# Revision of the Absolute Configuration at $\mathbf{C}(\mathbf{2 3})$ of Lanostanoids and Isolation of Secondary Metabolites from Formosan Soft Coral Nephthea erecta 

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#### Abstract

Three new oxygenated ergostanoids, $\mathbf{1}-\mathbf{3}$, one known ergostanoid, $\mathbf{4}$, one new trinoreudesmadienone, $\mathbf{5}$, one new calamenene type sesquiterpene, $\mathbf{6}$, and one known aristolane-type sesquiterpene, $(-)$ aristolone (7), have been isolated from the AcOEt extract of the soft coral Nephthea erecta. The structures of these compounds were determined by extensive spectroscopic and X-ray crystallographic analysis, as well as Mosher's method. We revised the absolute configuration at $\mathrm{C}(23)$ in the side chain of some lanostanoids as a result of the Mosher's products of $\mathbf{1}$ (i.e., $\mathbf{1 a}$ and $\mathbf{1 b}$ ). The cytotoxicities against selected cancer cells and the anti-inflammatory effects of these tested metabolites $\mathbf{1 - 7}$ were determined in vitro.


Introduction. - Soft corals belonging to the genus Nephthea have been found to be a rich source of terpenoids and steroids [1-13]. A number of terpenoids have shown an array of biological activities such as insecticidal [12] and cytotoxic activities [5][7][13]. Earlier studies of marine soft corals have led to the isolation of steroids, some of which have shown cytotoxic [6] and anti-inflammatory activities [14a]. In the course of our ongoing search to explore bioactive substances from marine resources, chromatographic separation of the AcOEt extracts from the soft coral N. erecta led to the isolation of three new oxygenated ergostanoids, 1-3, one known ergostanoid, 4, [14b], one new trinoreudesmadienone, $\mathbf{5}$, one new calamenene type sesquiterpene, $\mathbf{6}$, and one known aristolane type sesquiterpene, (-)-aristolone (7) [15]. We describe here the isolation, structural elucidation, cytotoxicity, and anti-inflammatory activity of these metabolites. In addition, we revised the absolute configuration at $\mathrm{C}(23)$ in the side chain of some lanostanoids isolated from the bark and leaves of Amentotaxus formosana [16] [17] by using a modified Mosher's method [18][19] for $\mathbf{1}$.

Results and Discussion. - (3 $3,23 S$ )-Ergosta-5,24(28)-diene-3,23-diol (1) was isolated as a white amorphous powder. HR-ESI-MS of $\mathbf{1}$ exhibited a pseudo-molecular ion peak at $m / z 437.3392\left([M+\mathrm{Na}]^{+}\right.$; calc. for 437.3395) and established a molecular

formula of $\mathrm{C}_{28} \mathrm{H}_{46} \mathrm{O}_{2}$, indicating six degrees of unsaturation. The ${ }^{13} \mathrm{C}$-NMR and DEPT spectra displayed 28 signals, including those of five Me , ten $\mathrm{CH}_{2}$, nine CH groups, and four quaternary C-atoms. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ Signals $(\delta(\mathrm{H}) 4.20(t, J=6.7,1 \mathrm{H})$ and $3.49-3.56$ $(m, 1 \mathrm{H}))\left(\right.$ Table 1) and an IR absorption at $3323 \mathrm{~cm}^{-1}$, together with the observation of two O-bearing C-atom resonances ( $\delta(\mathrm{C}) 74.4$ and 71.8 ) in ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum (Table 1), revealed the presence of two OH groups. Furthermore, a trisubstituted $\mathrm{C}=\mathrm{C}$ bond $(\delta(\mathrm{C}) 121.6$ and 140.7) and a terminal $\mathrm{C}=\mathrm{C}$ bond $(\delta(\mathrm{C}) 108.1$ and 159.2) were assigned on the basis of ${ }^{13} \mathrm{C}$-NMR and DEPT spectra of $\mathbf{1}$. The above functionalities accounted for two of the six degrees of unsaturation of $\mathbf{1}$, which possesses a tetracyclic skeleton. The structure of $\mathbf{1}$ was established by extensive 2D-NMR analysis $\left({ }^{1} \mathrm{H},{ }^{1} \mathrm{H}-\right.$ COSY, HMQC, and HMBC). Interpretation of the ${ }^{1} \mathrm{H},{ }^{1} \mathrm{H}-\mathrm{COSY}$ spectrum led to partial structures I, II, and III (Fig. 1). Partial structures I and II were connected, on the



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Fig. 1. Key ${ }^{1} H,{ }^{1} \mathrm{H}-\mathrm{COSY}(-)$ and HMBC correlations $(\mathrm{H} \rightarrow \mathrm{C})$ of $\mathbf{1}-\mathbf{3}, \mathbf{5}$, and $\mathbf{6}$

Table 1. ${ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}$-NMR Data of Compounds $\mathbf{1 - 3}$. $\delta$ in ppm, $J$ in Hz.

| Position | $\mathbf{1 a}^{\text {a }}$ ) |  | $\mathbf{2}^{\text {a }}$ ) |  | $3^{\text {b }}$ ) |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\delta(\mathrm{C})$ | $\delta(\mathrm{H})$ | $\delta(\mathrm{C})$ | $\delta(\mathrm{H})$ | $\delta(\mathrm{C})$ | $\delta(\mathrm{H})$ |
| 1 | 37.2 (t) | $\begin{aligned} & 1.04-1.14\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.82-1.92\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ | 37.2 (t) | $\begin{aligned} & 1.04-1.14\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.82-1.92\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ | 37.2 ( $t$ ) | $\begin{aligned} & 1.04-1.14\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.81-1.88\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ |
| 2 | 31.6 ( $t$ ) | $\begin{aligned} & 1.79-1.89\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.38-1.48\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ | 31.6 (t) | $\begin{aligned} & 1.81-1.91\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.37-1.47\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ | 31.6 (t) | $\begin{aligned} & 1.80-1.87\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.46-1.54\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ |
| 3 | 71.8 (d) | 3.49-3.56 (m) | 71.8 (d) | 3.49-3.56 (m) | 71.7 (d) | 3.51-3.57 (m) |
| 4 | 42.2 ( $t$ ) | $\begin{aligned} & 2.26-2.34\left(m, \mathrm{H}_{\alpha}\right), \\ & 2.19-2.26\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ | 42.3 ( $t$ ) | $\begin{aligned} & 2.26-2.34\left(m, \mathrm{H}_{\alpha}\right), \\ & 2.19-2.26\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ | 42.2 ( $t$ ) | $\begin{aligned} & 2.26-2.33\left(m, \mathrm{H}_{\alpha}\right), \\ & 2.19-2.26\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ |
| 5 | 140.7 (s) |  | 140.7 (s) |  | 140.6 (s) |  |
| 6 | 121.6 (d) | 5.35 (br. $d, J=4.9)$ | 121.7 (d) | 5.35 (br. $d, J=4.5)$ | 121.6 (d) | 5.36 (br. $d, J=5.0)$ |
| 7 | 31.8 (t) | $\begin{aligned} & 1.55-1.64\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.93-2.05\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ | 31.9 (t) | $\begin{aligned} & 1.59-1.68\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.95-2.07\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ | 31.8 (t) | $\begin{aligned} & 1.56-1.65\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.94-2.03\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ |
| 8 | 31.8 (d) | 1.47-1.54 (m) | 31.9 (d) | 1.48-1.55 (m) | 32.3 (d) | 1.45-1.52 (m) |
| 9 | 50.0 (d) | 0.86-1.00 (m) | 50.1 (d) | 0.90-1.04 (m) | 49.6 (d) | 0.94-1.02 (m) |
| 10 | 36.5 (s) |  | 36.5 (s) |  | 36.4 (s) |  |
| 11 | 21.0 (t) | $\begin{aligned} & 1.49-1.57\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.38-1.48\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ | 21.1 (t) | $\begin{aligned} & 1.51-1.59\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.41-1.51\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ | 20.9 (t) | $\begin{aligned} & 1.54-1.62\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.42-1.50\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ |
| 12 | 39.7 (t) | $\begin{aligned} & 1.12-1.24\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.98-2.09\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ | 39.8 (t) | $\begin{aligned} & 1.12-1.24\left(m, \mathrm{H}_{\alpha}\right), \\ & 2.00-2.11\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ | 32.3 (t) | $\begin{aligned} & 1.42-1.50\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.67-1.75\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ |
| 13 | 42.4 (s) |  | 42.5 (s) |  | 47.3 (s) |  |
| 14 | 56.7 (d) | 0.95-1.10 (m) | 56.7 (d) | 0.97-1.12 (m) | 50.9 (d) | 1.68-1.76 (m) |
| 15 | 24.2 (t) | $\begin{aligned} & 1.55-1.67\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.13-1.21\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ | 24.2 (t) | $\begin{aligned} & 1.57-1.69\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.13-1.21\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ | 23.6 (t) | $\begin{aligned} & 1.71-1.79\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.12-1.20\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ |
| 16 | 28.4 (t) | $\begin{aligned} & 1.84-1.98\left(m, \mathrm{H}_{\alpha}\right) \\ & 1.20-1.30\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ | 28.4 (t) | $\begin{aligned} & 1.82-1.96\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.22-1.32\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ | 38.0 (t) | $\begin{aligned} & 1.86-1.92\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.83-1.86\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ |
| 17 | 56.8 (d) | 1.10-1.26 (m) | 56.8 (d) | 1.13-1.29 (m) | 87.5 (s) |  |
| 18 | 11.8 (q) | 0.67 (s) | 11.9 (q) | 0.73 (s) | 14.5 (q) | 0.76 (s) |
| 19 | 19.4 (q) | 1.00 (s) | 19.4 (q) | 1.01 (s) | 19.4 (q) | 1.02 (s) |
| 20 | 34.1 (d) | 1.36-1.49 (m) | 33.0 (d) | 1.71-1.84 (m) | 41.6 (d) | 1.69-1.77 (m) |
| 21 | 19.5 (q) | $1.00(d, J=6.5)$ | 18.5 (q) | $1.02(d, J=6.5)$ | 7.5 (q) | $1.01(d, J=7.5)$ |
| 22 | 42.6 (t) | $\begin{aligned} & 1.72-1.83(\mathrm{~m}) \\ & 1.21-1.35(\mathrm{~m}) \end{aligned}$ | 43.6 (t) | $\begin{aligned} & 1.52-1.63(\mathrm{~m}), \\ & 1.11-1.25(\mathrm{~m}) \end{aligned}$ | 71.5 (d) | $4.19(d d, J=8.0,5.0)$ |
| 23 | 74.4 (d) | $4.20(t, J=6.7)$ | 71.1 (d) | $4.16(d, J=10.2)$ | 40.6 (t) | $\begin{aligned} & 2.28(d d, J=14.0,5.0) \\ & 2.10(d d, J=14.0,5.0) \end{aligned}$ |
| 24 | 159.2 (s) |  | 160.6 (s) |  | 152.7 (s) |  |
| 25 | 29.8 (d) | 2.18-2.40 (m) | 30.6 (d) | 2.09-2.31 (m) | 33.6 (d) | 2.18-2.28 (m) |
| 26 | 23.1 (q) | $1.08(d, J=6.8)$ | 22.5 (q) | $1.05(d, J=7.0)$ | 21.8 (q) | $1.04(d, J=6.5)$ |
| 27 | 23.6 (q) | $1.09(d, J=6.8)$ | 23.3 (q) | $1.08(d, J=7.0)$ | 21.9 (q) | $1.06(d, J=6.5)$ |
| 28 | 108.1 (t) | $5.02(s), 4.92(s)$ | 105.7 (t) | 5.06 (s), 4.88 (s) | 109.5 ( $t$ ) | 4.90 (s), 4.79 (s) |

$\left.{ }^{\text {a }}\right)$ Spectra were recorded in $\mathrm{CDCl}_{3}\left({ }^{1} \mathrm{H}: 300 \mathrm{MHz}\right.$ and $\left.\left.{ }^{13} \mathrm{C}: 75 \mathrm{MHz}\right) .{ }^{\mathrm{b}}\right)$ Spectra were recorded in $\mathrm{CDCl}_{3}$ ( ${ }^{1} \mathrm{H}: 500 \mathrm{MHz}$ and ${ }^{13} \mathrm{C}: 125 \mathrm{MHz}$ ).
basis of HMBC correlations, through $\mathrm{CH}_{2}(4) / \mathrm{C}(5)$ and $\mathrm{CH}_{2}(4) / \mathrm{C}(6)$. In addition, II and III were connected through the terminal methylidene based on HMBC correlations between $\mathrm{Me}(26), \mathrm{Me}(27) / \mathrm{C}(24), \mathrm{C}(25)$, and between $\mathrm{CH}_{2}(28) / \mathrm{C}(23), \mathrm{C}(24), \mathrm{C}(25)$. Rings $A$ and $B$ were elucidated on the basis of HMBC cross-peaks between $\mathrm{Me}(19) /$ $\mathrm{C}(1), \mathrm{C}(5), \mathrm{C}(9), \mathrm{C}(10)$ and $\mathrm{CH}_{2}(4) / \mathrm{C}(3)$, whereas rings $C$ and $D$ were determined
based on HMBC correlations between $\mathrm{Me}(18) / \mathrm{C}(12), \mathrm{C}(13), \mathrm{C}(14), \mathrm{C}(17)$ (Fig. 1). NOESY Correlations observed between $\mathrm{H}_{\beta}-\mathrm{C}(11)$ and $\mathrm{H}-\mathrm{C}(8), \mathrm{H}_{\beta}-\mathrm{C}(11)$ and $\mathrm{Me}(18), \mathrm{H}_{\beta}-\mathrm{C}(11)$ and $\mathrm{Me}(19), \mathrm{H}-\mathrm{C}(9)$ and $\mathrm{H}-\mathrm{C}(14), \mathrm{Me}(18)$ and $\mathrm{H}-\mathrm{C}(8)$, $\mathrm{Me}(19)$ and $\mathrm{H}-\mathrm{C}(8), \mathrm{Me}(18)$ and $\mathrm{H}-\mathrm{C}(20), \mathrm{H}-\mathrm{C}(3)$ and $\mathrm{H}_{a}-\mathrm{C}(1), \mathrm{Me}(21)$ and $\mathrm{H}_{\beta}-\mathrm{C}(12)$, and $\mathrm{Me}(19)$ and $\mathrm{H}_{\beta}-\mathrm{C}(4)$ in $\mathbf{1}$ confirmed the relative configurations for each ring junction and stereogenic center. By using a modified Mosher's method, the absolute configurations at $\mathrm{C}(3)$ and $\mathrm{C}(23)$ were established. Analysis of the $\Delta \delta\left(=\delta_{(S)}-\right.$ $\delta_{(R)}$ ) values (Table 2) according to the Mosher model [18][19] pointed to an ( $S$ )configuration for $\mathrm{C}(23)$ of $\mathbf{1}$, because $\mathrm{H}-\mathrm{C}(25), \mathrm{Me}(26)$, $\mathrm{Me}(27)$, and $\mathrm{CH}_{2}(28)$ of $(R)$ MTPA ester 1b were more shielded by the phenyl ring of MTPA ( $=\alpha$-methoxy- $\alpha$ (trifluoromethyl)phenylacetic acid) products. We thereby revised the absolute configuration at $\mathrm{C}(23)$ in the side chain of some lanostanoids which were isolated from the bark and leaves of Amentotaxus formosana [16][17]. Thus, the structure of $\mathbf{1}$ was established unambiguously.

Compound 2, a white amorphous powder, was found to possess the same molecular formula as that of $\mathbf{1}\left(\mathrm{C}_{28} \mathrm{H}_{46} \mathrm{O}_{2}\right)$, as revealed by HR-ESI-MS. The spectral data of $\mathbf{2}$ (Table 1) differed from 1 only for $\mathrm{C}(23)(\delta(\mathrm{H}) 4.20(t, J=6.7,1 \mathrm{H}), \delta(\mathrm{C}) 74.4$ of $\mathbf{1}$ and $\delta(\mathrm{H}) 4.16(d, J=10.2,1 \mathrm{H}), \delta(\mathrm{C}) 71.1$ of $\mathbf{2})$. The above result, together with the comparison of 2D-NMR spectral data of $\mathbf{2}$ with those of $\mathbf{1}$, disclosed that the two compounds were epimers. The significant ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ chemical-shift differences between $\mathbf{1}$ and $\mathbf{2}$ demonstrated that $\mathrm{C}(23)$ of $\mathbf{2}$ possessed the $(R)$-configuration. From

Table 2. Selected ${ }^{l} H-N M R$ Data of Mosher's Esters of $\mathbf{1}$ and 3. $\delta$ in ppm.

| Position | 1 ${ }^{\text {a }}$ ) |  |  | Position | $3^{\text {a }}$ ) |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\begin{aligned} & (S) \text {-MTPA } \\ & \mathbf{1 a} \end{aligned}$ | $(R) \text {-MTPA }$ 1b | $\Delta \delta$ |  | $\begin{aligned} & (S) \text {-MTPA } \\ & \mathbf{3 a} \end{aligned}$ | $(R) \text {-MTPA }$ $\mathbf{3 b}$ | $\Delta \delta$ |
| 2 | $\begin{aligned} & 1.9814, \\ & 1.5857 \end{aligned}$ | $\begin{aligned} & 2.0195, \\ & 1.6786 \end{aligned}$ | $\begin{aligned} & -0.0381, \\ & -0.0929 \end{aligned}$ | 2 | $\begin{aligned} & 1.8579, \\ & 1.5713 \end{aligned}$ | $\begin{aligned} & 1.9147, \\ & 1.6853 \end{aligned}$ | $\begin{aligned} & -0.0568 \\ & -0.1140 \end{aligned}$ |
| 4 | $\begin{aligned} & 2.5685, \\ & 2.5385 \end{aligned}$ | $\begin{aligned} & 2.5227, \\ & 2.4623 \end{aligned}$ | $\begin{aligned} & +0.0458 \\ & +0.0762 \end{aligned}$ | 4 | $\begin{aligned} & 2.4941, \\ & 2.4435 \end{aligned}$ | $\begin{aligned} & 2.4679 \\ & 2.4059 \end{aligned}$ | $\begin{aligned} & +0.0251 \\ & +0.0376 \end{aligned}$ |
| 6 | 5.4314 | 5.3997 | +0.0317 | 6 | 5.3490 | 5.3100 | +0.0390 |
| 7 | $\begin{aligned} & 1.9118, \\ & 1.5470 \end{aligned}$ | $\begin{aligned} & 1.8645, \\ & 1.5113 \end{aligned}$ | $\begin{aligned} & +0.0473 \\ & +0.0357 \end{aligned}$ | 7 | $\begin{aligned} & 1.8579, \\ & 1.5385 \end{aligned}$ | $\begin{aligned} & 1.8216, \\ & 1.5054 \end{aligned}$ | $\begin{aligned} & +0.0363 \\ & +0.0331 \end{aligned}$ |
| 18 | 0.6442 | 0.6682 | -0.0240 | 18 | 0.7751 | 0.7530 | $+0.0221$ |
| 19 | 0.9123 | 0.9161 | -0.0038 | 19 | 0.8976 | 0.8878 | +0.0098 |
| 20 | 1.4855 | 1.5391 | -0.0536 | 20 | 2.2400 | 2.1343 | +0.1057 |
| 21 | 1.0191 | 1.0375 | -0.0184 | 21 | 1.2850 | 1.2406 | $+0.0444$ |
| 22 | $\begin{aligned} & 1.9363, \\ & 1.6348 \end{aligned}$ | $\begin{aligned} & 2.0013, \\ & 1.6612 \end{aligned}$ | $\begin{aligned} & -0.0650 \\ & -0.0264 \end{aligned}$ | 23 | $\begin{aligned} & 2.4941, \\ & 2.2320 \end{aligned}$ | $\begin{aligned} & 2.6356, \\ & 2.4679 \end{aligned}$ | $\begin{aligned} & -0.1415, \\ & -0.1490 \end{aligned}$ |
| 25 | 2.4619 | 2.3914 | +0.0705 | 25 | 2.3801 | 2.4059 | -0.0258 |
| 26 | 1.1232 | 1.1113 | +0.0119 | 26 | 1.0140 | 1.0396 | -0.0256 |
| 27 | 1.1557 | 1.1332 | +0.0225 | 27 | 1.0366 | 1.0587 | -0.0221 |
| 28 | $\begin{aligned} & 5.3134, \\ & 5.1600 \end{aligned}$ | $\begin{aligned} & 5.2047, \\ & 5.0908 \end{aligned}$ | $\begin{aligned} & +0.1087 \\ & +0.0692 \end{aligned}$ | 28 | 4.8689 | 4.992 | -0.1231 |

[^0]the aforementioned data, the structure of $\mathbf{2}$ was assigned as ( $3 \beta, 23 R$ )-ergosta-5,24(28)-diene-3,23-diol.

Compound $\mathbf{3}$ had a molecular formula of $\mathrm{C}_{28} \mathrm{H}_{46} \mathrm{O}_{3}$ as determined by HR-ESI-MS. IR Spectrum showed the presence of OH group(s) $\left(3332 \mathrm{~cm}^{-1}\right)$. The presence of two $\mathrm{CH}-\mathrm{O}$ groups and one O-bearing quaternary C -atom were inferred from ${ }^{1} \mathrm{H}-\mathrm{NMR}$ $(\delta(\mathrm{H}) 3.51-3.57(m, 1 \mathrm{H}), 4.19(d d, J=8.0,5.0,1 \mathrm{H}))$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}(\delta(\mathrm{C}) 71.7,71.5$, and 87.5) spectra. ${ }^{1} \mathrm{H}-$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectra also exhibited signals due to five Me $\operatorname{groups}(\delta(\mathrm{H}) 0.76(s, 3 \mathrm{H}), 1.02(s, 3 \mathrm{H}), 1.04(d, J=6.5,3 \mathrm{H}), 1.06(d, J=6.5,3 \mathrm{H}), 1.01$ $(d, J=7.5,3 \mathrm{H})$ ), and a trisubstituted $\mathrm{C}=\mathrm{C}$ bond $(\delta(\mathrm{H}) 5.36($ br. $d, J=5.0,1 \mathrm{H}), \delta(\mathrm{C})$ 121.6 and 140.6), and a terminal $\mathrm{C}=\mathrm{C}$ bond $(\delta(\mathrm{H}) 4.90,4.79(s$, each 1 H$), \delta(\mathrm{C}) 109.5$ and 152.7). ${ }^{13} \mathrm{C}$-NMR and DEPT spectra of $\mathbf{3}$ also exhibited 28 signals, including those of five Me , ten $\mathrm{CH}_{2}$, eight CH groups, and five quaternary C -atoms. This evidence suggested that 3 possessed the similar steroid skeleton ( $A, B, C$, and $D$ rings) as of ergosta-5,24(28)-dien-3 $\beta$-ol except for the presence of additional OH groups at $\mathrm{C}(17)$ and $\mathrm{C}(22)$, which were confirmed by COSY and HMBC spectra of $\mathbf{3}$ (Fig. 1). By using Mosher's method [18][19], the absolute configuration at C(22) was readily defined by the analysis of NMR-shift data of the corresponding C(22) ( $S$ )- and ( $R$ )-MTPA esters. The significant ${ }^{1} \mathrm{H}$-NMR chemical-shift differences between the $(S)$ - and $(R)$-MTPA esters 3a and 3b demonstrated that $\mathrm{C}(22)$ possessed the ( $S$ )-configuration (Table 2), because $\mathrm{CH}(25), \mathrm{Me}(26), \mathrm{Me}(27)$, and $\mathrm{CH}_{2}(28)$ of $(R)$-MTPA ester $\mathbf{3 b}$ were more shielded by the phenyl ring of MTPA products. In addition, the $\Delta$ value $\left(\delta_{\mathrm{CDCl}_{3}}-\delta_{\mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N}}\right)$ of the $\mathrm{Me}(18) \mathrm{H}$-atoms ( 300 MHz ) of $\mathbf{3}$ in two different solvents $\left(\mathrm{CDCl}_{3}\right.$ and $\left(\mathrm{D}_{5}\right)$ pyridine $\left.\left(\mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N}\right)\right)$ is small $(-0.02 \mathrm{ppm})$, whereas the dihedral angle between the OH group and the Me group is large, indicating the $\alpha$-orientation of OH at $\mathrm{C}(17)$ [20]. Therefore, the structure of $\mathbf{3}$ was assigned as ( $3 \beta, 22 S$ )-ergosta-5,24(28)-diene-3,17,22-triol.

Compound 5 was obtained as a colorless oil. According to HR-EI-MS ( $\mathrm{m} / \mathrm{z}$ $\left.176.1202\left(M^{+}\right)\right)$and ${ }^{13} \mathrm{C}$-NMR data, its molecular formula was established as $\mathrm{C}_{12} \mathrm{H}_{16} \mathrm{O}$, indicating five degrees of unsaturation. UV Spectrum displayed a $\lambda_{\max }(\mathrm{MeOH})$ value of 241 nm , indicating the presence of a cyclohexa-2,5-dienone moiety, supported by an IR absorption at $1657 \mathrm{~cm}^{-1}$ (conjugated $\mathrm{C}=\mathrm{O}$ group). In addition, the ${ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}$-NMR spectra (Table 3) of $\mathbf{5}$ contained resonances for a cyclohexa-2,5-dienone moiety $(\delta(\mathrm{H}) 6.12(d, J=1.5,1 \mathrm{H}), 6.22(d d, J=10.0,1.5,1 \mathrm{H})$, and $6.78(d, J=10.0$, $1 \mathrm{H}) ; \delta(\mathrm{C}) 171.0,121.5,187.5,126.4$, and 158.1). Thus, 5 was suggested to be a bicyclic structure. From the COSY spectrum of $\mathbf{5}$ ( Fig. 1), it was possible to establish the Hatom sequence from $\mathrm{CH}_{2}(5)$ to $\mathrm{Me}(9) .{ }^{1} \mathrm{H},{ }^{1} \mathrm{H}$-COSY correlations further observed between $\mathrm{CH}(3)$ and $\mathrm{CH}(4)$ indicated the $(Z)$-form olefinic coupling ( $J=10.0 \mathrm{~Hz}$ ). The connectivities between $\mathrm{C}(4 \mathrm{a})$ and $\mathrm{C}(4), \mathrm{C}(5)$, and $\mathrm{C}(8 \mathrm{a})$ were confirmed by HMBC correlations of $\mathrm{Me}(10)$ with $\mathrm{C}(4), \mathrm{C}(4 a), \mathrm{C}(5)$, and $\mathrm{C}(8 \mathrm{a})$. Furthermore, HMBC correlations from $\mathrm{Me}(9)$ to $\mathrm{C}(8)$ and $\mathrm{C}(8 \mathrm{a})$, suggested that $\mathrm{C}(8)$ and $\mathrm{C}(8 \mathrm{a})$ were connected. The relative configuration of 5 was determined by NOESY data (Fig. 2). NOE Correlations from $\mathrm{H}-\mathrm{C}(8)$ to $\mathrm{H}_{\alpha}-\mathrm{C}(7), \mathrm{H}_{\alpha}-\mathrm{C}(5)$, and $\mathrm{Me}(10)$ suggested that these H -atoms were oriented on the same side. NOE Correlations also indicated that $\mathrm{H}_{\beta}-\mathrm{C}(5)$ and $\mathrm{H}_{\beta}-\mathrm{C}(6), \mathrm{H}_{\beta}-\mathrm{C}(6)$ and $\mathrm{H}_{\beta}-\mathrm{C}(7)$, and $\mathrm{H}_{\beta}-\mathrm{C}(7)$ and $\mathrm{Me}(9)$ were on the opposite side. From the aforementioned data, $\mathbf{5}$ was determined as ( $4 \mathrm{a} S^{*}, 8 S^{*}$ )-5,6,7,8-tetrahydro-4a,8-dimethylnaphthalen-2(4a $H$ )-one. An isomer $\left(4 \mathrm{a} R^{*}, 8 S^{*}\right)-5,6,7,8$-tetrahydro-4a,8-dimethylnaphthalen-2(4a $H$ )-one of 5 had been

Table 3. ${ }^{1} H$ - and ${ }^{13} C$-NMR Data of 5 and 6. At 500 and 300 MHz , resp., $\delta$ in ppm, $J$ in Hz.

| Position | $5^{\text {a }}$ ) |  | $\mathbf{6}^{\text {a }}$ ) |  |
| :---: | :---: | :---: | :---: | :---: |
|  | $\delta(\mathrm{C})$ | $\delta(\mathrm{H})$ | $\delta(\mathrm{C})$ | $\delta(\mathrm{H})$ |
| 1 | 121.5 (d) | $6.12(d, J=1.5)$ | 130.0 (d) | 7.73 (br.s) |
| 2 | 187.5 (s) |  | 141.1 (s) |  |
| 3 | 126.4 (d) | 6.22 (dd, $J=10.0,1.5)$ | 126.7 (d) | 7.64 (br. $d, J=7.9$ ) |
| 4 | 158.1 (d) | $6.78(d, J=10.0)$ | 127.9 (d) | $7.39(d, J=7.9)$ |
| 4a | 41.2 (s) |  | 150.7 (s) |  |
| 5 | 38.5 (t) | $\begin{aligned} & 1.80-1.86\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.33-1.35\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ | 33.4 (d) | 2.83-2.93 (m) |
| 6 | 21.0 ( $t$ ) | $\begin{aligned} & 1.78-1.85\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.66-1.73\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ | 30.1 (t) | $\begin{aligned} & 1.32-1.42\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.92-2.08\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ |
| 7 | 37.0 ( $t$ ) | $\begin{aligned} & 1.95-2.03\left(m, \mathrm{H}_{\alpha}\right), \\ & 1.04-1.10\left(m, \mathrm{H}_{\beta}\right) \end{aligned}$ | 21.1 (t) | $1.59-1.69\left(m, \mathrm{H}_{\beta}\right)$ |
| 8 | 34.1 (d) | 2.46-2.54 (m) | 43.8 (d) | 2.79 (br. $d d, J=12.5,6.3)$ |
| 8a | 171.0 (s) |  | 134.0 (s) |  |
| 9 | 17.7 (q) | $1.14(d, J=7.0)$ | 32.0 (d) | 2.22-2.32 (m) |
| 10 | 23.4 (q) | 1.27 (s) | 17.4 (q) | $0.73(d, J=6.8)$ |
| 11 |  |  | 21.2 (q) | 1.03 ( $d, J=6.8$ ) |
| 12 |  |  | 22.1 (q) | $1.31(d, J=7.0)$ |
| 13 |  |  | 192.5 (d) | 9.96 (s) |

${ }^{\text {a }}$ ) Spectra were recorded in $\mathrm{CDCl}_{3}$.
prepared as an intermediate in a spirovetivane synthesis [21]. The assignments of $\mathbf{5}$ were disclosed (Table 3) by extensive 2D-NMR data analyses for the first time.


5


6


7

Fig. 2. Selected NOESY correlations of $\mathbf{5}$ and $\mathbf{6}$, and X-ray ORTEP representation of $\mathbf{7}$

HR-ESI-MS of (+)-trans-calamenen-13-al (6), a canary yellow syrup, showed a molecular formula of $\mathrm{C}_{15} \mathrm{H}_{20} \mathrm{O}$, implying six degrees of unsaturation. IR Spectrum of 6 indicated the presence of a conjugated benzaldehyde function ( 2867 and $1695 \mathrm{~cm}^{-1}$ ). The UV absorption maximum at 262 nm confirmed the aforementioned function. The NMR spectra (Table 3) indicated a trisubstituted phenyl moiety ( ${ }^{1} \mathrm{H}-\mathrm{NMR}: \delta(\mathrm{H}) 7.39$, $7.64(d, J=7.9,1 \mathrm{H}$ each ), and 7.73 (br. $s, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR} \delta(\mathrm{C}) 127.9,126.7,130.0,141.1$, 134.0, and 150.7). The ${ }^{13} \mathrm{C}-\mathrm{NMR}$ and DEPT spectra revealed 15 signals, including those for three Me , two $\mathrm{CH}_{2}$, seven CH groups, and three quaternary C -atoms. The above
data of 6 were similar to those of ( - -trans-calamenene [22], except for the replacement of the Me by an carbaldehyde group $(\delta(\mathrm{H}) 9.96(s, 1 \mathrm{H})$ and $\delta(\mathrm{C}) 192.5$ (d)) at $\mathrm{C}(2)$. This was supported by the HMBC spectrum (Fig. 1) and long-range ${ }^{1} \mathrm{H},{ }^{13} \mathrm{C}$-correlations from $\mathrm{CH}(1)$ to $\left.\mathrm{C}(13)\right)$. NOESY Correlations (Fig. 2) between $\mathrm{H}_{\alpha}-\mathrm{C}(6)$ and the H -atoms of $\mathrm{H}_{\alpha}-\mathrm{C}(7)$ and $\mathrm{Me}(12)$, and between $\mathrm{H}_{\alpha}-\mathrm{C}(7)$ and $\mathrm{H}-\mathrm{C}(8)$ suggested that these H -atoms were oriented on the same side of the cyclohexene moiety, in which the i-Pr group at $\mathrm{C}(8)$ was oriented on the opposite side. The absolute configuration of $\mathbf{6}$ was further established by comparison of its optical rotation with that of ( - -trans-calamenene [22]. On the basis of these evidences, the structure of $\mathbf{6}$ was established unambiguously.

Compound 7, identified by comparison of spectral data with those of $(-)$-aristolone [15], was obtained as a colorless hexagonal crystal from hexane. The absolute configuration of aristol-9-en-8-one (7) was further confirmed by X-ray diffraction analyses (Fig. 2) [23].

Compounds 1-4 and $\mathbf{7}$ were tested for cytotoxicity against P-388 (mouse lymphocytic leukemia) and HT-29 (human colon adenocarcinoma) cancer cell lines. The results showed that compounds $\mathbf{1}, \mathbf{2}, \mathbf{4}$, and $\mathbf{7}$ were not cytotoxic against these two cancer cell lines. However, compound $\mathbf{3}$ exhibited cytotoxicity against P-388 cell line with an $E D_{50}$ value of $3.7 \mu \mathrm{~g} / \mathrm{ml}$. The anticancer agent mithramycin was used as the positive control and exhibited an $E D_{50}$ value of 0.06 and $0.08 \mu \mathrm{~g} / \mathrm{ml}$ against P-388 and HT-29 cells, respectively. In addition, as shown in Fig. 3, the in vitro anti-inflammatory activity of compounds $\mathbf{1 - 7}$ was tested using LPS-stimulated cell. Stimulation of RAW 264.7 cell with LPS resulted in up-regulation of the pro-inflammatory iNOS (inducible nitric oxide synthetase) and COX-2 (cyclooxygenase-2) protein. Both compounds $\mathbf{1}$ and $\mathbf{3}$, at a concentration of $10 \mu \mathrm{~m}$, significantly reduced the levels of the iNOS ( $45.8 \pm$ 9.9 and $33.6 \pm 20.6 \%$, resp.) and COX-2 protein ( $68.1 \pm 2.3$ and $10.3 \pm 6.2 \%$, resp.) compared with the control cells stimulated with LPS. At the same concentration, compounds 2 and $\mathbf{4}$ did not inhibit the COX-2 protein expression, but significantly


Fig. 3. Effect of compounds 1-7 at $10 \mu \mathrm{~m}$ concentration on the LPS-induced pro-inflammatory iNOS and COX-2 protein expression of RAW264.7 macrophage cells by immunoblot analysis. a) Immunoblot of iNOS; $b$ ) immunoblot of COX-2. The values are mean $\pm$ SEM $(n=5)$. Relative intensity of the LPS-alone-stimulated group was taken as $100 \%$. *: Significantly different from LPS-stimulated (control) group ( $P<0.05$ ).
inhibited iNOS protein expression ( $62.8 \pm 1.3$ and $15.6 \pm 4.1 \%$, resp., compared with control cells) by LPS stimulation. Moreover, cell survival was not changed by the presence of compounds 1-7 at the same concentration. Under the same experimental conditions, $10 \mu \mathrm{~m}$ CAPE (caffeic acid phenylthyl ester; Sigma Chemical. Company, St. Louis, MO) reduced the levels of the iNOS and COX-2 protein to $1.5 \pm 2.1$ and $70.2 \pm$ $11.5 \%$, respectively, relative to the control cells stimulated with LPS.

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## Experimental Part

General. Silica gel 60 (Merck, 230-400 mesh) was used for column chromatography (CC); precoated Si gel plates (Merck, Kieselgel $60 F_{254}, 0.25 \mathrm{~mm}$ ) were used for TLC. High-performance liquid chromatography (HPLC): a Hitachi L-7420 UV detector L-7100 pump apparatus equipped with Merck Hibar RP-18e column $(250 \times 10 \mathrm{~mm}, 5 \mu \mathrm{~m})$. Optical rotations: JASCO P1020 polarimeter. UV Spectra: Hitachi U-3210 spectrophotometer. IR Spectra: JASCO FT/IR-4100 spectrophotometer. NMR Spectra: Bruker Avance 300 NMR spectrometer at 300 MHz for ${ }^{1} \mathrm{H}$ and 75 MHz for ${ }^{13} \mathrm{C}$, or Varian Unity INOVA 500 FT-NMR spectrometer at 500 MHz for ${ }^{1} \mathrm{H}$ and 125 MHz for ${ }^{13} \mathrm{C}$, with TMS as internal standard; chemical shifts in $\delta(\mathrm{ppm})$ and coupling constants $J$ in Hz. ESI-MS: ESI-FT-MS on a Bruker APEX II mass spectrometer. EI-MS: JEOL JMSSX/SX $102 A$ mass spectrometer at 70 eV .

Soft Coral Material. The soft coral N. erecta was collected by hand using scuba at Green Island located in the southeast coast of Taiwan, in July 2005, at a depth of 10 m , and was stored in a freezer for 5 weeks until extraction. This specimen was identified by Prof. Chang-Feng Dai, and a voucher specimen (GN-80) was deposited with the Department of Marine Biotechnology and Resources, National Sun Yatsen University.

Extraction and Isolation. A specimen of N. erecta was extracted sequentially with fresh acetone and MeOH at r.t. The combined acetone extracts were concentrated to a brown gum, which was partitioned between $\mathrm{H}_{2} \mathrm{O}$ and AcOEt. The AcOEt extract ( 35.0 g ) was chromatographed over a silica-gel column using hexane, hexane/ AcOEt , and $\mathrm{AcOEt} / \mathrm{MeOH}$ mixtures of increasing polarity. Elution with hexane gave fractions containing compounds 6 and 7, and elution with hexane/AcOEt $90: 10$ gave fractions containing compounds $\mathbf{1}, \mathbf{2}, \mathbf{3}$, and $\mathbf{4}$. Compounds $\mathbf{6}(5 \mathrm{mg})$ and $\mathbf{7}(5 \mathrm{mg})$ were further purified by HPLC (Si) by eluting with hexane. Compounds $\mathbf{1}(7 \mathrm{mg}), \mathbf{2}(2 \mathrm{mg}), \mathbf{3}(14 \mathrm{mg})$, and $\mathbf{4}(10 \mathrm{mg})$ were further purified by $R P-18 \mathrm{HPLC}\left(95 \% \mathrm{MeOH}\right.$ in $\left.\mathrm{H}_{2} \mathrm{O}\right)$. In addition, the combined MeOH extract was concentrated to a brown gum, which was partitioned between $\mathrm{H}_{2} \mathrm{O}$ and AcOEt. The AcOEt extract was chromatographed over a silica-gel column using $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and MeOH mixtures of increasing polarity. A fraction (eluted from $90 \% \mathrm{CH}_{2} \mathrm{Cl}_{2}$ in MeOH ) was subjected to repeated $R P-18 \mathrm{HPLC}(60 \% \mathrm{MeOH}$ in $\left.\mathrm{H}_{2} \mathrm{O}\right)$ to afford $5(1 \mathrm{mg})$.
(3 $\beta, 23 \mathrm{~S}$ )-Ergosta-5,24(28)-diene-3,23-diol (1). White amorphous powder. $[\alpha]_{\mathrm{D}}^{23}=+16(c=0.7$, $\mathrm{CHCl}_{3}$ ). IR (KBr): 3323, 2925, 1465, 1374, 1055, 901, 799. ${ }^{1} \mathrm{H}-$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ : see Table 1. ESI-MS (pos.): $437\left([M+\mathrm{Na}]^{+}\right)$. HR-ESI-MS: $437.3392\left([M+\mathrm{Na}]^{+}, \mathrm{C}_{28} \mathrm{H}_{46} \mathrm{NaO}_{2}^{+}\right.$; calc. 437.3395).
(3ß,23R)-Ergosta-5,24(28)-diene-3,23-diol (2). White amorphous powder. $[\alpha]_{\mathrm{D}}^{23}=+96(c=0.2$, $\mathrm{CHCl}_{3}$ ). IR (KBr): 3332, 2929, 1465, 1374, 1059, 902, 803. ${ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ : see Table 1. ESI-MS (pos.): $437\left([M+\mathrm{Na}]^{+}\right)$. HR-ESI-MS: $437.3394\left([M+\mathrm{Na}]^{+}, \mathrm{C}_{28} \mathrm{H}_{46} \mathrm{NaO}_{2}^{+}\right.$; calc. 437.3395).
(3 $3,22 \mathrm{~S}$ )-Ergosta-5,24(28)-diene-3,17,22-triol (3). White amorphous powder. $[\alpha]_{\mathrm{D}}^{23}=+14(c=0.7$, $\mathrm{CHCl}_{3}$ ). IR (KBr): 3332, 2929, 1438, 1369, 1095, 1054, 892, 767. ${ }^{1} \mathrm{H}-$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ : see Table 1. ESI-MS (pos.): $453\left([M+\mathrm{Na}]^{+}\right)$. HR-ESI-MS: $453.3345\left([M+\mathrm{Na}]^{+}, \mathrm{C}_{28} \mathrm{H}_{46} \mathrm{NaO}_{3}^{+}\right.$; calc. 453.3344$)$.
$\left(4 a \mathrm{~S}^{*}, 8 \mathrm{~S}^{*}\right)-5,6,7,8$-Tetrahydro-4a,8-dimethylnaphthalen-2(4aH)-one (5). Colorless oil. $[\alpha]_{\mathrm{D}}^{23}=+128$ $\left(c=0.1, \mathrm{CHCl}_{3}\right) . \mathrm{UV}(\mathrm{MeOH}): 241$ (3.98). IR (KBr): 2926, 1657, 1461, 1378, 749. ${ }^{1} \mathrm{H}-$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ : see Table 3. EI-MS: 176 ( $M^{+}$). HR-EI-MS: $176.1202\left(M^{+}, \mathrm{C}_{12} \mathrm{H}_{16} \mathrm{O}^{+}\right.$; calc. 176.1201).
(5S,8R)-5,6,7,8-Tetrahydro-5-methyl-8-(1-methylethyl)naphthalene-2-carbaldehyde (6). Canary yellow syrup. $[\alpha]_{\mathrm{D}}^{23}=+42\left(c=0.5, \mathrm{CHCl}_{3}\right)$. UV ( MeOH ): 262 (3.18). IR ( KBr ): 2952, 2867, 1695, 1603,

1564, 1456, 1379, 902, 824. ${ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}$-NMR: see Table 3. ESI-MS (pos.) : $217\left([M+\mathrm{H}]^{+}\right)$. HR-ESI-MS: $217.1594\left([M+\mathrm{H}]^{+}, \mathrm{C}_{15} \mathrm{H}_{21} \mathrm{O}^{+}\right.$; calc. 217.1592).

Preparation of $(\mathrm{R})$ - and (S)-MTPA ( $=\alpha$-Methoxy- $\alpha$-(trifluoromethyl)phenylacetic Acid) Esters of $\mathbf{1}$ and 3. Two aliquots of $\mathbf{1}(1 \mathrm{mg})$ were dissolved in $\left(\mathrm{D}_{5}\right)$ pyridine $\left(\mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N} ; 0.6 \mathrm{ml}\right)$ and allowed to react overnight with $(R)$ - and ( $S$ )-MTPA chloride (one drop), affording the $(S)$ - and ( $R$ )-MTPA esters 1a and 1b, resp. In the same manner, $\mathbf{3}(1 \mathrm{mg})$ was dissolved in $\mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N}(0.6 \mathrm{ml})$ and reacted with $(R)$ - and $(S)$ MTPA chloride to afford $(S)$-MTPA ester 3a and $(R)$-MTPA ester 3b, resp. Selected $\Delta \delta$ values were shown in Table 2.

Cytotoxicity Testing. P-388 Cells were kindly supplied by Prof. J. M. Pezzuto, formerly of the Department of Medicinal Chemistry and Pharmacognosy, University of Illinois at Chicago; HT-29 cells were purchased from the American Type Culture Collection (ATCC). Cytotoxicity assays of the tested compounds were performed on the basis of a MTT assay method. The experimental details about the biological assay were carried out according to the procedure described in [24].

In vitro Anti-Inflammatory Assay. The anti-inflammatory assay was modified from Ho et al. [25] and Park et al. [26]. Murine RAW 264.7 macrophages were obtained from ATCC (No. TIB-71). The cells were activated by incubation in a medium containing Escherichia coli LPS $(0.01 \mu \mathrm{~g} / \mathrm{ml}$; Sigma $)$ for 16 h in the presence or absence of various compounds. Then, cells were washed with ice-cold phosphate-buffered saline (PBS), lysed in ice cold lysis buffer, and then centrifuged at 20000 g for 30 min at $4^{\circ}$. The supernatant was decanted from the pellet and retained for Western blot analysis. Protein concentrations were determined by the DC protein assay kit (Bio-Rad) modified by the method of Lowry et al. [27]. Samples containing equal quantities of protein were subjected to sodium dodecyl sulfate (SDS)polyacrylamide gel electrophoresis, and the separated proteins were electrophoretically transferred to polyvinylidene difluoride membranes (PVDF; Immobilon-P, Millipore, $0.45-\mu \mathrm{m}$ pore size). The resultant PVDF membranes were incubated with blocking soln. and incubated for 180 min with antibody against inducible nitric oxide synthase (iNOS; 1:1000 dilution; Transduction Laboratories) and cyclooxygenase2 (COX-2; 1:1000 dilution; Cayman Chemical) protein. The blots were detected using enhanced chemiluminescence (ECL) detection reagents (Perkin-Elmer, Western Blot Chemiluminescence Reagent Plus) according to the manufacturer instructions.

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[^0]:    $\left.{ }^{\text {a }}\right)$ Spectra were recorded in $\mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N}(300 \mathrm{MHz})$.

