

Mechanical Properties of Al₂O₃-NiAl Composites

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The mechanical properties of the Al₂O₃-NiAl system are investigated in the present study. Specimens containing 0 to 100 vol% NiAl in Al₂O₃ were prepared by hot pressing. Both the strength and toughness of the Al₂O₃-NiAl composites are higher than the values predicted by the rule of mixtures. The grain growth of Al₂O₃ and NiAl in the composites is constrained by each component. The increase in strength is thus partly attributed to microstructural refinement. The toughness enhancement is contributed by a combination of crack deflection and crack bridging.

I. Introduction

THE mechanical properties of ceramics can be improved by adding metallic reinforcements, such as Al,¹ Ni,² Ag,³ Cu,⁴ Fe,⁵ Ni₃Al,⁶ NiAl,⁷ and Nb₃Al.⁸ Among the metals investigated, NiAl has potential for high-temperature applications due to its high melting point, 1640°C. Furthermore, the oxidation and corrosion resistance of NiAl are superior among metals.⁹ For metal-toughened ceramics, the toughening effect is contributed either by crack bridging¹⁻⁸ or by crack deflection.^{5,7} Experimental studies have also suggested that the toughening behavior depends strongly on the shape of the metallic reinforcements. The strength of alumina can also be enhanced by adding metallic particles.^{2-8,10} The strengthening effect is qualitatively attributed to the refinement of matrix grains.^{2-5,7,10} In the present study, NiAl is chosen as the reinforcement material for Al₂O₃. Furthermore, monolithic Al₂O₃ and monolithic NiAl are also prepared with the same processing technique. The strength and toughness of the metal-toughened aluminas are determined.

II. Experimental Procedure

Alumina (TM-DR, Taimei Chemical Co., Ltd., Tokyo, Japan) and various amounts of nickel aluminide (NiAl, Xform Inc., New York) were milled together in ethyl alcohol with an attriator (Model 01-HD, Union Process Inc., U.S.A.) for 12 h. The amount of powder mixtures was around 100 g for each milling run. The rotation speed of the shaft was 300 rpm. Zirconia balls with diameters of 5 mm were used as grinding media. The particle size distribution of the as-received Al₂O₃ and NiAl powders was determined with a laser particle size analyzer (LS 230, Coulter Co., U.S.A.).

Sintering was performed by hot pressing at 1450°C in a graphite die for 1 h, under an applied pressure of 24.5 MPa. The vapor pressure during hot pressing was kept below 5×10^{-3} torr. The dimensions of the hot-pressed specimen were 50 mm in diameter and roughly 4.5 mm in thickness. Hot-pressed specimens were cut into rectangular bars with a diamond saw.

Rectangular specimens were then machined longitudinally with a diamond wheel. The final dimensions of the specimens were 4 mm × 3 mm × 34 mm. The final density was determined by the water displacement method. Phase identification was performed by X-ray powder diffractometry (XRD) with CuK α radiation. The polished surface was prepared by grinding with a diamond slurry to 6 μ m and polishing with silica suspension to 0.05 μ m. The microstructure was observed by SEM. The volume fraction of NiAl in the planes perpendicular and parallel to the hot-pressing direction was determined with an image analyzer (Quantimet 520, Cambridge, Co., England). The size of NiAl grains was determined with the line intercept procedure. The interconnectivity of NiAl grains in the composites was determined by measuring their electrical resistivity at room temperature. The strength of the specimens was determined by the four-point bending technique at ambient conditions. The inner and outer spans were 10 and 30 mm, respectively. The rate of loading was 0.5 mm/min. Fracture toughness was determined by the single-edge-notched-beam (SENB) technique. The notch was generated by cutting with a diamond saw. The thickness of the blade was 0.33 mm. The width of the notch was 0.4 mm.

III. Results and Discussion

The particle size analysis reveals that the median size of the as-received Al₂O₃ and NiAl particles is 0.2 and 5.9 μ m, respectively. The XRD analysis indicates that no phases other than α -Al₂O₃ and β -NiAl are produced in composites after hot pressing. The density of the hot-pressed Al₂O₃-NiAl composites is higher than 97.8%. The microstructures of the planes perpendicular and parallel to the hot-pressing direction for the composite containing 50 vol% NiAl are shown in Fig. 1. Elongated NiAl particles are found within the Al₂O₃ matrix. Compressive and shear stresses are applied on the NiAl particles during attrition milling. The shape of the as-received NiAl particles is angular. The shape of NiAl particles is changed to elongated flakes during attrition milling. The elongated NiAl flakes tend to stay on the plane that is perpendicular to the hot-pressing direction during hot pressing. Therefore, the long axes of most NiAl particles are perpendicular to the hot-pressing direction. The line intercept of the NiAl particles is shown as a function of NiAl content in Fig. 2. The intercept of the NiAl particles on the planes parallel to the hot-pressing direction is taken along the hot-pressing direction. The intercept can be treated as the thickness of the NiAl flakes for most flakes are in a plane perpendicular to the hot-pressing direction. The long axes of the NiAl flakes on the plane perpendicular to the hot-pressing direction are randomly distributed (Fig. 1). The aspect ratio of the elongated flake can be estimated by dividing the size of NiAl on the perpendicular plane to the intercept on the parallel plane. The aspect ratio of the NiAl flakes in the composites varies in a relatively narrow range, from 3 to 4.

The electrical resistivity of the composites is shown as a function of NiAl content in Fig. 3. It suggests that the NiAl grains are interconnected when more than 20 vol% NiAl is added to alumina. The presence of NiAl particles can increase the interdiffusion distance between Al₂O₃ grains. Furthermore, a small amount of Ni from NiAl may be dissolved in Al₂O₃

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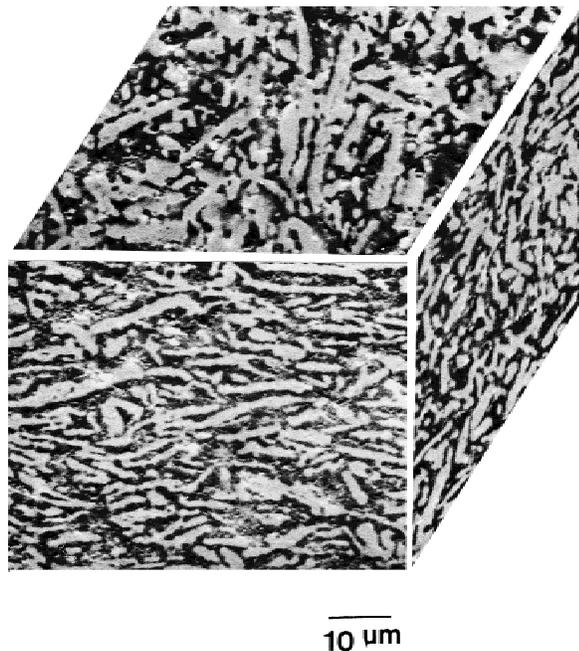


Fig. 1. SEM micrographs of the composite containing 50 vol% NiAl. The bright phase in the micrographs is NiAl. The micrographs were taken separately from the planes perpendicular and parallel to the hot-pressing direction. The micrographs were then placed to give a 3-D image of the hot-pressed composite.

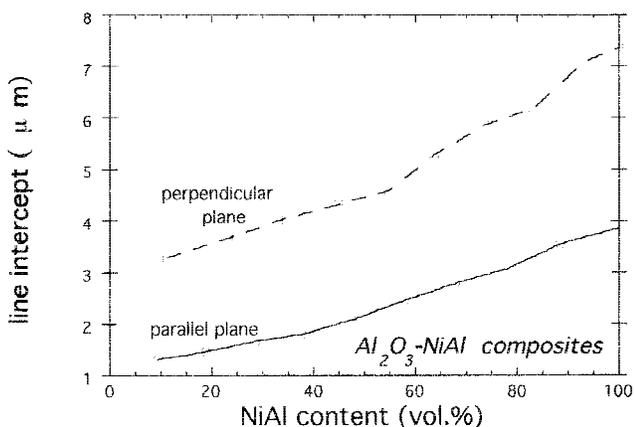


Fig. 2. Line intercept of NiAl particles on the planes perpendicular and parallel to the hot-pressing direction as a function of NiAl content.

during sintering. The nickel ions prefer to segregate at the grain boundaries of alumina.¹¹ The size of Al₂O₃ grains is thus reduced because of the presence of NiAl particles.⁷

The flexural strength of the Al₂O₃-NiAl composites is shown as a function of NiAl content in Fig. 4. The values in the figure indicate an average of six to eight specimens. Error bars show the maximum and minimum values measured. The straight line predicted by the rule of mixtures is also shown for comparison. The strength of the composite is a complex combination of matrix strength, reinforcement strength, interface strength, and the residual stresses induced by thermal expansion mismatch. The rule of mixtures reflects only the contribution from matrix strength and reinforcement strength. Therefore, the values predicted by the rule of mixtures can only be treated as a basis for comparison. The grain growth of each phase in the composites is constrained by its counterpart. The strengthening effect can thus be attributed partly to the microstructural refinement.

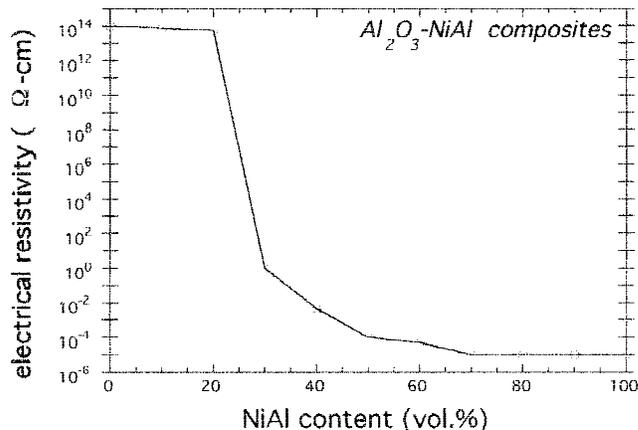


Fig. 3. Electrical resistivity of Al₂O₃-NiAl composites as a function of NiAl content.

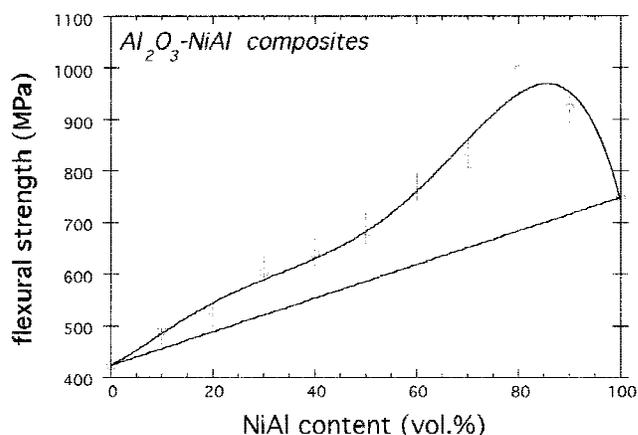


Fig. 4. Flexural strength of the Al₂O₃-NiAl composites as a function of NiAl content.

The toughness of the composites is shown as a function of NiAl content in Fig. 5. The values in the figure show the average values of six to eight specimens. The notch width used in the present study is large. It can lead to overestimation of the toughness. Nevertheless, the trend of the increase in toughness with the increase of NiAl content is clear. The values predicted by the rule of mixtures are shown for comparison. The tough-

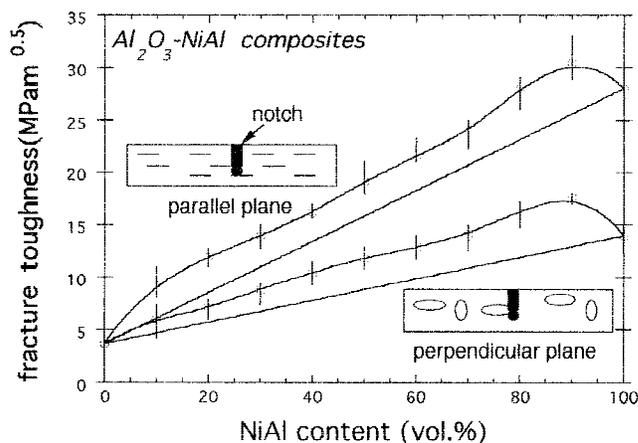


Fig. 5. Fracture toughness of the Al₂O₃-NiAl composites as a function of NiAl content. The notch is either parallel or perpendicular to the hot-pressing direction.

ness of the composites is higher than the values predicted by the rule of mixtures. Furthermore, strong dependence on the orientation is observed. Crack deflection can take place for the $\text{Al}_2\text{O}_3/\text{NiAl}$ interface is relatively weak.⁷ The crack surfaces can also be bridged by the interconnected NiAl grains (Fig. 6). The toughness of Al_2O_3 is therefore enhanced.

IV. Conclusions

The microstructure and mechanical properties of Al_2O_3 -NiAl composites have been investigated in the present study. The strength of the composites is high because of their refined microstructures. Elongated NiAl flakes are produced after at-

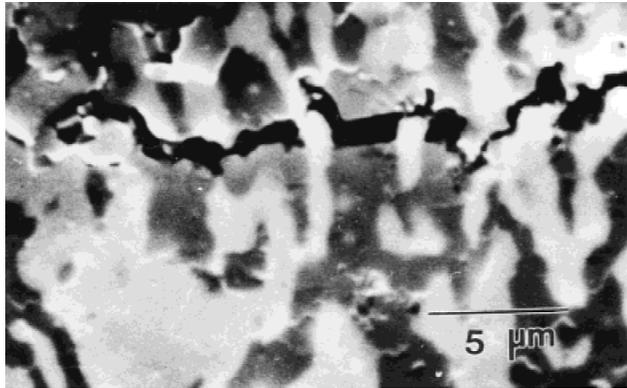


Fig. 6. Interactions between a crack generated by indentation and NiAl grains. The composite contains 50 vol% NiAl. Note that the crack surfaces are bridged by the interconnected NiAl particles.

trition milling. These flakes lie in the plane which is perpendicular to the hot-pressing direction. Therefore, the toughness of the composites shows a strong dependence on orientation. The toughening effect is contributed by a combination of crack deflection and crack bridging due to the presence of weak $\text{Al}_2\text{O}_3/\text{NiAl}$ interface and the interconnected NiAl particles.

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