N$-methylanilino) acetonitrile ( $730 \mathrm{mg}, 5 \mathrm{mmol}$ ) was then added dropwise at $-40^{\circ} \mathrm{C}$ to the preceding solution and the mixture was stirred for 20 min ; a THF solution ( $3 \mathrm{~cm}^{3}$ ) of diphenyl disulfide ( $1.2 \mathrm{~g}, 5 \mathrm{mmol}$ ) was then added to it. After 45 min , the reaction was quenched by addition of saturated aqueous $\mathrm{NH}_{4} \mathrm{Cl}$. The volatile components of the mixture were removed by rotary evaporation, and the residue was extracted with EtOAc. The combined extracts were washed with $5 \%$ aqueous NaOH , dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated and the residue was purified by silica-gel column chromatography ( EtOAc -hexane, $1: 9$ ) to give $1(1.17 \mathrm{~g}, 91 \%$ ) as colourless crystals, m.p. $61-61.5^{\circ} \mathrm{C}$ (from hexane); $v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 2237$ $(\mathrm{CN}) ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 3.05\left(\mathrm{~s}, \mathrm{NCH}_{3}\right), 5.80(\mathrm{~s}), 6.78-7.02(3 \mathrm{H}, \mathrm{m})$, $7.18-7.42(5 \mathrm{H}, \mathrm{m})$ and $7.48-7.62(2 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 35.5(\mathrm{q})$, 61.7 (d), 116.5 (s, CN), 121.5 (d), 129.5 (d), 129.6 (d), 129.9 (d) 135.7 (s) and $146.7(\mathrm{~s}) ; m / z 254\left(\mathrm{M}^{+}, 2 \%\right), 145$ (100) (Found: C, $70.8 ; \mathrm{H}, 5.6 ; \mathrm{N}, 11.05 . \mathrm{C}_{15} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{~S}$ requires $\mathrm{C}, 70.83 ; \mathrm{H}, 5.55$; N, $11.01 \%$ ).
( N -Methylanilino)prop-2-enenitrile 2a. ${ }^{11}$-To a solution of $\mathrm{Bu}{ }^{t} \mathrm{OK}(113 \mathrm{mg}, 1 \mathrm{mmol})$ in THF ( $8 \mathrm{~cm}^{3}$ ) was added a solution of 2-( $N$-methylanilino)-2-phenylsulfanylacetonitrile ( $254 \mathrm{mg}, 1$ $\mathrm{mmol})$ in THF ( $2 \mathrm{~cm}^{3}$ ) at room temperature. After 20 min , the resulting pale yellow solution was treated with iodomethane ( $1.1 \mathrm{mmol}, 0.07 \mathrm{~cm}^{3}$ ). The brownish yellow turbid mixture was stirred at room temperature for $1-12 \mathrm{~h}$ and then quenched with saturated aqueous $\mathrm{NH}_{4} \mathrm{Cl}$. After removal of THF from the mixture under reduced pressure, the residue was extracted with EtOAc. The combined extracts were concentrated and passed through a column of silica gel to give the desired product $\mathbf{2 a}(131 \mathrm{mg}, 83 \%)$. The spectral data were previously described. ${ }^{11}$

2-( N -Methylanilino)but-2-enenitrile $\mathbf{2 b}$.-The reaction of $\mathbf{1}$ and iodoethane, by a procedure similar to that for $\mathbf{2 a}$, gave $\mathbf{2 b}$ in $82 \%$ yield. A mixture of $E$ - and $Z$-isomers ( $6: 1$ ); liquid, TLC (EtOAc-hexane, 2:98) $R_{\mathrm{f}} 0.2 ; v_{\max }$ (neat) $/ \mathrm{cm}^{-1} 2213(\mathrm{CN}) ; m / z$ $172\left(\mathrm{M}^{+}, 100 \%\right), 163(10)$ and $157(50) ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right)$ 1.98/1.72 (3 $\mathrm{H}, \mathrm{d}, J 7.2), 3.06(3 \mathrm{H}, \mathrm{s}), 5.94 / 6.31(1 \mathrm{H}, \mathrm{q}, J 7.2), 6.85-7.05(3 \mathrm{H}$, $\mathrm{m})$ and $7.20-7.40(2 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 14.7 / 13.5(\mathrm{C}-4), 39.6 / 37.6$ $\left(\mathrm{NCH}_{3}\right), 114.4 / 114.1(\mathrm{CN}), 116.2(Z), 118.4,119.6(Z), 120.3$ $(Z), 121.6(E), 121.7(Z), 122.3(E), 126.4(E), 127.6(E), 128.9$
(E), $129.0(Z), 129.1(Z) 129.3(E), 129.8(E), 139.8(Z), 145.8$ and 146.5 (Found: $\mathrm{M}^{+}, 172.0984$. Calc. for $M, 172.1000$ ).

2-( N -Methylanilino)hex-2-enenitrile $\mathbf{2 c} .^{12}$-The reaction of $\mathbf{1}$ and iodobutane, by a procedure similar to that for 2 a , gave $\mathbf{2 c}$ ( $E$-configuration) as a liquid in $61 \%$ yield. The spectral data were previously described; ${ }^{12}$ liquid, $v_{\text {max }}$ (neat) $/ \mathrm{cm}^{-1} 2240(\mathrm{CN})$ and $1610 ; m / z 200\left(\mathrm{M}^{+}, 35 \%\right)$ and $171(100) ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.00(3$ $\mathrm{H}, \mathrm{t}, J 6.1), 1.53(2 \mathrm{H}, \mathrm{m}), 2.40(2 \mathrm{H}, \mathrm{dt}, J 8.4,8.0), 3.12(3 \mathrm{H}, \mathrm{s})$, $5.91(1 \mathrm{H}, \mathrm{t}, J 8.4), 6.90-7.05(3 \mathrm{H}, \mathrm{m})$ and $7.20-7.40(2 \mathrm{H}, \mathrm{m})$.

5-Bromo-2-( N -methylanilino)pent-2-enenitrile 2d.-The reaction of 1 and 1 -bromo- 3 -iodopropane, by a procedure similar to that for 2 a except that heating ( $60^{\circ} \mathrm{C}, 24 \mathrm{~h}$ ) was applied, gave $\mathbf{2 d}$ ( $E$-configuration) in $62 \%$ yield; liquid, TLC (EtOAc-hexane, 3:97), $R_{\mathrm{f}} 0.3 ; v_{\max }(\mathrm{neat}) / \mathrm{cm}^{-1} 2238(\mathrm{CN})$ and $1600 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right)$ $2.95(2 \mathrm{H}, \mathrm{dt}, J 8.1,7.9), 3.16(3 \mathrm{H}, \mathrm{s}), 3.47(2 \mathrm{H}, \mathrm{t}, J 7.9), 5.65(1$ $\mathrm{H}, \mathrm{t}, J 8.1), 7.02-7.15(2 \mathrm{H}, \mathrm{m})$ and 7.20-7.40 ( $3 \mathrm{H}, \mathrm{m}$ ); m/z 266 $(20 \%), 264\left(\mathrm{M}^{+}, 20\right)$ and $171(100)$ (Found: $\mathrm{M}^{+}, 264.0258$. Calc. for $M, 264.0262$ ).

7-Iodo-2-( N -methylanilino)hept-2-enenitrile $\mathbf{2 e}$.-The reaction of $\mathbf{1}$ and 1,5 -diiodopentane, by a procedure similar that for 2a, gave 2 e ( $E$-configuration) in $50 \%$ yield; liquid, TLC (EtOAchexane, 2:98), $R_{\mathrm{f}} 0.3 ; \nu_{\max }($ neat $) / \mathrm{cm}^{-1} 2221(\mathrm{CN})$ and 1593; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.62(2 \mathrm{H}, \mathrm{m}), 1.90(2 \mathrm{H}, \mathrm{m}), 2.45(2 \mathrm{H}, \mathrm{dt}, J 7.1,6.8)$, $3.12(3 \mathrm{H}, \mathrm{s}), 3.2(2 \mathrm{H}, \mathrm{t}, J 7.0), 5.77(1 \mathrm{H}, \mathrm{t}, J 7.1), 6.8-7.12(2 \mathrm{H}$, $\mathrm{m})$ and $7.20-7.40(3 \mathrm{H}, \mathrm{m}) ; m / z 340\left(\mathrm{M}^{+}, 20 \%\right), 171$ ( 100 ); (Found: $\mathrm{M}^{+}, 340.0426$. Calc. for $M, 340.0438$ ).

Ethyl 3-(N-Methylanilino)-3-cyanoprop-2-enoate 2f.--The reaction of 1 and ethyl iodoacetate, by a procedure similar to that for 2 a except that heating $\left(60^{\circ} \mathrm{C}, 24 \mathrm{~h}\right)$ was applied, gave 2 f ( $E$-configuration) in $72 \%$ yield; liquid, TLC (EtOAc-hexane, $15: 85), R_{\mathrm{f}} 0.2 ; v_{\text {max }}($ neat $) / \mathrm{cm}^{-1} 2223(\mathrm{CN})$ and $1700 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right)$ $1.30(3 \mathrm{H}, \mathrm{t}, J 8.1), 3.35(3 \mathrm{H}, \mathrm{s}), 4.21(2 \mathrm{H}, \mathrm{q}, J 8.1), 5.28(1$ $\mathrm{H}, \mathrm{s}), 7.12-7.25(2 \mathrm{H}, \mathrm{m})$ and $7.30-7.50(3 \mathrm{H}, \mathrm{m}) ; m / z 230$ $\left(\mathrm{M}^{+}, 100 \%\right)$ and 157 (96) (Found: $\mathrm{M}^{+}, 230.1053$. Calc. for M, 230.1054).

2-( N -Methylanilino)-3-phenylprop-2-enenitrile 2g. ${ }^{12}$ - The reaction of 1 and benzyl bromide, by a procedure similar to that for 2a, gave $\mathbf{2 g}$ ( $E$-configuration) as a liquid in $64 \%$ yield. The spectral data have been described earlier. ${ }^{12}$

2-( N -Methylanilino)penta-2,4-dienenitrile $\mathbf{2 h} .^{13}$-The reaction of 1 and allyl bromide, by a procedure similar to that for 2a, gave 2 h ( $2 E$-configuration) in $61 \%$ yield; liquid, HPLC (EtOAc-hexane, 2:98) $t_{\mathrm{R}} 4.8 \mathrm{~min} ; v_{\text {max }}($ neat $) / \mathrm{cm}^{-1} \quad 2223$; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 3.21(3 \mathrm{H}, \mathrm{s}), 5.21(1 \mathrm{H}, \mathrm{brd}, J 10.2), 5.31(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J$ 16.8), $6.13(1 \mathrm{H}, \mathrm{d}, \mathrm{J} 11.1), 6.58-6.77(1 \mathrm{H}, \mathrm{m}), 7.04-7.15(3 \mathrm{H}, \mathrm{m})$ and $7.26-7.37(2 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 40.3\left(\mathrm{NCH}_{3}\right), 114.3(\mathrm{CN})$, 118.4 (C-5), 122.8 (C-4), 123.1 (C-3), 124.5 (d), 129.3 (d), 132.3 (d), 138.7 (C-2) and 145.7 (s); $m / z 183\left(100 \%, \mathbf{M}^{+}-1\right)$ and 168 (25) (Found: C, 77.9; H, 6.6; N, 15.2. $\mathrm{C}_{12} \mathrm{H}_{12} \mathrm{~N}_{2}$ requires C, 78.23; H, 6.57; N, $15.20 \%$ ).

2-( N -Methylanilino)-5-phenylpenta-2,4-dienenitrile 2i. ${ }^{14}{ }^{14}$ The reaction of 1 and cinnamyl chloride, by a procedure similar to that for 2 a except that heating ( $60^{\circ} \mathrm{C}, 48 \mathrm{~h}$ ) was applied, gave $\mathbf{2 i}$ ( $2 E, 4 E$-configuration) as a liquid in $63 \%$ yield. The spectral data have been described earlier. ${ }^{14}$

2-( N -Methylanilino) hexa-2,4-dienenitrile $\mathbf{2 j} .^{14}$-The reaction of $\mathbf{1}$ and crotyl chloride, by a procedure similar to that for $\mathbf{2 a}$ except that heating $\left(60^{\circ} \mathrm{C}, 24 \mathrm{~h}\right)$ was applied, gave $2 \mathrm{j}(2 E, 4 E-$ and $2 Z, 4 E$-isomers, $2: 1$ ) as a liquid in $61 \%$ yield. The spectral data have been described earlier. ${ }^{14}$

2-( N -Methylanilino)-5-bromopenta-2,4-dienenitrile $2 \mathbf{k}$.-The reaction of 1 and 1,3 -dibromopropene, by a procedure similar to that for 2 a , gave $\mathbf{2 k}$ in $72 \%$ yield; liquid, TLC (EtOAchexane, 5:95), $R_{\mathrm{f}} 0.3$; $v_{\text {max }}$ (neat) $/ \mathrm{cm}^{-1} 2227$ (CN) and 1587; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 3.22(3 \mathrm{H}, \mathrm{s}), 5.93(1 \mathrm{H}, \mathrm{d}, J 11.1), 6.36(1 \mathrm{H}, \mathrm{d}, J$ 13.4), $7.02-7.25(4 \mathrm{H}, \mathrm{m})$ and $7.30-7.45(2 \mathrm{H}, \mathrm{m}) ; m / z 264(20 \%)$, $262\left(\mathrm{M}^{+}, 20\right)$ and $183(100)$ (Found: $\mathrm{M}^{+}, 262.0100$. Calc. for $M$, 262.0106).

## 2-( N -Methylanilino)-6-phenylsulfanylhexa-2,4-dienenitrile

 21.-The reaction of 1 and 1,4 -dibromobut-2-ene, by a procedure similar to that for 2a, gave 21 in $69 \%$ yield. The reaction involved a counterattack of benzenethiolate ion. $2 E, 4 E$-Isomer, liquid; $v_{\text {max }}$ (neat) $/ \mathrm{cm}^{-1} 2225(\mathrm{CN})$ and 1578 ; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 3.16(3 \mathrm{H}, \mathrm{s}), 3.65(2 \mathrm{H}, \mathrm{dd}, J 7.0,6.1), 5.80(1 \mathrm{H}, \mathrm{dt}, J$ 14.6, 7.6), $6.05(1 \mathrm{H}, \mathrm{d}, J 11.2), 6.48(1 \mathrm{H}, \mathrm{dd}, J 14.6,11.2)$ and 7.02-7.40 ( $10 \mathrm{H}, \mathrm{m}$ ); m/z $306\left(\mathrm{M}^{+}, 44 \%\right)$ and 197 (100) (Found: $\mathrm{M}^{+}, 306.1199$. Calc. for $M, 306.1191$. Found: C, 74.3; H, 5.9; $\mathrm{N}, 9.2 . \mathrm{C}_{19} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{~S}$ requires $\mathrm{C}, 74.47 ; \mathrm{H}, 5.92 ; \mathrm{N}, 9.14 \%$ ).
## 2-( N -Methylanilino)-5-phenylsulfanylpenta-2,4-dienenitrile

$\mathbf{2 m}$.-The reaction of $\mathbf{1}$ and 3-bromo-1-(trimethylsilyl)prop-1yne, by a procedure similar to that for 2 a , gave 2 m in $55 \%$ yield. The reaction involved a counterattack by benzenethiolate ion. The $2 E, 4 E$ - and $2 Z, 4 E$-isomers ( $3: 2$ ) were separated by HPLC (EtOAc-hexane, 1:9) appearing at $t_{\mathrm{R}} 10.0$ and 8.5 min , respectively. $2 E, 4 E$-Isomer: $\nu_{\text {max }}($ neat $) / \mathrm{cm}^{-1} 2219,1583$ and $1543 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 3.22(3 \mathrm{H}, \mathrm{s}), 6.21(1 \mathrm{H}, \mathrm{d}, J 10.3), 6.52(1 \mathrm{H}$, d, $J 17.0), 6.73(1 \mathrm{H}, \mathrm{dd}, J 10.3,17.0), 7.02-7.2(3 \mathrm{H}, \mathrm{m})$ and 7.3-7.45 ( $7 \mathrm{H}, \mathrm{m}$ ); $m / z$ (Found: $\mathrm{M}^{+}$, 292.1032. Calc. for $M$, 292.1034). $2 E, 4 Z$-Isomer: $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 3.28(3 \mathrm{H}, \mathrm{s}), 6.32(1 \mathrm{H}$, d, $J 10.0$ ), $6.47(1 \mathrm{H}, \mathrm{d}, J 12.5), 6.74(1 \mathrm{H}, \mathrm{dd}, J 10.0,12.5), 7.10-$ $7.22(3 \mathrm{H}, \mathrm{m})$ and $7.25-7.45(7 \mathrm{H}, \mathrm{m}) ; m / z 292\left(\mathrm{M}^{+}, 100 \%\right)$ and 183 (92) (Found: $\mathrm{M}^{+}, 292.1024$. Calc. for $M, 292.1034$ ).

Methyl N -Methyl- N -phenylcarbamate 3a. ${ }^{15}$-To a THF solution ( $5 \mathrm{~cm}^{3}$ ) of $\mathbf{1}(254 \mathrm{mg}, 1 \mathrm{mmol})$ was added $\mathrm{CuI}(20 \mathrm{mg})$ and MeONa ( $65 \mathrm{mg}, 1.2 \mathrm{mmol}$ ). After 10 min , a stream of oxygen was bubbled into the solution. The mixture was stirred at room temperature for 12 h and quenched by addition to it of aqueous KI. The mixture was filtered, concentrated and extracted with EtOAc. The combined extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, concentrated and separated on a silica gel column (EtOAc-hexane, $1: 19$ ) to give 3a ( $180 \mathrm{mg}, 71 \%$ ); liquid, TLC (EtOAc-hexane, $1: 19$ ), $R_{\mathrm{f}} 0.2$. The spectral data have been described earlier. ${ }^{15}$

Ethyl N-Methyl-N-phenylcarbamate 3b. ${ }^{16}$-The autoxidation of 1 in the presence of EtONa, by a procedure similar to that for 3a, gave 3b as liquid in $64 \%$ yield, the spectral data have been described earlier. ${ }^{16}$

Isopropyl N -Methyl- N -phenylcarbamate $3 \mathrm{3c} .^{17}$-The autoxidation of $\mathbf{1}$ in the presence of $\mathrm{Pr}{ }^{\mathrm{O} O L i}$ (prepared from $\mathrm{Pr}^{\mathrm{i} O H}$ and BuLi ), by a procedure similar to that for $\mathbf{3 a}$, gave $\mathbf{3 c}$ as liquid in $74 \%$ yield. The spectral data have been described earlier. ${ }^{17}$

Allyl N-Methyl-N-phenylcarbamate 3d.-The autoxidation of $\mathbf{1}$ in the presence of lithium allyl oxide (prepared from allyl alcohol and BuLi), by a procedure similar to that for 3a, gave 3d in $65 \%$ yield; liquid, TLC (EtOAc-hexane, 1:19), $R_{\mathrm{f}} 0.3$; $v_{\text {max }}$ (neat) $/ \mathrm{cm}^{-1} 1710(\mathrm{C}=\mathrm{O})$ and $1598 ; \mathrm{m} / \mathrm{z} 191\left(\mathrm{M}^{+}, 37 \%\right)$ and $105(100) ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 3.31(3 \mathrm{H}, \mathrm{s}), 4.60(2 \mathrm{H}, \mathrm{m}), 5.18(2 \mathrm{H}, \mathrm{m})$, $5.88(1 \mathrm{H}, \mathrm{m})$ and $7.16-7.39(5 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 37.6\left(\mathrm{NCH}_{3}\right)$, 66.1 (C-1), 117.1 (C-3), 125.7 (C-2'), 126.0 (C-4'), 128.7 (C-3'), 132.7 (C-2), $143.1\left(\mathrm{C}-1^{\prime}\right)$ and $155.3(\mathrm{C}=\mathrm{O})$ (Found: $\mathrm{M}^{+}$, 191.0952. Calc. for $M, 191.0946)$.

N -Methyl-N-phenylcyanamide $\mathbf{4}$ and S-Phenyl N -Methyl- N phenylthiocarbamate 5.-To a cold ( $-78^{\circ} \mathrm{C}$ ) THF solution (2 $\mathrm{cm}^{3}$ ) of $1(254 \mathrm{mg}, 1 \mathrm{mmol})$ was added $\mathrm{Bu}^{t} \mathrm{OK}(60 \mathrm{mg})$ in THF $\left(5 \mathrm{~cm}^{3}\right)$. The solution was saturated with oxygen, warmed to room temperature, and stirred for 18 h . Aqueous KI was added to the mixture to quench the reaction. The mixture was concentrated and extracted with EtOAc. The combined extracts were washed with aqueous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ and brine, dried $\left(\mathrm{Na}_{2}{ }^{-}\right.$ $\mathrm{SO}_{4}$ ), filtered and concentrated. The residue was separated by silica-gel chromatography (EtOAc-hexane, 1:19) to give 4 ( 86 $\mathrm{mg}, 54 \%$ ) and 5 ( $58 \mathrm{mg}, 24 \%$ ). Compound 4: $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1}$ $2230(\mathrm{CN})$ and $1698(\mathrm{C}=\mathrm{O}) ; m / z 160\left(\mathrm{M}^{+}, 100 \%\right)$ and $132(24)$; $\delta_{\mathbf{H}}\left(\mathrm{CDCl}_{3}\right) 3.37(3 \mathrm{H}, \mathrm{s})$ and $7.20-7.45(5 \mathrm{H}, \mathrm{m})$ (Found: $\mathrm{M}^{+}$, 160.0642. Calc. for $M, 160.0637$ ). Compound 5: $v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1}$ $1668 ; m / z 243\left(\mathrm{M}^{+}, 22 \%\right)$ and $134(100) ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 3.35(3 \mathrm{H}, \mathrm{s})$ and 7.25-7.50 ( $5 \mathrm{H}, \mathrm{m}$ ); $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 38.5$ (s), 128.3 (d), 128.6 (d), 128.8 (d), 129.0 (d), 129.4 (s), 129.5 (d), 135.4 (d), 141.8 (s) and 167.4 (s, C=O) (Found: $\mathrm{M}^{+}, 243.0713$. Calc. for $M, 243.0718$ ).

2-( N -Methylanilino)propanenitrile 6 ..$^{18}$--To a cold ( $-40^{\circ} \mathrm{C}$ ) THF solution $\left(5 \mathrm{~cm}^{3}\right)$ of $1(178 \mathrm{mg}, 0.7 \mathrm{mmol})$ and $\mathrm{CuI}(10 \mathrm{mg})$ was added dropwise $\mathrm{MeMgCl}\left(3.0 \mathrm{~mol} \mathrm{dm}^{-3} \mathrm{THF}\right.$ solution; 1.4 $\mathrm{mmol}, 0.47 \mathrm{~cm}^{3}$ ). The mixture was warmed to room temperature and stirred for 2 h . After addition of ice-water to the mixture it was extracted with EtOAc. The combined extracts were washed with water and brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, filtered and concentrated. The residue was purified by silica-gel chromatography (EtOAc-hexane, 1:9) to give $6 \mathbf{a}$ ( $111 \mathrm{mg}, 99 \%$ ); liquid, TLC (EtOAc-hexane, $1: 9$ ), $R_{\mathrm{f}} 0.2$. The spectral data have been reported earlier. ${ }^{18}$

2-(N-Methylanilino)butanenitrile 6b.-Compound 6b was prepared in $81 \%$ yield from 1 and EtMgBr by a procedure similar to that for 6a; liquid, TLC (EtOAc-hexane, 1:19), $R_{\mathrm{f}}$ $0.2 ; v_{\text {max }}($ neat $) / \mathrm{cm}^{-1} 2227(\mathrm{CN}) ; m / z 174\left(\mathrm{M}^{+}, 60 \%\right)$ and 146 (100); $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.90(3 \mathrm{H}, \mathrm{t}, J 7), 1.95(2 \mathrm{H}, \mathrm{m}), 2.88(3 \mathrm{H}, \mathrm{s})$, $4.36(1 \mathrm{H}, \mathrm{t}, J 7)$ and $6.88-7.33(5 \mathrm{H}, \mathrm{m})$ (Found: $\mathrm{M}^{+}, 174.1156$. Calc. for $M, 174.1157$ ).

2-( N -Methylanilino)hexanenitrile 6c.-Compound 6c was prepared in $84 \%$ yield from 1 and BuLi by a procedure similar to that for 6a; liquid, TLC (EtOAc-hexane, 1:19), $R_{\mathrm{f}} 0.3$; $v_{\text {max }}($ neat $) / \mathrm{cm}^{-1} 2220(\mathrm{CN}) ; m / z 202\left(\mathrm{M}^{+}, 56 \%\right)$ and $145(100)$; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 0.93(3 \mathrm{H}, \mathrm{t}, J 7), 1.45(4 \mathrm{H}, \mathrm{m}), 1.95(2 \mathrm{H}, \mathrm{q}, J 7), 2.90$ $(3 \mathrm{H}, \mathrm{s}), 4.45(1 \mathrm{H}, \mathrm{t}, J 7)$ and $6.85-7.35(5 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 13.8$ (C-6), $22.0(\mathrm{C}-4), 27.8(\mathrm{C}-5), 31.4(\mathrm{C}-3), 34.2\left(\mathrm{NCH}_{3}\right), 54.0(\mathrm{C}-2)$, 116.8 (C-2'), 118.0 (CN), 120.7 (C-4'), 129.3 (C-3') and 149.3 (C-1') (Found: $\mathrm{M}^{+}, 202.1474$. Calc. for $M, 202.1470$ ).

2-( N -Methylanilino)pent-4-enenitrile 7 .-To a cold ( $-78^{\circ} \mathrm{C}$ ) $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ solution ( $4 \mathrm{~cm}^{3}$ ) of $1(254 \mathrm{mg}, 1 \mathrm{mmol})$ was added sequentially $\mathrm{TiCl}_{4}\left(0.15 \mathrm{~cm}^{3}, 1.5 \mathrm{mmol}\right)$ and allylsilane $\left(0.2 \mathrm{~cm}^{3}\right.$, 1.27 mmol ). The mixture was stirred at $-78^{\circ} \mathrm{C}$ for 40 min , quenched by addition of aqueous $\mathrm{NaOH}\left(5 \%, 0.5 \mathrm{~cm}^{3}\right)$ to it and extracted with $\mathrm{CHCl}_{3}$. The extract was washed with brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, concentrated and purified by silica-gel chromatography (EtOAc-hexane, 1:19) to give 7 ( $143 \mathrm{mg}, 77 \%$ ); liquid, TLC (EtOAc-hexane, $1: 19$ ), $R_{\mathrm{f}} 0.19 ; v_{\text {max }}($ neat $) / \mathrm{cm}^{-1}$ $2231(\mathrm{CN})$ and $1641 ; m / z 186\left(\mathrm{M}^{+}, 12 \%\right)$ and 145 (100); $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 2.62(2 \mathrm{H}, \mathrm{t}, J 7.5), 2.89(3 \mathrm{H}, \mathrm{s}), 4.49(1 \mathrm{H}, \mathrm{t}, J 7.5)$, 5.17-5.30 ( $2 \mathrm{H}, \mathrm{m}$ ), 5.70-5.91 ( $1 \mathrm{H}, \mathrm{m}$ ) and 6.82-7.33 ( $5 \mathrm{H}, \mathrm{m}$ ); $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 34.3\left(\mathrm{NCH}_{3}\right), 35.8(\mathrm{C}-3), 54.0(\mathrm{C}-2), 116.6\left(\mathrm{C}-2^{\prime}\right)$, 117.2 (CN), 119.5 (C-5), 120.8 (C-4'), 129.2 (C-3'), 131.5 (C-4) and 148.9 (C-1') (Found: $\mathrm{M}^{+}, 186.1151$. Calc. for $M, 186.1157$ ).

## Methyl 3-Cyano-2,2-dimethyl-3-(N-methylanilino)propanoate

 8.-To a cold $\left(-78^{\circ} \mathrm{C}\right)$ THF solution ( $10 \mathrm{~cm}^{3}$ ) of LDA ( 10 mmol ), prepared from $\operatorname{BuLi}\left(1.6 \mathrm{~mol} \mathrm{dm}^{-3}\right.$ hexane solution; 7.7$\mathrm{cm}^{3}$ ) and diisopropylamine ( $1.5 \mathrm{~cm}^{3}$ ), was added dropwise methyl isobutyrate ( $1.15 \mathrm{~cm}^{3}, 10 \mathrm{mmol}$ ). After 30 min , chlorotrimethylsilane ( $1.4 \mathrm{~cm}^{3}, 11 \mathrm{mmol}$ ) was added to the mixture which was then stirred at $-78^{\circ} \mathrm{C}$ for 5 min , and warmed to room temperature over 1 h . Saturated aqueous $\mathrm{NH}_{4} \mathrm{Cl}\left(0.5 \mathrm{~cm}^{3}\right)$ was added to the mixture which was then concentrated and extracted with $\mathrm{Et}_{2} \mathrm{O}$. The combined extracts were washed with brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, concentrated and distilled under reduced pressure to give the corresponding silyl ketene acetal, b.p. $46^{\circ} \mathrm{C} / 30$ Torr.
To a cold $\left(-78{ }^{\circ} \mathrm{C}\right) \mathrm{CH}_{2} \mathrm{Cl}_{2}$ solution $\left(2 \mathrm{~cm}^{3}\right)$ of $\mathbf{1}(254 \mathrm{mg}, 1$ $\mathrm{mmol})$ was added $\mathrm{TiCl}_{4}\left(0.15 \mathrm{~cm}^{3}, 1.5 \mathrm{mmol}\right)$ and the freshly prepared silyl ketene acetal ( $191 \mathrm{mg}, 1.1 \mathrm{mmol}$ ). After the addition, the mixture was warmed to room temperature and stirred for 1 h . The solution was poured into cold $\left(0^{\circ} \mathrm{C}\right)$ aqueous $\mathrm{NaOH}(5 \%)$, and extracted with EtOAc. The combined extracts were washed with brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated, and the residue separated by silica-gel chromatography (EtOAc-hexane, 1:9) to give 8 ( $71 \mathrm{mg}, 30 \%$ ), accompanied by recovery of $1(25 \%)$. Compound 8 : liquid, TLC (EtOAc-hexane, 1:9) $R_{\mathrm{f}} 0.23 ; v_{\max }($ neat $) / \mathrm{cm}^{-1} 2250(\mathrm{CN})$ and $1735\left(\mathrm{CO}_{2} \mathrm{Me}\right) ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.40(3 \mathrm{H}, \mathrm{s}), 1.47(3 \mathrm{H}, \mathrm{s}), 2.97(3 \mathrm{H}$, s), $3.65(3 \mathrm{H}, \mathrm{s}), 4.95(3 \mathrm{H}, \mathrm{s}), 6.88-7.02(3 \mathrm{H}, \mathrm{m})$ and $7.18-7.42$ $(2 \mathrm{H}, \mathrm{m}) ; m / z 246\left(\mathrm{M}^{+}, 20 \%\right)$ and 1.45 (100) (Found: $\mathrm{M}^{+}$, 246.1375. Calc. for $M, 246.1368$ ) (Found C, 68.1; H, 7.35; N, 11.3. $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{2}$ requires $\mathrm{C}, 68.27 ; \mathrm{H}, 7.37 ; \mathrm{N}, 11.37 \%$ ).

3-Benzoyl-2-( N -methylanilino)butanenitrile 9a.--To a cold $\left(-78^{\circ} \mathrm{C}\right) \mathrm{CH}_{2} \mathrm{Cl}_{2}$ solution ( $15 \mathrm{~cm}^{3}$ ) of propiophenone $(402 \mathrm{mg}$, 3 mmol ) was added $\mathrm{TiCl}_{4}\left(0.31 \mathrm{~cm}^{3}, 3.1 \mathrm{mmol}\right)$ under $\mathrm{N}_{2}$. The yellow slurry was stirred for 2 min after which $N, N$-diisopropylethylamine ( $0.56 \mathrm{~cm}^{3}, 3.2 \mathrm{mmol}$ ) was added dropwise to it; the resulting deep red solution was stirred at $-78^{\circ} \mathrm{C}$ for 1.5 h . $\mathrm{A} \mathrm{CH}_{2} \mathrm{Cl}_{2}$ solution ( $2 \mathrm{~cm}^{3}$ ) of $\mathbf{1}(254 \mathrm{mg}, 1 \mathrm{mmol})$ was added dropwise to the mixture which was then warmed to room temperature and stirred for 1.5 h . The mixture was worked up by a procedure similar to that for 8 to give 9 a $(659 \mathrm{mg}, 80 \%$ ) as a mxiture of two diastereoisomers ( $62: 38$ ). Liquid, TLC (EtOAchexane $12: 88), R_{\mathrm{f}} 0.19 ; v_{\max }($ neat $) / \mathrm{cm}^{-1} 2228$ and 1678 ; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.37 / 1.45(3 \mathrm{H}, \mathrm{d}, J 6), 2.97 / 2.81(3 \mathrm{H}, \mathrm{s}), 4.13-4.29(1$ $\mathrm{H}, \mathrm{m}), 4.96 / 5.05(1 \mathrm{H}, \mathrm{d}, J 6), 6.92-7.06(3 \mathrm{H}, \mathrm{m}), 7.23-7.28(2 \mathrm{H}$, $\mathrm{m}), 7.44-7.57(3 \mathrm{H}, \mathrm{m})$ and $7.90-7.94(2 \mathrm{H}, \mathrm{m}) ; m / z 278\left(\mathrm{M}^{+}\right.$, $15 \%$ ) and 145 (100) (Found: C, 77.3; H, 6.5; N, 9.85. $\mathrm{C}_{18} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}$ requires C, $\left.77.67 ; \mathrm{H}, 6.52 ; \mathrm{N}, 10.06 \%\right)$.

3-Methyl-2-( N -methylanilino)-4-oxohexanenitrile $9 \mathbf{9 b}$.-Compound 9 b was prepared by condensation of pentan-3-one and 1 in $65 \%$ yield by a procedure similar to that for 9 a. Compound 9b: a mixture of two diastereoisomers ( $60: 40$ ), liquid, TLC (EtOAc-hexane, 12:88), $R_{\mathrm{f}} 0.2 ; v_{\max }($ neat $) / \mathrm{cm}^{-1}$ 2229 and $1710 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.05(3 \mathrm{H}, \mathrm{t}, J 6), 1.28 / 1.19(3 \mathrm{H}, \mathrm{d}, J$ 6), 2.31-2.76 ( $2 \mathrm{H}, \mathrm{m}$ ), $2.86 / 2.85(3 \mathrm{H}, \mathrm{s}), 3.23-3.34$ ( $1 \mathrm{H}, \mathrm{m}$ ), $4.78 / 4.86(1 \mathrm{H}, \mathrm{d}, J 12), 6.89-7.02(3 \mathrm{H}, \mathrm{m})$ and $7.24-7.33(2 \mathrm{H}$, $\mathrm{m}) ; m / z 230\left(\mathrm{M}^{+}, 10 \%\right)$ and $145(100)$ (Found: C, $72.9 ; \mathrm{H}, 7.7$; N , 11.8. $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}$ requires $\mathrm{C}, 73.01 ; \mathrm{H}, 7.88 ; \mathrm{N}, 12.16 \%$ ).

2,3-Bis( N -methylanilino)butane-1,4-dinitrile 10.-To a mild refluxing $\left(80^{\circ} \mathrm{C}\right)$ solution of $1(127 \mathrm{mg}, 0.5 \mathrm{mmol})$ in benzene ( $3 \mathrm{~cm}^{3}$ ) was added dropwise a benzene solution ( $7 \mathrm{~cm}^{3}$ ) of $\mathrm{Bu}_{3} \mathrm{SnH}\left(0.28 \mathrm{~cm}^{3}, 1 \mathrm{mmol}\right)$, methyl acrylate ( $0.09 \mathrm{~cm}^{3}, 1 \mathrm{mmol}$ ) and azoisobutyronitrile ( 16.4 mg ) over a period of 2 h . The mixture was heated for 6 h at reflux, cooled and concentrated by rotary evaporation. The residue was passed through a silica gel column by elution with hexane to remove benzenethiol, and by elution with EtOAc-hexane (5:95) to give methyl 3-(tributylstannyl)propanoate ( $67.7 \mathrm{mg}, 37 \%$ ), 2-( $N$-methylanilino)acetonitrile ( $8.8 \mathrm{mg}, 12 \%$ ) and $10(63 \mathrm{mg}, 44 \%$, containing two isomers in equal amounts). Compound 10: oil, TLC (EtOAc-
hexane, $10: 90$ ), $R_{\mathrm{f}} 0.19 ; \mathrm{m} / \mathrm{z} 290\left(\mathrm{M}^{+}, 17 \%\right), 263$ (11) and 145 (100); $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 7.34-6.84(10 \mathrm{H}, \mathrm{m}), 4.97(2 \mathrm{H}, \mathrm{s}) 4.85(2$ $\mathrm{H}, \mathrm{s})$ and $2.89(6 \mathrm{H}, \mathrm{s}) / 3.03(6 \mathrm{H}, \mathrm{s}) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 148.1 / 148.7(\mathrm{~s})$, 129.6/129.7 (d), 122.0/123.3 (d), 117.0/118.9 (d), 114.7/114.2 (s, CN ), $56.0 / 58.1$ (d, C-2) and 35.5/36.7 (q) (Found: $\mathrm{M}^{+}$, 290.1510. Calc. for $M, 290.1531$ ).

2-[ N -Methyl-(o-propenylphenyl)amino]-2-phenylsulfanylacetonitrile 11.-Treatment of $2-[N$-methyl-(o-propenylphenyl)amino]acetonitrile with LDA and diphenyl disulfide, by a procedure similar to that for 1, gave 11 in $74 \%$ yield; oil, TLC (EtOAc-hexane, 5:95), $R_{\mathrm{f}} 0.22 ; v_{\text {max }}($ neat $) / \mathrm{cm}^{-1} 2211$; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.82(3 \mathrm{H}, \mathrm{dd}, J 6.6,1.0), 3.02(3 \mathrm{H}, \mathrm{s}), 5.42(1 \mathrm{H}, \mathrm{s})$, $6.06(1 \mathrm{H}, \mathrm{dq}, J 16.0,6.6), 6.44(1 \mathrm{H}, \mathrm{dd}, J 16.0,1.0)$ and $7.12-$ $7.54(9 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 18.8(\mathrm{q}), 36.2$ (q), 64.1 (d), 115.5 (s), 121.7 (d), 125.2 (d), 127.2 (d), 127.5 (d), 127.8 (d), 128.1 (d), 129.1 (d), 129.2 (d, 2 C), 131.2 (s), 132.8 (s), 134.6 (d, 2 C) and 145.1 (s); $m / z 294\left(1 \%, \mathrm{M}^{+}\right)$and 185 (100) (Found: $\mathrm{M}^{+}$, 294.1176. Calc. for $M, 294.1191$ ).

3-Ethyl-1-methyl-2,3-dihydroindole-2-carbonitrile 12.-To a mild refluxing ( $80^{\circ} \mathrm{C}$ ) solution of $11(150 \mathrm{mg}, 0.51 \mathrm{mmol})$ in benzene ( $15 \mathrm{~cm}^{3}$ ) was added dropwise a benzene solution ( 15 $\left.\mathrm{cm}^{3}\right)$ of $\mathrm{Bu}_{3} \mathrm{SnH}\left(0.17 \mathrm{~cm}^{3}, 0.61 \mathrm{mmol}\right)$ and azoisobutyronitrile $(17 \mathrm{mg})$ over a period of 2.5 h . The mixture was heated for 6 h and concentrated. The residue was chromatographed on a silica gel column by elution with hexane to remove tin compounds and followed by elution with EtOAc. The EtOAc phase was treated with $\mathrm{Et}_{3} \mathrm{~N}\left(0.5 \mathrm{~cm}^{3}\right)$ and the white precipitate was filtered off. The filtrate was concentrated and separated by HPLC (EtOAc-hexane, 5:95) to give cis-12 (53 mg) and trans$12(28 \mathrm{mg})$ in $91 \%$ total yield. Compound trans-12: oil, TLC (EtOAc-hexane, 5:95), $R_{\mathrm{f}} 0.20 ; v_{\max }($ neat $) / \mathrm{cm}^{-1} 2248 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right)$ $1.08(3 \mathrm{H}, \mathrm{t}, J 7.4), 1.61-1.76(1 \mathrm{H}, \mathrm{m}), 1.81-1.96(1 \mathrm{H}, \mathrm{m}), 2.88(3$ $\mathrm{H}, \mathrm{s}), 3.40-3.50(1 \mathrm{H}, \mathrm{m}), 3.93(1 \mathrm{H}, \mathrm{d}, J 6), 6.54(1 \mathrm{H}, \mathrm{d}, J 8), 6.80$ ( 1 H , ddd, $J 8,8,1$ ), 7.09 ( 1 H , dd, $J 8,1$ ) and 7.16 ( 1 H , dd, $J 8$, $8) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 11.4(\mathrm{q}), 26.2(\mathrm{t}), 34.2(\mathrm{q}), 48.5(\mathrm{~d}, \mathrm{C}-3), 61.4$ (d, C-2), 108.2 (d), 118.6 (s, CN), 119.5 (d), 123.8 (d), 128.5 (d), 130.3 (s) and 150.1 (s); $m / z 186\left(33 \%, \mathrm{M}^{+}\right)$and 157 (100). Compound cis-12: oil, TLC (EtOAc-hexane, 5:95), $R_{\mathrm{f}} 0.17$; $v_{\max }($ neat $) / \mathrm{cm}^{-1} 2219(\mathrm{CN}) ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.11(3 \mathrm{H}, \mathrm{t}, J 7.4), 1.80-$ $1.96(1 \mathrm{H}, \mathrm{m}), 2.00-2.18(1 \mathrm{H}, \mathrm{m}), 2.87(3 \mathrm{H}, \mathrm{s}), 3.28-3.41(1 \mathrm{H}$, m), $4.42(1 \mathrm{H}, \mathrm{d}, J 8), 6.56(1 \mathrm{H}, \mathrm{d}, J 8), 6.81(1 \mathrm{H}, \mathrm{ddd}, J 8,8,1)$, $7.09(1 \mathrm{H}, \mathrm{dd}, J 8,1)$ and $7.16(1 \mathrm{H}$, ddd, $J 8,8,1) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right)$ 11.9 (t), 22.6 (t), 34.1 (q), 45.4 (d), 62.4 (d), 108.3 (d), 116.1 (s), 119.7 (d), 123.5 (d), 128.4 (d), 130.6 (s) and 150.3 (s); $m / z 186$ $\left(34 \%, \mathrm{M}^{+}\right)$and 157 (100) (Found: $\mathrm{M}^{+}, 186.1145$. Calc. for $M$, 186.1157).

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## References

1 S. Arseniyadis, K. S. Kyler and D. S. Watt, Org. Reactions, 1984, 31, 1
2 J.-M. Fang and C.-C. Chen, J. Chem. Soc., Perkin Trans. I, 1990, 3365; T.-H. Chuang, C.-C. Yang, C.-J. Chang and J.-M. Fang, Synlett, 1990, 733.
3 J. H. Boyer and J. Kooi, J. Am. Chem. Soc., 1976, 98, 1099.
4 A. Padwa, W. Dent, H. Nimmesgern, M. K. Venkatramanan and G. S. K. Wong, Chem. Ber., 1986, 119, 813.

5 H. G. Aurich, Tetrahedron Lett., 1964, 657; N. Rabjohn and C. A. Harbert, J. Org. Chem., 1970, 35, 3240.
6 H. Ahlbrecht and H. Dollinger, Synthesis, 1985, 743; W. H. Bunnelle and C. G. Shevlin, Tetrahedron Lett., 1989, 30, 4203
7 T. Hayashi, M. Konish and M. Kumada, J. Am. Chem. Soc., 1982, 104, 4963; J. M. McNamara and Y. Kishi, J. Am. Chem. Soc., 1982, 104, 7371; P. A. Bartlett, W. S. Johnson and J. D. Elliott, J. Am. Chem. Soc., 1983, 105, 2088; M. T. Reetz and K. Kesseler, J. Org. Chem., 1985, 264, 99
8 T. Mukaiyama, Org. Reactions, 1982, 28, 238; D. A. Evans, D. L. Rieger, M. T. Bilodeau and F. Urpi, J. Am. Chem. Soc., 1991, 113, 1047.

9 J. D. Albright, Tetrahedron, 1983, 39, 3207.
10 H. G. Viehe, Z. Janousek and R. Merenyi, Acc. Chem. Res., 1985, 18, 148.

11 H. Ahlbrecht and K. Pfaff, Synthesis, 1978, 897; 1985, 421; J.-M. Fang and H.-T. Chang, J. Chem. Soc., Perkin Trans. 1, 1988, 1945.

12 K. Takahashi, K. Shibasaki, K. Ogura and H. Iida, J. Org. Chem., 1983, 48, 3566.
13 C.-C. Lin, M.S. Thesis, National Taiwan University, 1989.
14 J.-M. Fang, C.-C. Yang and Y.-W. Wang, J. Org. Chem., 1989, 54, 477.
15 Y. Tsujimoto, Y. Nishimura, A. Kosaka, H. Kiriyama, Y. Miyamoto and Y. Odaira, Tetrahedron Lett., 1979, 4, 373.
16 E. A. Parfenov and V. A. Fomin, Zh. Obshch. Khim., 1981, 51, 1144.

17 M. J. Beck, Biotechnol. Lett., 1986, 8, 513.
18 H. Ahlbrecht, W. Raab and C. Vonderheid, Synthesis, 1979, 2, 127.

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