Polyoxygenated Steroids from the Gorgonian Isis hippuris

Chih-Hua Chao,[†] Long-Fei Huang,[†] Yi-Lea Yang,[†] Jui-Hsin Su,[†] Guey-Horng Wang,[‡] Michael Y. Chiang,[§] Yang-Chang Wu,[⊥] Chang-Feng Dai,[∥] and Jyh-Horng Sheu^{*,†}

Department of Marine Resources, National Sun Yat-Sen University, Kaohsiung 804, Taiwan, Republic of China, Center of General Education, Hsing-Kuo University, Tainan 709, Taiwan, Republic of China, Department of Chemistry, National Sun Yat-Sen University, Kaohsiung 804, Taiwan, Republic of China, Graduate Institute of Natural Products, Kaohsiung Medical University, Kaohsiung 807, Taiwan, Republic of China, and Institute of Oceanography, National Taiwan University, Taipei 106, Taiwan, Republic of China.

Received January 28, 2005

Eleven new hippuristanols (1-4, 7-9, and 11-14), along with eight known metabolites (5, 6, 10, and 15-19), have been isolated from the gorgonian coral *Isis hippuris*. Single-crystal X-ray diffraction analyses supported the structure elucidation of known steroids 5 and 10. The absolute structures of hippuristanols were established by application of modified Mosher's method on 19. Compounds 14-19 have been found to exhibit significant cytotoxicity against several cancer cell lines.

Previous studies on *Isis hippuris* have resulted in the isolation of a series of novel metabolites, including highly oxygenated spiroketal steroids that were named hippurins or hippuristanols,^{1–7,9} polyoxygenated gorgosteroids,^{5,6} (22*R*,-23*S*,24*S*)-polyoxygenated steroids,^{7–9} and suberosane-type sesquiterpenes.¹⁰ Some hippuristanols⁹ and suberosane-type sesquiterpenes have been reported to have significant cytotoxicity against several cancer cell lines. Our continuing investigation on the chemical constituents of *I. hippuris*, collected by hand using scuba at Green Island, located off the southeast coast of Taiwan, in February 1999, has again afforded a series of hippuristanols. We describe herein the isolation, structure elucidation, and biological activity of these compounds.

Results and Discussion

The gorgonian coral *I. hippuris* was frozen immediately after collection, and the freeze-dried organism was extracted successively with *n*-hexane and CH_2Cl_2 to afford a crude extract. The crude extract was repeatedly purified by extensive column chromatography on silica gel and afforded 11 new (1-4, 7-9, and 11-14) and eight known steroids (5, 6, 10, and 15-19, see Figure 1).

Both steroids **5** and **10** have been reported in the literatures,³ but without supporting their structures by X-ray diffraction analyses. The results (Figure 2) of our study on single-crystal X-ray diffraction for both **5** and **10** further confirmed the overall structures of these two metabolites.

The absolute configuration of hippuristanols was determined by application of the modified Mosher's method.¹³ Compound **19** was treated with (*R*)- or (*S*)- α -methoxy- α trifluoromethylphenylacetic acid [(*R*)- or (*S*)-MTPA] in the presence of 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide hydrochloride (EDC·HCl) and 4-(dimethylamino)pyridine (4-DMAP) to yield the (*R*)- and (*S*)-MTPA esters (**19b** and **19a**), respectively. The MTPA esters selectively formed at C-2 were elucidated from the ¹H NMR chemical shifts and



Figure 1. Structures of metabolites 1-20.

coupling constants of H-2 in **19a** and **19b** (**19a**, δ 5.23, 1H, br d, J = 12.2 Hz, H-2; **19b**, δ 5.23, 1H, br d, J = 12.2 Hz, H-2), due to less hindrance of the equatorial hydroxy group attached at C-2. Comparison of the ¹H NMR chemical shifts for **19a** and **19b** (Δ values shown in Figure 3) led to the assignment of the *R*-configuration at C-2. Therefore, the absolute structure of **19** was determined as shown in formula **19**. Because of biogenic considerations, the absolute configurations at C-3, C-5, C-8, C-9, C-10, C-13, C-14, C-16, C-17, C-20, and C-24 of other hippuristanols, reported or discovered by the present study, were assumed to be identical with those of **19**.

Compound 1 was isolated as a white powder. The HRFABMS of 1 established a molecular formula of $C_{32}H_{46}O_{9}$.

10.1021/np050033y CCC: \$30.25 © 2005 American Chemical Society and American Society of Pharmacognosy Published on Web 05/25/2005

^{*} To whom correspondence should be addressed. Tel: 886-7-5252000 ext. 5030. Fax: 886-7-5255020. E-mail: sheu@mail.nsysu.edu.tw, sheu@ mail.nsysu.edu.tw.

[†] Department of Marine Resources, National Sun Yat-Sen University. [‡] Hsing-Kuo University.

[§] Department of Chemistry, National Sun Yat-Sen University.

¹ Kaohsiung Medical University.

[&]quot;National Taiwan University.

	1	1	, ,			
C #	1^b	2^{a}	3^{a}	4^{b}	13^{a}	14^{a}
1	$37.5 (CH_2)^c$	$40.2 (CH_2)^c$	$36.6 (CH_2)^c$	$41.2 (CH_2)^c$	$31.0 (CH_2)^c$	$41.2 ({ m CH_2})^c$
2	69.7 (CH)	68.8 (CH)	72.4 (CH)	67.8 (CH)	$29.0 (CH_2)$	67.9 (CH)
3	69.4 (CH)	69.0 (CH)	67.4 (CH)	72.9 (CH)	66.4 (CH)	73.0 (CH)
4	$31.8 (CH_2)$	$33.4 (CH_2)$	$33.3 (CH_2)$	$31.7 (CH_2)$	$35.4 (CH_2)$	$31.8 (CH_2)$
5	40.1 (CH)	38.7 (CH)	38.7 (CH)	40.0 (CH)	39.0 (CH)	40.2 (CH)
6	$26.6 (CH_2)$	$26.7 (CH_2)$	$26.6 (CH_2)$	$26.6 (CH_2)$	$27.9 (CH_2)$	$26.9 (CH_2)$
7	$32.2 (CH_2)$	$32.2 (CH_2)$	$32.2 (CH_2)$	$32.3 (CH_2)$	$32.7 (CH_2)$	$32.3 (CH_2)$
8	29.1 (CH)	29.1 (CH)	29.1 (CH)	29.1 (CH)	35.9 (CH)	29.6 (CH)
9	57.7 (CH)	57.6 (CH)	57.7 (CH)	57.7 (CH)	64.7 (CH)	58.1 (CH)
10	37.6 (C)	37.3 (C)	37.5 (C)	37.7 (C)	35.9 (C)	37.1 (C)
11	67.0 (CH)	67.0 (CH)	66.9 (CH)	67.0 (CH)	210.5 (C)	68.1 (CH)
12	$40.7 (CH_2)$	$40.6 (CH_2)$	$40.6 (CH_2)$	$40.7 (CH_2)$	$58.5 (CH_2)$	$49.9 (CH_2)$
13	53.0 (C)	53.0 (C)	52.9 (C)	53.1 (C)	46.0 (C)	42.2 (C)
14	56.7 (CH)	56.7 (CH)	56.6 (CH)	56.7 (CH)	56.2 (CH)	56.9 (CH)
15	$35.0 (CH_2)$	$34.9 (CH_2)$	$34.9 (CH_2)$	$35.0 (CH_2)$	$31.5 (CH_2)$	$34.0 (CH_2)$
16	80.3 (CH)	80.2 (CH)	80.2 (CH)	80.3 (CH)	79.2 (CH)	80.1 (CH)
17	60.0 (CH)	59.9 (CH)	59.9 (CH)	60.0 (CH)	63.2 (CH)	66.1 (CH)
18	182.4 (C)	182.5(C)	182.4 (C)	182.5(C)	$17.7 (CH_3)$	18.6 (CH ₃)
19	$15.1 (CH_3)$	$14.9 (CH_3)$	$14.8 (CH_3)$	$15.2 (CH_3)$	$11.0 (CH_3)$	$15.3 (CH_3)$
20	90.3 (C)	90.2 (C)	90.2 (C)	90.2 (C)	81.9 (C)	79.2 (C)
21	$18.9 (CH_3)$	$18.8 (CH_3)$	$18.8 (CH_3)$	$18.9 (CH_3)$	$25.9 (CH_3)$	$29.1 (CH_3)$
22	116.7 (C)	116.7 (C)	116.7 (C)	116.7 (C)	118.7 (C)	115.3 (C)
23	$38.6 (CH_2)$	$38.5 (CH_2)$	$38.5 (CH_2)$	$38.6 (CH_2)$	$39.6 (CH_2)$	$40.9 (CH_2)$
24	41.2 (CH)	41.1 (CH)	41.1 (CH)	41.2 (CH)	41.0 (CH)	41.9 (CH)
25	85.7 (C)	85.6 (C)	85.6 (C)	85.7 (C)	84.5 (C)	84.6 (C)
26	$29.2 (CH_3)$	$29.0 (CH_3)$	$29.0 (CH_3)$	$29.0 (CH_3)$	$29.1 (CH_3)$	$28.4 (CH_3)$
27	$23.1 (CH_3)$	$23.0 (CH_3)$	$23.0 (CH_3)$	$23.1 (CH_3)$	$23.0 (CH_3)$	$23.0 (CH_3)$
28	$14.0 (CH_3)$	$13.9 (CH_3)$	$13.9 (CH_3)$	$14.0 (CH_3)$	$14.0 (CH_3)$	$14.7 (CH_3)$
OAc	170.4 (C)		169.9 (C)	171.5 (C)		171.5 (C)
	170.3 (C)					
OAc	$21.3 (CH_3)$		$21.3 (CH_3)$	$21.4 (CH_3)$		$21.4 (CH_3)$
	$21.1 (CH_3)$					

^{*a*} Spectra recorded at 125 MHz in CDCl₃ at 25°C. ^{*b*} 75 MHz in CDCl₃ at 25°C. ^{*c*} Multiplicity deduced by DEPT and indicated by the usual symbols. The values are in ppm downfield from TMS.



Figure 2. Computer-generated ORTEP drawings of compounds 5 and 10. Hydrogen atoms were omitted for clarity.

This compound was shown to be a member of hippurins by the presence of a spiroketal functionality ($\delta_{\rm C}$ 116.7, s).³ The NMR spectral data (Tables 1 and 3) also showed the presence of a lactone carbonyl ($\delta_{\rm C}$ 182.4, s) and were found to be similar to those of **5**.³ Furthermore, the presence of an additional acetoxy group at C-2 ($\delta_{\rm H}$ 5.02, br d, J = 11.4Hz; $\delta_{\rm C}$ 69.7, d) was supported by COSY (H₂-1/H-2; H-2/H-3) and HMBC cross-peaks (H₃-19/C-1). The orientations of two acetoxy groups in ring A were determined to be α , as evidenced by a large coupling constant of H-2 (br d, J =11.4 Hz) and a small coupling constant of H-3 (br d, J =2.2 Hz). Also, by comparison of the NMR data, including chemical shifts and coupling constants, with those of the known steroid 3-acetyl-22-*epi*-hippurin-1 (**20**),⁴ which possessed the same structure as that of **1** in ring A, the



$\Delta \delta = \delta(S) - \delta(R) (ppm)$

Figure 3. ¹H NMR chemical shift differences [$\delta(S)$ -MTPA – $\delta(R)$ -MTPA] of the MTPA esters.

 α -orientation for both acetoxy groups in ring A was unambiguously determined. Thus, the structure of **1** was deduced as (22*S*)-2 α ,3 α -diacetoxy-11 β -hydroxy-24-methyl-22,25-epoxy-5 α -furostan-18,20 β -lactone (1).

Compound **2** was isolated as a white powder. The HRFABMS of **2** established a molecular formula of $C_{28}H_{42}O_7$. By comparison of the NMR data of **2** with those of **1** the structure of **2** was deduced as $(22S)-2\alpha,3\alpha$ -dihydroxy-11 β -hydroxy-24-methyl-22,25-epoxy-5 α -furostan-18,20 β -lactone (**2**). Moreover, hydrolysis of **1** was found to get **2** as the major product and further confirmed the structure of **2**.

The molecular formula $C_{30}H_{44}O_8$ of **3** was established by the HRFABMS spectrum. By comparison of NMR data of **3** with those of **1** and **2**, together with the elucidation of the COSY (H₂-1/H-2; H-2/H-3) and HMBC cross-peaks (H₃-19/C-1), the structure of **3** was fully established. Furthermore, hydrolysis of **1** also formed **3** as a minor component.

Table 2. ¹³C NMR Spectral Data of Compounds 7-9, 11, and 12

C #	7 ^b	8 ^a	9 <i>a</i>	11 ^b	12^{b}
1	$38.3 (CH_2)^c$	$41.7 (CH_2)^c$	$37.3 (CH_2)^c$	$32.6 (CH_2)^c$	$37.8~({ m CH_2})^c$
2	69.6 (CH)	67.6 (CH)	72.4 (CH)	$25.7 (CH_2)$	69.8 (CH)
3	69.2 (CH)	72.7 (CH)	67.3 (CH)	70.1 (CH)	69.4 (CH)
4	$31.8 (CH_2)$	$31.6 (CH_2)$	$33.3 (CH_2)$	$32.2 (CH_2)$	$31.8 (CH_2)$
5	39.5 (CH)	39.4 (CH)	38.1 (CH)	40.6 (CH)	40.0 (CH)
6	$27.5 (CH_2)$	$27.4 (CH_2)$	$27.5 (CH_2)$	$27.5 (CH_2)$	$26.6 (CH_2)$
7	$32.2 (CH_2)$	$32.1 (CH_2)$	$32.2 (CH_2)$	$33.8 (CH_2)$	$32.0 (CH_2)$
8	39.0 (CH)	38.9 (CH)	38.9 (CH)	31.2 (CH)	30.5 (CH)
9	57.9 (CH)	57.7 (CH)	57.8 (CH)	58.6 (CH)	58.6 (CH)
10	37.4 (C)	37.1 (C)	37.4 (C)	36.0 (C)	37.3 (C)
11	81.0 (CH)	81.0 (CH)	80.9 (CH)	66.1 (CH)	66.2 (CH)
12	39.6 (CH ₂)	$39.5 (CH_2)$	39.6 (CH ₂)	$39.3 (CH_2)$	$39.2 (CH_2)$
13	63.7 (C)	63.7 (C)	63.7 (C)	56.1 (C)	55.9 (C)
14	49.6 (CH)	49.4 (CH)	49.5 (CH)	56.3 (CH)	56.4 (CH)
15	$36.2 (CH_2)$	$36.2 (CH_2)$	$36.2 (CH_2)$	$33.8 (CH_2)$	$33.9 (CH_2)$
16	81.6 (CH)	81.9 (CH)	81.6 (CH)	80.7 (CH)	80.7 (CH)
17	57.4 (CH)	57.3 (CH)	57.3 (CH)	64.6 (CH)	64.4 (CH)
18	107.7 (CH)	107.5 (CH)	107.6 (CH)	101.4 (CH)	101.6 (CH)
19	$12.8 (CH_3)$	$13.0 (CH_3)$	$12.8 (CH_3)$	$14.5 (CH_3)$	$15.6 (CH_3)$
20	93.5 (C)	93.4 (C)	93.4 (C)	90.8 (C)	91.1 (C)
21	$20.8 (CH_3)$	$20.7 (CH_3)$	$20.8 (CH_3)$	$19.7 (CH_3)$	$19.7 (CH_3)$
22	117.1 (C)	117.0 (C)	117.0 (C)	118.0 (C)	118.0 (C)
23	$39.0 (CH_2)$	$39.0 (CH_2)$	$38.9 (CH_2)$	$38.9 (CH_2)$	$38.9 (CH_2)$
24	41.0 (CH)	40.9 (CH)	41.0 (CH)	41.2 (CH)	41.2 (CH)
25	85.2 (C)	85.1 (C)	85.1 (C)	84.8 (C)	84.9 (C)
26	29.3 (CH ₃)	29.2 (CH ₃)	$29.2 (CH_3)$	$29.2 (CH_3)$	$29.2 (CH_3)$
27	$23.1 (CH_3)$	$23.0 (CH_3)$	$23.0 (CH_3)$	$23.0 (CH_3)$	$23.0 (CH_3)$
28	$14.1 (CH_3)$	$14.0 (CH_3)$	$14.0 (CH_3)$	$14.1 (CH_3)$	$14.3 (CH_3)$
OAc	170.4 (C)	171.5 (C)	169.9 (C)	170.7 (C)	170.5 (C)
	170.3 (C)				170.5 (C)
OAc	$21.2 (CH_3)$ $21.1 (CH_2)$	$21.3 \left(CH_3 \right)$	$21.3 \; (CH_3)$	$21.6 (CH_3)$	$21.3 (CH_3)$ $21.2 (CH_3)$

^{*a*} Spectra recorded at 125 MHz in CDCl₃ at 25°C. ^{*b*} 75 MHz in CDCl₃ at 25°C. ^{*c*} Multiplicity deduced by DEPT and indicated by the usual symbols. The values are in ppm downfield from TMS.

Table 3.	¹ H NMR	Spectral	Data	of Com	pounds	1 - 4	13,	and 1	14
----------	--------------------	----------	------	--------	--------	-------	-----	-------	----

	•	•	, ,			
C #	1^{b}	2^{a}	3^{a}	4^{b}	13^{a}	14^{a}
1	1.58m	1.40 t (12.0)	1.56 m	1.35 t (12.1)	1.21m	1.31 m
	1.83 m	1.94 dd (12.0, 5.0)	1.92 dd (11.5, 4.5)	2.06 m	2.27 m	1.98 m
2	5.02 br d (11.4)	3.86 br d (12.0)	5.04 br d (12.0)	3.92 br d (12.2)	1.74 m	3.90 m
					1.54 m	
3	5.28 br d (2.2)	3.97 br s	4.06 br s	5.14 br d (2.3)	4.04 br s	5.12 br s
4	1.53 m	1.54 m	2.35 m	1.59 m	1.53 m	1.55 m
	1.36 m		1.59 m		1.37 m	1.57 m
5	1.49 m	1.50 m	1.59 m	1.52 m	1.52 m	1.53 m
6	1.28 m	1.27 m	1.27 m	1.25 m	1.19 m	1.26 m
7	1.95 m	1.85 m	1.85 m	1.85 m	1.80 m	1.81 m
	0.88 m	0.89 m	0.91 m	0.88 m	1.13 m	0.90 m
8	2.50 m	2.48 m	2.50 m	2.50 m	1.90 m	1.95 m
9	0.90 m	0.93 m	0.94 m	0.94 m	1.70 m	0.81 m
10						
11	4.13 dd (12.6, 3.3)	4.20 br d (12.0)	4.15 dd (13.0, 2.5)	4.20 br d (12.6)		4.29 br s
12	2.40 m	1.70 dd (14.0, 4.0)	1.68 dd (14.0, 4.0)	1.71 dd (14.0, 3.6)	2.51 d (12.0)	2.18 br d (14.0)
	1.71 m	2.37 dd (14.0, 2.0)	2.36 dd (14.0, 2.0)	2.38 br d (14.2)	2.25 m	1.40 m
13						
14	1.51 m	1.49 m	1.50 m	1.50 m	1.67 m	0.88 m
15	2.26 m	2.32 m	2.31 m	2.31 m	2.10 m	2.02 m
	1.37 m	1.23 m	1.23 m	1.23 m	1.40 m	1.43 m
16	4.66 dt (6.6, 6.6)	4.68 dt (8.0, 7.5)	4.66 dt (8.0, 7.0)	4.67 dt (7.4, 6.9)	4.49 dt (8.0, 7.5)	4.32 m
17	2.66 d (8.6)	2.67 d (8.5)	2.66 d (8.0)	2.67 d (8.3)	2.12 d (8.0)	1.74 d (7.0)
18					1.05 s	1.39 s
19	1.17 s	1.11 s	1.16 s	1.13 s	1.01 s	$1.07 \ {\rm s}$
20						
21	$1.47 \mathrm{~s}$	$1.48 \mathrm{~s}$	$1.48 \mathrm{~s}$	1.48 s	$1.29 \mathrm{~s}$	$1.31 \mathrm{~s}$
22						
23	2.11 m	2.12 dd (13.0, 6.0)	2.12 dd (13.0, 6.5)	2.12 m	2.02 dd (13.0, 7.0)	2.40 dd (12.0, 5.0)
	1.79 m	1.80 t (13.0)	1.80 t (13.0)	1.80 t (13.0)	1.77 t (13.0)	1.70 m
24	2.30 m	2.31 m	2.30 m	2.30 m	2.23 m	1.88 m
25						
26	1.29 s	1.30 s	1.30 s	1.30 s	$1.28 \mathrm{~s}$	$1.22 \ s$
27	0.99 s	1.00 s	1.00 s	1.00 s	0.98 s	1.20 s
28	0.96 d (6.8)	0.96 d (6.5)	0.96 d (6.5)	0.97 d (6.9)	0.94 d (7.0)	0.98 d (7.0)
11-OH	3.95 d (12.6)	4.00 d (12.5)	4.00 d (12.5)	3.98 d (12.7)	* <i>*</i>	3.19 s (20-OH)
OAc	2.04 s		2.01 s	2.10 s		2.11 s
	1.97 s					

^a Spectra recorded at 500 MHz in CDCl₃ at 25 °C. ^b 300 MHz in CDCl₃ at 25 °C. The values are in ppm downfield from TMS.

Table 4. ¹ H NMR Spectral Data of Compounds 7–9, 11, and	l 12
--	------

C #	7^{b}	8^{a}	9^{a}	11^b	12^b
1	1.58 m	1.20 t (12.0)	1.41 t (12.0)	1.34 m	1.50 m
	1.80 m	1.94 dd (12.0, 4.5)	1.80 m	1.70 m	1.87 m
2	5.02 br d (11.1)	3.92 br d (12.0)	5.01 br d (12.0)	1.70 m	5.00 br d (11.3)
3	5.29 br d (2.6)	5.13 br d (3.0)	4.05 br d (2.0)	5.01 br s	5.28 br s
4	1.55 m	1.50 m	1.54 m	1.46 m	1.54 m
5	1.53 m	1.33 m	1.54 m	1.47 m	1.47 m
6	1.26 m	1.24 m	1.36 m	1.18 m	1.26 m
7	1.78 m	1.80 m	1.78 m	1.82 m	1.87 m
	1.10 m	1.06 m	1.07 m	0.94 m	0.97 m
8	1.70 m	1.65 m	1.64 m	1.88 m	1.89 m
9	0.85 d (10.8)	0.88 br d (11.0)	0.90 br d (11.0)	0.81 br d (11.1)	0.86 br d (10.8)
10					
11	4.77 br d (5.4)	4.82 br d (5.5)	4.76 br d (5.5)	4.29 br s	$4.24 \mathrm{\ br\ s}$
12	2.30 m	2.30 dd (11.5, 5.5)	2.28 dd (10.5, 5.5)	2.71 d (12.0)	2.72 d (13.9)
	1.40 m	1.38 br d (11.0)	1.37 br d (10.0)	1.68 m	1.68 m
13					
14	1.75 m	1.75 m	1.74 m	1.48 m	1.48 m
15	2.22 m	2.35 m	2.34 m	2.11 m	2.09 m
	1.37 m	1.53 m	1.56 m	1.30 m	1.30 m
16	4.58 dt (6.8, 1.7)	4.59 dt (7.0, 2.0)	4.58 dt (7.0, 2.0)	4.53 dt (7.9, 7.2)	4.54 dt (8.0, 6.6)
17	2.61 d (6.9)	2.62 d (6.5)	2.61 d (7.0)	2.62 d (8.0)	2.62 d (8.3)
18	$5.31~\mathrm{s}$	$5.31~\mathrm{s}$	$5.31 \mathrm{~s}$	$5.32 \mathrm{s}$	$5.33 \mathrm{s}$
19	$0.91 \mathrm{s}$	0.94 s	0.97 s	1.02 s	1.09 s
20					
21	$1.37 \mathrm{~s}$	$1.37 \mathrm{~s}$	$1.37 \mathrm{~s}$	$1.43 \mathrm{~s}$	$1.43 \mathrm{~s}$
22					
23	2.05 m	2.04 dd (13.5, 6.5)	2.05 dd (13.0, 6.5)	2.09 m	2.10 m
	1.75 m	1.72 m	1.74 m	1.72 m	1.75 m
24	2.25 m	2.24 m	2.25 m	2.28 m	2.26 m
25					
26	$1.28 \mathrm{~s}$	$1.28 \mathrm{~s}$	$1.28 \mathrm{~s}$	$1.30 \mathrm{~s}$	$1.30 \mathrm{~s}$
27	0.98 s	0.98 s	0.97 s	$0.97 \mathrm{s}$	0.98 s
28	0.94 d (6.8)	0.94 d (6.5)	0.94 d (7.0)	0.94 d (6.9)	0.91 d (7.2)
OAc	$2.04 \mathrm{~s}$	$2.10 \mathrm{~s}$	2.08 s	$2.04 \mathrm{~s}$	2.09 s
	1.98 s				1.99 s

^a Spectra recorded at 500 MHz in CDCl₃ at 25°C. ^b 300 MHz in CDCl₃ at 25 °C. The values are in ppm downfield from TMS.

Thus, the structure of **3** was deduced as (22S)- 2α -acetoxy- 3α -hydroxy- 11β -hydroxy-24-methyl-22,25-epoxy- 5α -furostan- $18,20\beta$ -lactone (**3**).

Compound 4 was obtained as a white powder. It gave the same formula, $C_{30}H_{44}O_8$, as that of 3 from the HR-FABMS spectrum and revealed that 4 was an isomer of 3. Hydrolysis of 1 also obtained 4 as a minor component. Thus, from the above observations and by comparison of the ¹H and ¹³C NMR spectral data (Tables 1 and 3) with those of metabolites 1–3, the structure of 4 was deduced as (22S)-2 α -hydroxy-3 α -acetoxy-11 β -hydroxy-24-methyl-22,25-epoxy-5 α -furostan-18,20 β -lactone (4).

The formula of compound **7** was found to be $C_{32}H_{46}O_8$ as deduced from the HRFABMS spectrum. It was found to contain an additional acetoxy group at C-2 (δ_H 5.02, br d, J = 11.1 Hz, and δ_C 69.6, d) by COSY (H₂-1/H-2; H-2/H-3) and HMBC cross-peaks (H₃-19/C-1) as compared to **10**.³ The two acetoxy groups attached at C-2 and C-3 were both placed on the α face by comparison of the chemical shifts and large coupling constants of H-2 and H-3 with those of **1**. Furthermore, the NMR spectra of **7** are similar to those of **10**³ (Tables 2 and 4), except that an additional acetoxy group was found to be present at C-2 of **7**. Thus, the structure of **7** was deduced as (22S)-2 α ,3 α -diacetoxy-24methyl-11 β ,18;18, 20 β ;22,25-triepoxy-5 α -furostane (**7**).

The molecular formula of compound 8, $C_{30}H_{44}O_7$, was deduced from the HRFABMS spectrum. It was found that 8 contained an additional oxygen atom as compared to 10.³ The above evidence together with the comparison of the NMR spectral data between 8 and 10³ (Tables 2 and 4) revealed that 8 had one more hydroxy group than 10. Furthermore, the analyses on the 2D NMR spectral data of 8 suggested that the hydroxy group should be positioned

	cancer cell line ^{<i>a</i>} (IC ₅₀ , μ g/mL)					
compound	Hep G2	Hep 3B	MCF-7	MDA-MB-231		
1	>20	b	19.59	>20		
2	>30	23.85	>30	>30		
3	27.60	>30	>30	>30		
5	>20	_	>20	>20		
6	>20	_	>20	>20		
7	>20	_	>20	>20		
10	>20	_	>20	>20		
11	>20	_	12.72	>20		
12	>20	_	11.39	>20		
14	0.72	0.46	1.07	0.21		
15	2.06	1.46	2.41	0.74		
16	0.56	0.10	0.53	0.41		
17	4.64	0.68	4.54	2.64		
18	0.08	0.10	0.20	0.13		
19	0.62	0.77	0.59	0.75		

Table 5. Cytotoxicities of 1-3, 5-7, 10, 12, and 14-19

 a Human hepatocellular carcinoma Hep G2 and Hep 3B; human breast carcinoma MCF-7 and MDA-MB-231, and human lung carcinoma A-549. b "—" not tested.

at C-2 ($\delta_{\rm C}$ 67.6, d). The α -orientation of 2-OH was determined by NOE correlations between both H-1 β and H₃-19 and H-2, and the large coupling constant of H-2 (J = 12.0 Hz). Thus, the structure of **8** was deduced as (22S)-2 α -hydroxy-3 α -acetoxy-24-methyl-11 β ,18;18,20 β ;22,25-triepoxy-5 α -furostane (**8**).

Compound **9**, $C_{30}H_{44}O_7$, was found to be an isomer of **8** from the HRFABMS spectrum, and the ¹H NMR data of **9** show clear differences at H-2 (5.01, br d, J = 12.0 Hz) and H-3 (4.05, br d, J = 2.0 Hz) as compared with those of **8** (see Tables 2 and 4). It was found that **9** has the same structure in ring A by comparison of the NMR spectral data with those of **3**. Therefore, the structure of **9** was concluded

to be (22S)-2 α -acetoxy,3 α -hydroxy-24-methyl-11 β ,18;18,-20 β ;22,25-triepoxy-5 α -furostane (**9**).

Compound 11 was isolated as a white powder and was found to be more polar than 10, a known metabolite also isolated in the present study. Its HRFABMS established the molecular formula C₃₀H₄₆O₇, implying eight degrees of unsaturation. Furthermore, an acetoxy group was observed from NMR signals appearing at $\delta_{\rm H}$ 2.04 (s), $\delta_{\rm C}$ 170.7 (s), and 21.6 (q). The above data suggested a heptacyclic structure in the molecule of 11. The NMR data of 11 were found to be close to those of 10,3 except that the carbon signal of C-18 (δ 107.7, d) in 10³ was converted to 101.8 (d) in **11** and that of C-11 (δ 81.0, d) in **10**³ was converted to 66.4 in 11 (see Tables 2 and 4). H-18 was assigned as having a β -orientation from the key NOE correlation between H-18 and H-8. Thus, the structure of 11 was established as (22S)-3 α -acetoxy-11 β ,18 α -dihydroxy-24methyl-18,20 β ;22,25-diepoxy-5 α -furostane (11).

Compound 12 was obtained as a white powder. The HREIMS of 12 established a molecular formula of $C_{32}H_{48}O_9$. The ¹H and ¹³C NMR spectral data of 12 were similar to those of 11 and revealed the presence of an additional acetoxy group as compared to 11. By comparison of NMR data of 12 in ring A with those of 1 and 7 (see Tables 1–4), the structure of 12 was fully determined and assigned as (22S)-2 α ,3 α -diacetoxy-11 β ,18 α -dihydroxy-24-methyl-18,-20 β ;22,25-diepoxy-5 α -furostane (12).

Compound 13 was obtained as a white powder, which gave a $[M + H]^+$ peak at m/z 461.3267 in the HRFABMS and thus established the molecular formula $C_{28}H_{44}O_5$. By comparison of the NMR spectral data of 13 with those of a known steroid, 22-*epi*-hippuristanol (18),² the structure of 13 was determined to be 22-*epi*-hippuristan-11-one, as the NMR spectrum of 13 (see Tables 2 and 4) did not show the signals at δ_H 4.30 (br s) and δ_C 68.0, attributable to the presence of a hydroxyl group at C-11 of 18,⁹ and instead showed the presence of a ketone functionality at δ_C 210.5 (s).

Compound 14 was obtained as a white powder that gave a $[M + H]^+$ peak at m/z 521.3478 in the HRFABMS spectrum. Thus, the molecular formula $C_{30}H_{48}O_7$ was established. By comparison of the NMR data of 14 with those of a known steroid, 3-acetyl-2-desacetyl-22-*epi*-hippurin-1 (15),⁴ it was found that the structure of 14 is similar to that of 15, except that the 22S configuration in 15 should be converted to 22R in 14, as the carbon signal at δ_C 118.6 (attributable to the presence of the 22S configuration) was replaced by a carbon signal at δ_C 115.3 (s) (attributable to the presence of the 22R configuration)⁹ (see Tables 2 and 4).

Except for compounds 4, 8, 9, and 13, the hippuristanols obtained in this study have been submitted for cytotoxicity evaluation toward cancer cell lines. The investigations showed that compounds possessing an ether or ester ring that links C-18 to C-20, such as 1-3, 5-7, and 10-12, would possess only weak cytotoxicity. Also, it was found that the hydroxy groups attached at both C-11 and C-20, such as in 14-19, could significantly enhance the cytotoxicity against the proliferation of the tested cancer cell lines.

Experimental Section

General Experimental Procedures. Optical rotations were measured on a Jasco DIP-1000 digital polarimeter. IR spectra were recorded on a Jasco FT-5300 infrared spectrophotometer. NMR spectra were recorded on a Bruker Avance DPX300 FT-NMR at 300 MHz for ¹H and 75 MHz for ¹³C or on a Varian Unity INOVA 500 FT-NMR at 500 MHz for ¹H and 125 MHz for 13 C, respectively, in CDCl₃ using TMS as internal standard. FABMS was obtained with a VG Quattro GC/MS spectrometer. HRMS spectra were recorded on a Finnigan MAT-95XL mass spectrometer. Silica gel (Merck, 230–400 mesh) was used for column chromatography. Precoated silica gel plates (Merck, Kieselgel 60 F-254, 0.2 mm) were used for analytical TLC. High-performance liquid chromatography (HPLC) was performed on a Hitachi L-7100 apparatus equipped with a Bischoff refractive index detector or a Hitachi L-7400 UV detector and with the Merck Hibar Si-60 column (250 \times 21 mm, 7 μ m).

Animal Material. The gorgonian coral *I. hippuris* was collected by hand using scuba at the Green Island, which is located off the southeast coast of Taiwan, in February 1999, at a depth of 25 m, and was stored in a freezer until extraction. A voucher specimen was deposited in the Department of Marine Resources, National Sun Yat-Sen University (specimen no. GISC-102).

Extraction and Isolation. The gorgonian coral (4.3 kg fresh wt) was collected and freeze-dried. The freeze-dried organism was minced and extracted exhaustively with nhexane and CH₂Cl₂. The combined organic extract was evaporated to give a dark green residue (37.0 g), which was chromatographed on a SiO₂ column using solvents of increasing polarity from *n*-hexane to EtOAc to obtain fractions 1-31. Fraction 21 was subjected to normal-phase HPLC column chromatography (gradient EtOAc/CH₂Cl₂, 7-10%) to afford compounds 1 (15 mg) and 5 (13 mg). Compounds 13 (2 mg), 18 (5 mg), and 20 (6 mg) were obtained from fraction 22 by repeated HPLC column chromatography (acetone/hexane, 15%). Fraction 23 was subjected to repeated normal-phase HPLC column chromatography (gradient acetone/CH₂Cl₂, 13-17%) to afford compounds 7 (30 mg), 10 (8 mg), 6 (9 mg), and 11 (9 mg). Similarly, fraction 24 was chromatographed (acetone/ CH₂Cl₂, 14%) to yield compound **12** (4 mg). Compounds **3**, **4**, 8, and 9 (each 1 mg) were eluted with MeOH/CH $_2$ Cl $_2$ (4–8%) from fraction 25. Repeated chromatography of fraction 28 over HPLC column (acetone/hexane, 25%) led to the isolation of compounds 14-17 (each 3 mg). Compounds 2 (1 mg) and 19 (6 mg) were both obtained by elution with MeOH/CH₂Cl₂ (gradient, 3-10%) from fraction 31 and fraction 30, respectively

(22S)-2α,3α-Diacetoxy-11β-hydroxy-24-methyl-22,25epoxy-5α-furostan-18,20β-lactone (1): white powder; mp 270–272 °C; $[\alpha]_D = 8^\circ$ (*c* 0.32, CHCl₃); IR (KBr) ν_{max} 3479, 1736, and 1263 cm⁻¹; ¹H NMR and ¹³C NMR data, see Tables 1 and 3; FABMS *m/z* 575 ([M + H]⁺, 2); HRFABMS *m/z* 575.3223 [M + H]⁺ (calcd for C₃₂H₄₇O₉, 575.3221).

(22S)-2α,3α-Dihydroxy-11β-hydroxy-24-methyl-22,25epoxy-5α-furostan-18,20β-lactone (2): white powder; mp 253–254 °C; $[α]_D = 27^\circ$ (*c* 0.92, CHCl₃); IR (KBr) ν_{max} 3445, and 1734 cm⁻¹; ¹H NMR and ¹³C NMR data, see Tables 1 and 3; FABMS *m/z* 491 ([M + H]⁺, 1), 473 (3), 455 (1), 437 (2), 391 (3), and 149 (57); HRFABMS *m/z* 491.3012 [M + H]⁺ (calcd for C₂₈H₄₃O₇, 491.3010).

(22S)-2α-Acetoxy-3α-hydroxy-11β-hydroxy-24-methyl-22,25-epoxy-5α-furostan-18,20β-lactone (3): white powder; mp 204–205 °C; $[\alpha]_D - 17^\circ$ (*c* 1.28, CHCl₃); IR (KBr) ν_{max} 3481 and 1736 cm⁻¹; ¹H NMR and ¹³C NMR data, see Tables 1 and 3; FABMS *m/z* 533 ([M + H]⁺, 48), 515 (82), 455 (25), and 437 (53); HRFABMS *m/z* 533.3108 [M + H]⁺ (calcd for C₃₀H₄₅O₈, 533.3116).

(22S)-2α-Hydroxy,3α-acetoxy-11β-hydroxy-24-methyl-22,25-epoxy-5α-furostan-18,20β-lactone (4): white powder; mp 244–245 °C; $[\alpha]_D$ +4° (*c* 1.40, CHCl₃); IR (KBr) ν_{max} 3449 and 1736 cm⁻¹; ¹H NMR and ¹³C NMR data, see Tables 1 and 3; FABMS *m/z* 533 ([M + H]⁺, 3); HRFABMS *m/z* 533.3116 [M + H]⁺ (calcd for C₃₀H₄₅O₈, 533.3116).

(22S)-2α,3α-Diacetoxy-24-methyl-11β,18;18,20β;22,25triepoxy-5α-furostane (7): white powder; mp 252–253 °C; $[α]_D - 22^\circ$ (*c* 0.31, CHCl₃); IR (KBr) ν_{max} 1726 and 1263 cm⁻¹; ¹H NMR and ¹³C NMR data, see Tables 2 and 4; FABMS *m/z* 559 ([M + H]⁺, 44), 449 (10), 439 (18); HRFABMS *m/z* 559.3271 [M + H]⁺ (calcd for C₃₂H₄₇O₈, 559.3272).

(22S)-2 α -Hydroxy-3 α -acetoxy-24-methyl-11 β ,18;18,-20β;22,25-triepoxy-5α-furostane (8): white powder; mp 193–194 °C; [α]_D –30° (c 0.83, CHCl₃); IR (KBr) ν_{max} 3447, 1734, and 1246 cm⁻¹; ¹H NMR and ¹³C NMR data, see Tables 2 and 4; FABMS m/z 517 ([M + H]+, 0.5), 499 (2), and 439 (0.6); HRFABMS m/z 517.3155 [M + H]⁺ (calcd for C₃₀H₄₅O₇, 517.3167).

(22S)-2 α -Acetoxy, 3 α -hydroxy-24-methyl-11 β , 18; 18,-20β;22,25-triepoxy-5α-furostane (9): white powder; mp 236-237 °C; [a]_D -27° (c 0.96, CHCl₃); IR (KBr) v_{max} 3447 $\rm cm^{-1};\,{}^{1}\!H$ NMR and $\rm {}^{13}\!C$ NMR data, see Tables 2 and 4; FABMS m/z 517 ([M + H]⁺, 2), 499 (3), and 439 (0.6); HRFABMS m/z517.3165 $[M + H]^+$ (calcd for $C_{30}H_{45}O_7$, 517.3167).

(22S)-3α-Acetoxy-11β,18α-dihydroxy-24-methyl-18,-**20β;22,25-diepoxy-5α-furostane** (11): white powder; mp 269–271 °C; [α]_D –43° (c 0.31, CHCl₃); IR (KBr) ν_{max} 3406, 1726, and 1265 cm⁻¹; ¹H NMR and ¹³C NMR data, see Tables 2 and 4; FABMS m/z 519 ([M + H]⁺); HRFABMS m/z 519.3328 $[M + H]^+$ (calcd for C₃₀H₄₇O₇, 519.3323).

(22S)-2α,3α-Diacetoxy-11β,18α-dihydroxy-24-methyl-18,20β;22,25-diepoxy-5α-furostane (12): white powder; mp 273–275 °C; $[\alpha]_D$ –23° (c 0.38, CHCl₃); IR (KBr) ν_{max} 3406, 1726, and 1265 cm⁻¹; ¹H NMR and ¹³C NMR data, see Tables 2 and 4; FABMS m/z 577 ([M + H]⁺, 10); HREIMS m/z 576.3300 $[M]^+$ (calcd for $C_{32}H_{48}O_9$, 576.3299).

22-epi-Hippuristan-11-one (13): white powder; mp 165-167 °C; $[\alpha]_{\rm D}$ +38° (c 0.16, CHCl₃); IR (KBr) $\nu_{\rm max}$ 3460 cm⁻¹; ¹H NMR and ¹³C NMR data, see Tables 1 and 3; FABMS m/z 461 $([M + H]^+, 4);$ HRFABMS m/z 461.3267 $[M + H]^+$ (calcd for $C_{28}H_{45}O_5, \ 461.3269).$

3-Acetyl-2-desacetylhippurin-1 (14): white powder; mp >300 °C; $[\alpha]_{\rm D}$ +27° (c 0.3, CHCl₃); IR (KBr) $\nu_{\rm max}$ 3435 and 1726 cm⁻¹; ¹H NMR and ¹³C NMR data, see Tables 1 and 3; FABMS m/z 521 ([M + H]⁺, 4); HRFABMS m/z 521.3478 [M + H]⁺ (calcd for $C_{30}H_{49}O_7$, 521.3480).

Hydrolysis of 1 to Compounds 2-4. A mixture of 1 (25.0 mg), LiOH (0.5 mg), and THF (1 mL) was stirred at room temperature for 2 h, followed by removal of the solvent under reduced pressure. Then, the resultant mixture was washed with brine (5 mL) and extracted with EtOAc (3×5 mL). The extracts were combined and dried over anhydrous MgSO₄. The solvent was removed under reduced pressure to gave a mixture of 1-4, which were chromatographed on normal-phase HPLC using MeOH/CH₂Cl₂ (gradient, 4-10%) to yield **2** (9.5 mg), **3** (4.1 mg), 4 (1.2 mg), and the reactant 1 (5 mg).

(R)- and (S)-MTPA Derivatives of 19. To a solution of compound 19 (5.0 mg, 1.0×10^{-2} mmol) in CHCl₃ (1.0 mL) at room temperature were added (R)-MTPA acid (11.7 mg, $5.0 \times$ 10⁻² mmol), 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (EDC) (9.6 mg, 5.0×10^{-2} mmol), and 4-(dimethylamino)pyridine (DMAP) (0.6 mg, 5.0×10^{-3} mmol), and the resultant mixture was stirred for 24 h at room temperature. The reaction mixture was concentrated under reduced pressure to give a crude product. Further purification was performed by a short silica gel column with *n*-hexane/acetone (2:1) to give 19b (1.5 mg) as a colorless oil. The (S)-MTPA ester **19a** (1.2 mg) was prepared in the same way. Selected $\Delta \delta$ values $[\delta(S) - \delta(R)]$ are as follows: H-11 = +0.02, H₂-1 = +0.09 and +0.01, H-3 = -0.10, H-4 = -0.02.

(S)-MTPA ester of 19: ¹H NMR (CDCl₃, 300 MHz), δ 5.23 (1H, br d, J = 12.2 Hz, H-2), 4.43 (1H, dt, J = 7.5, 5.4 Hz, H-16), 4.23 (1H, br s, H-11), 4.02 (1H, br s, H-3), 1.95 (1H, m, H-1a), 1.69 (1H, m, H-1b), 1.57 (2H, m, H-4), 1.34 (3H, s, H₃-18), 1.30 (3H, s, H_3 -21), 1.27 (3H, s, H_3 -26), 1.13 (3H, s, H_3 -19), 0.98 (3H, s, H_3 -27), 0.94 (3H, d, J = 6.8 Hz, H_3 -28).

(*R*)-MTPA ester of 19: ¹H NMR (CDCl₃, 300 MHz), δ 5.22 (1H, br d, J = 12.2 Hz, H-2), 4.43 (1H, dt, J = 7.5, 5.4 Hz, H-16), 4.21 (1H, br s, H-11), 4.12 (1H, br s, H-3), 1.94 (1H, m, H-1a), 1.60 (1H, m, H-1b), 1.59 (2H, m, H-4), 1.34 (3H, s, H₃-18), 1.30 (3H, s, H₃-21), 1.27 (3H, s, H₃-26), 1.13 (3H, s, H₃-19), 0.98 (3H, s, H₃-27), 0.94 (3H, d, J = 6.8 Hz, H₃-28).

Cytotoxicity Assays. Compounds were assayed for cytotoxicity against Hep G2, Hep 3B, A549, MCF-7, and MDA-MB-231 cells using the MTT [3-(4,5-dimethylthiazol-2-yl)-2,5diphenyltetrazolium bromide] method.¹⁴ Freshly trypsinized cell suspensions were seeded in 96-well microtiter plates at densities of 5000-10 000 cells per well with tested compounds added from DMSO-diluted stock. After 3 days in culture, attached cells were incubated with MTT (0.5 mg/mL, 1 h) and subsequently dissolved in DMSO. The absorbency at 550 nm was then measured using a microplate reader. The IC_{50} is the concentration of agent that reduced cell growth by 50% under the experimental conditions.

Acknowledgment. Financial support was provided by National Science Council of Taiwan (NSC92-2113-M-110-009) awarded to J.-H.S.

Supporting Information Available: This material is available free of charge via the Internet at http://pubs.acs.org.

References and Notes

- Kazlauskas, R.; Murphy, P. T.; Quinn, R. J.; Wells, R. J.; Schönholzer, P. Tetrahedron Lett. 1977, 4439–4442.
- (2) Higa, T.; Tanaka, J.; Tsukitani, Y.; Kikuchi, H. Chem. Lett. 1981, 11, 1647–1650.
- (3) Higa, T.; Tanaka, J.; Tachibana, K. Tetrahedron Lett. 1981, 29, 2777-2780.
- Rao, C. B.; Ramana, K. V.; Rao, D. V.; Fahy, E.; Faulkner, D. J. J. Nat.Prod. 1988, 51, 954-958.
 Tanaka, J.; Trianto, A.; Musman, M.; Yoshida, W. Y.; Ohtani, I. I.;
- Ichiba, T.; Higa, T.; Yoshida, W. Y.; Scheuer, P. J. Tetrahedron 2002 58, 6259-6266.
- (6) Tanaka, J.; Higa, T.; Tachibana, K.; Iwashita, T. Chem. Lett. 1982, 1295 - 1296.
- Sheu, J.-H.; Chen, S.-P.; Sung, P.-J.; Chiang, M. Y.; Dai, C.-F. *Tetrahedron Lett.* **2000**, *41*, 7885–7888.
 Sheu, J.-H.; Huang, L.-F.; Chen, S.-P.; Yang, Y.-L.; Sung, P.-J.; Wang, G.-H.; Su, J.-H.; Chao, C.-H.; Hu, W.-P.; Wang, J.-J. *J. Nat. Prod.* **2007** 2017 (2017) 2003, 66, 917-921.
- (9) González, N.; Barral, M. A.; Rodríguez, J.; Jiménez, C. Tetrahedron 2001, 57, 3487–3497.
- (10) Sheu, J.-H.; Hung, K.-C.; Wang, G.-H.; Duh, C.-Y. J. Nat. Prod. 2000, 63, 1603-1607.
- (11) Sheu, J.-H.; Chao, C.-H.; Wang, G..-H.; Hung, K.-C.; Duh, C.-Y.; Chiang, M.-Y.; Wu, Y.-C.; Wu, C.-C. Tetrahedron Lett. 2004, 45, 6413-6416.
- (12) Crystallography data (excluding structure factors) of 5 and 9 have been deposited with the Cambridge Crystallographic Data Center as supplementary publication numbers CCDC259656 and CCDC259655, (13) Ohtani, I.; Kusumi, T.; Kashman, Y.; Kakisawa, H. J. Am. Chem. Soc. 1991, 113, 4092–4096.
- Alley, M. C.; Acudiero, D. A.; Monks, A.; Hursey, M. L.; Czerwinski, M. J.; Fine, D. L.; Abbott, B. J.; Mayo, J. G.; Shoemark, R. H.; Boyd, M. R. *Cancer Res.* **1988**, *48*, 589–601. (14)

NP050033Y