Shwu-Li Wu^{a,b} (吳淑黎), Guey-Horng Wang^a (王貴弘), Chang-Feng Dai^c (戴昌鳳) and Jyh-Horng Sheu^a* (許志宏) ^aDepartment of Marine Resources, National Sun Yat-Sen University, Kaohsiung 804, Taiwan, R.O.C. ^bCenter for General Education, National Kaohsiung Institute of Marine Technology, Kaohsiung 811, Taiwan, R.O.C.

^cInstitute of Oceanography, National Taiwan University, Taipei 106, Taiwan, R.O.C.

A new steroid, 11α , 15α -diacetoxy- 17α -pregna-4,20-dien-3-one (1), and a known one, 17α -pregna-4,20-dien-3-one (2), have been isolated from a Formosan gorgonian *Subergorgia mollis*. The structures of both compounds were determined on the basis of extensive NMR experiments, including HMQC, HMBC, ¹H-¹H COSY and NOESY. Metabolite 2 has not been isolated from a natural source before. The detailed ¹H and ¹³C NMR spectral data of 2 are reported for the first time.

Keywords: Gorganian; Subergorgia mollis; Pregnane-Based Steroids.

INTRODUCTION

Previous investigations on secondary metabolites from Gorgonian corals of the genus *Subergorgia* (phylum Cnidaria, class Anthozoa, subclass Octocorallia, order Gorgonacea, family Subergorgiidae) have shown the presence of a variety of sesquiterpenes¹⁻⁸ and steroids.⁹ The Gorgonian coral *Subergorgia mollis* (Nutting, 1910) occurs widely in Taiwanese tropical waters, and its chemical constituents have not been reported. In this paper, we wish to describe the isolation and structural elucidation of a new steroid, 11α , 15α -diacetoxy-17 α -pregna-4,20-dien-3-one (1), and a known one, 17 α pregna-4,20-dien-3-one (2), from a Formosan gorgonian, *S. mollis*. The structures of the two pregnane-based steroids were determined on the basis of extensive NMR experiments (HMQC, HMBC, ¹H-¹H COSY and NOESY). Metabolite 2 was isolated as a natural product for the first time.

RESULTS AND DISCUSSION

The new metabolite 1 was obtained as a white solid. Its HRFABMS showed a pseudomolecular ion peak, $[M + H]^+$ at m/z 415.2474, appropriate for the molecular formula $C_{25}H_{34}O_5$ and thus requiring nine degrees of unsaturation. The ¹³C NMR spectrum displayed signals for 25 carbons, which were assigned by the DEPT spectrum into four methyls, seven methylenes, and eight methines (Table 1). The re-



maining six signals appearing in the broad-band spectrum were attributed to the quaternary carbon atoms. An IR absorption band at 1726 cm⁻¹ and the significant FABMS peaks at $m/z 415 [M + H]^+$, 355 $[M + H - HOAc]^+$, and 295 $[M + H - HOAc]^+$ 2 HOAc^+ , revealed the presence of two acetoxy groups in 1. The presence of two acetoxy moieties was further confirmed by ¹³C NMR spectrum, with signals at δ 170.0 (s) and 170.9 (s), and by ¹H NMR spectrum (Table 2) with two acetyl methyl singlets at δ 2.02 and 2.03. An IR absorption band at 1667 cm⁻¹ indicated the existence of a conjugate enone structural unit, which was further confirmed by a UV absorption at 248 nm. The signal at δ 5.76 (1H, s) was assigned for the proton attached to the α -carbon of the enone moiety. The presence of the enone unit could also be proved by the ¹³C NMR spectrum, which showed signals at δ 199.3 (s), 124.7 (d), and 169.4 (s). The presence of a vinyl group was deduced from ${}^{1}\text{H}$ NMR signals at δ 5.65 (1H, ddd, J = 17.1, 10.2, 7.0 Hz), 5.05 (1H, d, J = 10.2 Hz), and 5.01 (1H, d, J = 17.1 Hz), and car-

* Corresponding author. Tel: +886-7-5252000 ext. 5030; fax: +886-7-5255020; e-mail: sheu@mail.nsysu.edu.tw

Position	1 ^a	2 ^a
1	$36.5(t)^{b}$	35.7 (t)
2	34.1 (t)	33.9 (t)
3	199.3 (s)	199.7 (s)
4	124.7 (d)	123.8 (d)
5	169.4 (s)	171.6 (s)
6	33.2 (t)	32.9 (t)
7	31.2 (t)	32.0 (t)
8	34.7 (d)	35.7 (d)
9	55.6 (d)	54.0 (d)
10	39.8 (s)	38.7 (s)
11	70.5 (d)	20.6 (t)
12	44.1 (t)	37.2 (t)
13	43.6 (s)	43.4 (s)
14	57.4 (d)	55.0 (d)
15	75.5 (d)	24.7 (t)
16	36.8 (t)	27.1 (t)
17	52.6 (d)	55.2 (d)
18	14.6 (q)	12.8 (q)
19	18.5 (q)	17.4 (q)
20	136.7 (d)	139.5 (d)
21	116.6 (t)	114.8 (t)
acetate	21.3 (q)	
methyls	21.8 (q)	
carbonyls	170.0 (s)	
	170.9 (s)	

Table 1. ¹³C NMR Data of Compounds 1 and 2

^a Spectra recorded at 125 MHz in CDCl₃ at 25 °C.

^b Multiplicity deduced by DEPT and indicated by usual symbols.

The values are in ppm downfield from TMS.

bons at δ 136.7 (d), and 116.6 (t). The molecular framework was established by ¹H-¹H COSY and HMBC spectra (Fig. 1). From the ¹H-¹H COSY spectrum of **1**, it was possible to establish the sequential proton sets from H_2 -1 to H_2 -2; H_2 -6 to H-9; H-8 to H-14; H-9 to H-11; H-11 to H₂-12; H-14 to H-17; H-17 to H-20; and H-20 to H₂-21. The HMBC spectrum of 1 showed ¹H-¹³C long range correlations of H₃-19 with C-1, C-5, C-9 and C-10, and H₃-18 with C-12, C-13, C-14 and



Fig. 1. ¹H-¹H COSY and HMBC Correlations for 1.

Table 2. ¹ H NMR Data of Compounds 1 and 2		
Position	1 ^a	2 ^a
1α	1.95 m	1.70 m
β	1.99 m	2.02 m
2α	2.40 m	2.42 m
β	2.38 m	2.35 m
4	5.76 s	5.73 s
6α	2 20	2.29 m
β	2.30 m	2.39 m
7α	1.25 m	1.04 m
β	1.82 m	1.86 m
8	1.84 m	1.68 m
9	1.51 m	0.95 ddd
		(11.5, 11.5, 4.5)
11α		1.40 m
β	5.22 ddd	1.55 m
	$(10.5, 10.5, 5.5)^{\rm b}$	
12α	1.21 m	1.05 m
β	2.08 m	1.72 m
14	1.51 m	1.02 m
15α		1.22 m
β	4.92 ddd	1.71 m
	(8.5, 3.5, 3.5)	
16α	1.68 m	1.80 m
β	2.31 m	1.57 m
17	2.28 m	1.97 m
18	0.78 s	0.64 s
19	1.28 s	1.25 s
20	5.65 ddd	5.75 ddd
	(17.1, 10.2, 7.0)	(17.5, 10.4, 7.5)
21	5.05 d (10.2)	4.98 d (10.4)
	5.01 d (17.1)	4.97 d (17.5)
acetate	2.02 s	
methyls	2.03 s	

^a Spectra recorded at 500 MHz in CDCl₃ at 25 °C.

^b J values (in Hz) in parentheses.

C-17, suggesting that the ring-junctured C-18 and C-19 methyl groups should be positioned at C-13 and C-10, respectively. Comparison of the ¹H NMR spectral data of compound 1 with those of compound 2, which was also isolated from the present study (discussed below), revealed that 1 is the diacetoxy derivative of 2. The placement of acetoxy groups at C-11 and C-15 was confirmed from the HMBC connectivities of H-11 (δ 5.22) and H-15 (δ 4.92) with the carbonyl carbons of esters resonating at δ 170.0 (s) and 170.9 (s), respectively. Furthermore, both H₂-1 and H₂-2 have been shown to exhibit HMBC correlations with the enone carbonyl carbon at δ 199.3, revealing the ketonic functionality of C-3. On the basis of the above observations, metabolite 1 was determined to be the 11,15-diacetoxy derivative of pregna-

4,20-dien-3-one.

The relative stereochemistry of **1** was determined by the NOE correlations observed in the NOESY spectrum of **1** (Fig. 2). It was found that H-8 had NOE interactions with H₃-18 and H₃-19, but not with H-14 and H-9, suggesting the β -orientations of H-8, H₃-19, and H₃-18, and the α -orientations of H-9 and H-14. The H₃-19 showed NOE interaction with H-11, and H₃-18 showed NOE interactions with H-11, H-15, and H-20, indicating that H-11, H-15, and the vinyl group attached at C-17 should be in the β face. Finally, it was found that H-14 exhibited NOE crosspeaks with both H-9 and H-17, confirming the α -orientations of these three protons. On the basis of the above analyses, the structure of **1** was established unambiguously as 11α , 15α -diacetoxy- 17α -pregna-4,20-dien-3-one.

The other isolated compound, 17α -pregna-4,20-dien-3-one (2), has been obtained by chemical transformations.¹⁰⁻¹⁴ To the best of our knowledge, 2 has not been isolated previously from natural sources. The ¹³C NMR data and the detailed assignments of ¹H NMR spectral data for compound 2 are disclosed (Tables 1 and 2) for the first time by extensive analyses of a series of 2D NMR (HMQC, HMBC, ¹H-¹H COSY and NOESY) spectral data.

EXPERIMENTAL SECTION

General Methods

Melting points were determined using a Fisher-Johns melting point apparatus and were uncorrected. Optical rotations were measured on a Jasco DIP-370 digital polarimeter. UV spectra were taken in MeOH on a Perkin Elmer UV/VIS Lambda EZ 201 spectrometer. LRMS were obtained with a VG Quattro GC/MS spectrometer. HRMS were recorded on a JEOL JMS-SX/SX 102A mass spectrometer. The NMR spectra were recorded on a Varian UNITY INOVA 500 FT-NMR at 500 MHz for ¹H and 125 MHz for ¹³C, in CDCl₃ using TMS



Fig. 2. Key NOESY Correlations for 1.

as an internal standard. Silica gel (Merck, 230-400 mesh) was used for column chromatography. Precoated silica gel plates (Merck, Kieselgel 60 F_{254} , 0.2 mm) were used for analytical TLC. High performance liquid chromatography (HPLC) was achieved on a Hitachi L-7100 apparatus equipped with a Hitachi L-7400 UV detector, by using normal phase column (250 × 4.6 mm 5 µm Hypersil HS silica).

Organism

The gorgonian *S. mollis* was collected by hand using scuba along the coast of southern Taiwan, in February 2001, at a depth of 18 m, and was stored in a freezer until extraction. A voucher specimen was deposited in the Department of Marine Resources, National Sun Yat-Sen University (specimen no. SC020704).

Extraction and Isolation

The gorgonian *S. mollis* (776 g fresh wt) was freezedried, minced and subsequently extracted with EtOAc (3 L × 6). The organic extract was evaporated under reduced pressure to give a residue (8.11 g). The residue was separated by Si gel column chromatography using *n*-hexane and *n*-hexane-EtOAc mixtures of increasing polarity to yield 32 fractions. Fraction 8 (52 mg) eluted with hexanes/EtOAc (18:1) was rechromatographed to yield compound **2** (20.5 mg). Fraction 22 (75 mg) eluted with *n*-hexane/EtOAc (5:1) was further chromatographed on Si gel by HPLC using *n*-hexanes-EtOAc (7:3) to yield **1** (1.3 mg).

11α , 15α -diacetoxy- 17α -pregna-4, 20-dien-3-one (1)

White solid; mp: 100-102 °C; $[\alpha]_D^{27}$: +61.4° (*c* 0.44, CHCl₃); UV (MeOH) λ_{max} (log ε) 243 (4.08) nm; IR (CHCl₃) ν_{max} 1726, 1667 cm⁻¹; ¹H and ¹³C NMR spectral data, see Tables 1 and 2; FABMS *m/z* (rel. int %) 415 [0.2, (M + H)⁺], 355 [0.2, (M + H - HOAc)⁺], 295 [0.3, (M + H - 2 HOAc)⁺]; HRFABMS *m/z* 415.2474 (calcd. for C₂₅H₃₄O₅ + H, 415.2486).

17α -Pregna-4,20-dien-3-one (2)

White solid; mp: 72-73 °C; $[\alpha]_D^{27}$: +57.0° (*c* 2.1, CHCl₃); UV (MeOH) λ_{max} (log ε) 241 (4.09) nm; IR (CHCl₃) ν_{max} 1663 cm⁻¹; ¹H and ¹³C NMR spectral data, see Tables 1 and 2; EIMS *m/z* (rel. int %) 298 [28, M⁺], 283 (21), 229 (22), 121 (31), 91 (100); HREIMS *m/z* 298.2306 (calcd. for C₂₁H₃₀O, 298.2298).

Received March 5, 2003.

ACKNOWLEDGEMENT

We gratefully acknowledge the financial support of this work by a grant from the National Science Council of Taiwan, Republic of China (Contract No. NSC-90-2113-M-110-023) awarded to J.-H. Sheu.

REFERENCES

- 1. Kashman, Y. Tetrahedron 1979, 35, 263.
- 2. Bunko, J. D.; Ghisalberti, E. L.; Jefferies, P. R. Aust. J. Chem. 1981, 34, 2237.
- Groweiss, A.; Fenical, W.; He, C.-H.; Clardy, J.; Wu, Z.; Yiao, Z.; Long, K. *Tetrahedron Lett.* 1985, 20, 2379.
- Bokesch, H. R.; McKee, T. C.; Cardellina II, J. H.; Boyd, M. R. *Tetrahedron Lett.* **1996**, *37*, 3259.
- 5. Parameswaran, P. S.; Naik, C. G.; Kamat, S. Y.; Puar, S. Y.;

Das, P.; Hegde, V. R. J. Nat. Prod. 1998, 61, 832.

- Bokesch, H. R.; Blunt, J. W.; Westergaard, C. K.; Cardellina II, J. H.; Johnson, T. R.; Michael, J. A.; McKee, T. C.; Hollingshead, M. G.; Boyd, M. R. *J. Nat. Prod.* 1999, *62*, 633.
- Wang, G.-H.; Ahmed, A. F.; Sheu, J.-H.; Duh, C.-Y.; Shen, Y.-C.; Wang, L.-T. J. Nat. Prod. 2002, 65, 887.
- Wang, G.-H.; Ahmed, A. F.; Kuo, Y.-H.; Sheu, J.-H. J. Nat. Prod. 2002, 65, 1033.
- Anjaneyulu, A. S. R.; Kameswara Rao, N. S.; Venkateswara Rao, G. *Indian J. Chem.* 1997, 36B, 418.
- Julian, P. L.; Meyer, E. W.; Printy, H. C. J. Am. Chem. Soc. 1948, 70, 887.
- 11. Krieger, B.; Kaspar, E. Chem. Ber. 1967, 100, 1169.
- Krubiner, A. M.; Gottfried, N.; Oliveto, E. P. J. Org. Chem. 1969, 34, 3502.
- 13. Steven, R. S.; Trevor, C. M. Steroids. 1977, 30, 389.
- Johnson, W. S.; Chen, Y.-Q.; Kellogg, M. S. C. J. Am. Chem. Soc. 1983, 105, 6653.

